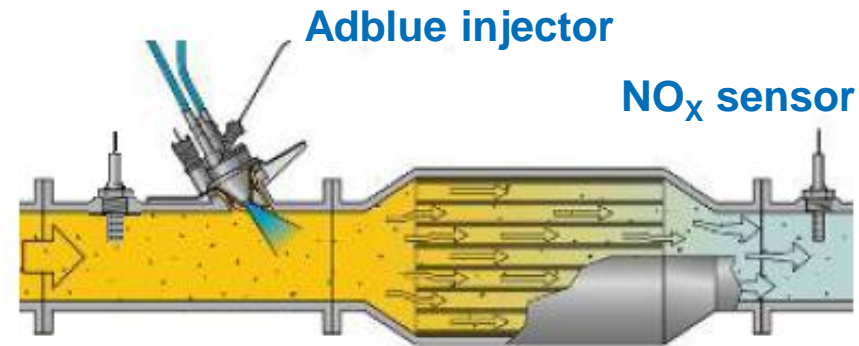


Hydrothermal aging modeling for SCR catalysts

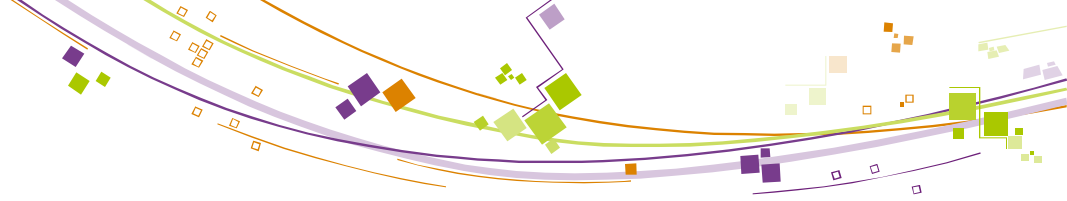
Simon Dosda
2015 CLEERS WORKSHOP

Interest of SCR modeling for control purpose

- Urea-SCR is one of the most common technology used by car and truck manufacturers to meet the latest NO_x emission standards
 - requires to optimize urea injection in order to limit ammonia slip while keeping a good NO_x reduction efficiency



- In this context, system simulation has significant advantages to set up optimal injection strategy:
 - *Physical modeling*: modeling level is adapted for control laws design in order to obtain accurate and consistent results in every real life case
 - *Fast calculation time*: about 10 times faster than the real time for a complex exhaust line coupled with a high frequency control



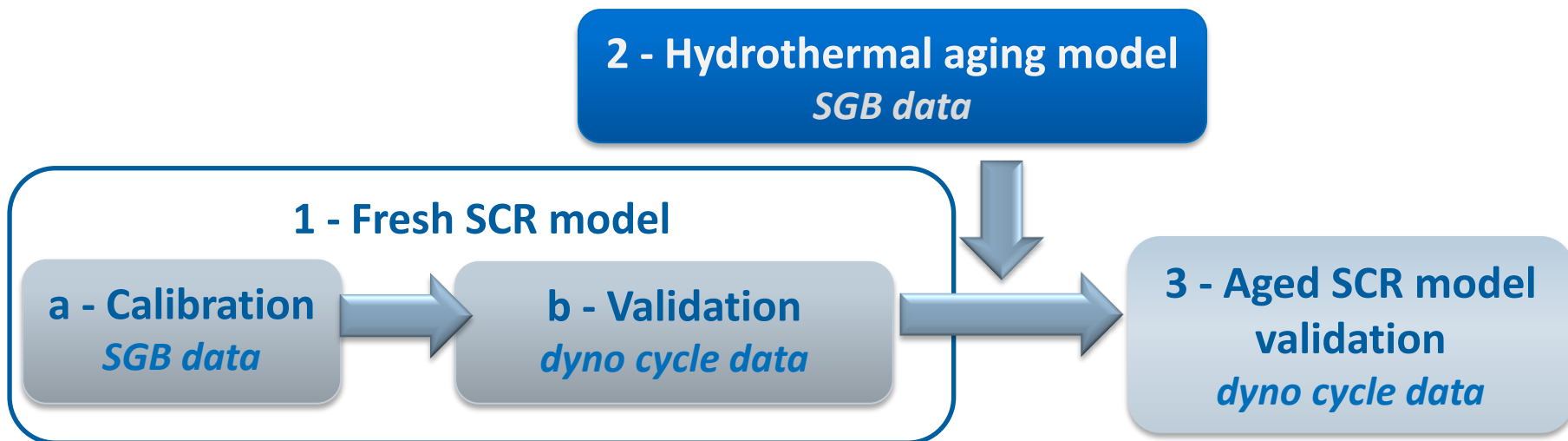
Importance of aging on SCR performances

- **During engine lifetime, performance degradation of the SCR catalyst is observed due to harsh exhaust conditions**
 - frequent filter regeneration subjects the SCR catalyst to high temperatures
 - hydrothermal aging will strongly decrease DeNO_x and NH₃ storage properties of the catalyst
- **Modeling hydrothermal aging impact on the catalyst allows to adapt urea injection strategy and avoid excessive amount of ammonia slip**
 - only way to adapt injection strategy over the complete vehicle lifetime

→ *Is there a way to model efficiently hydrothermal aging impact on a SCR catalyst ?*



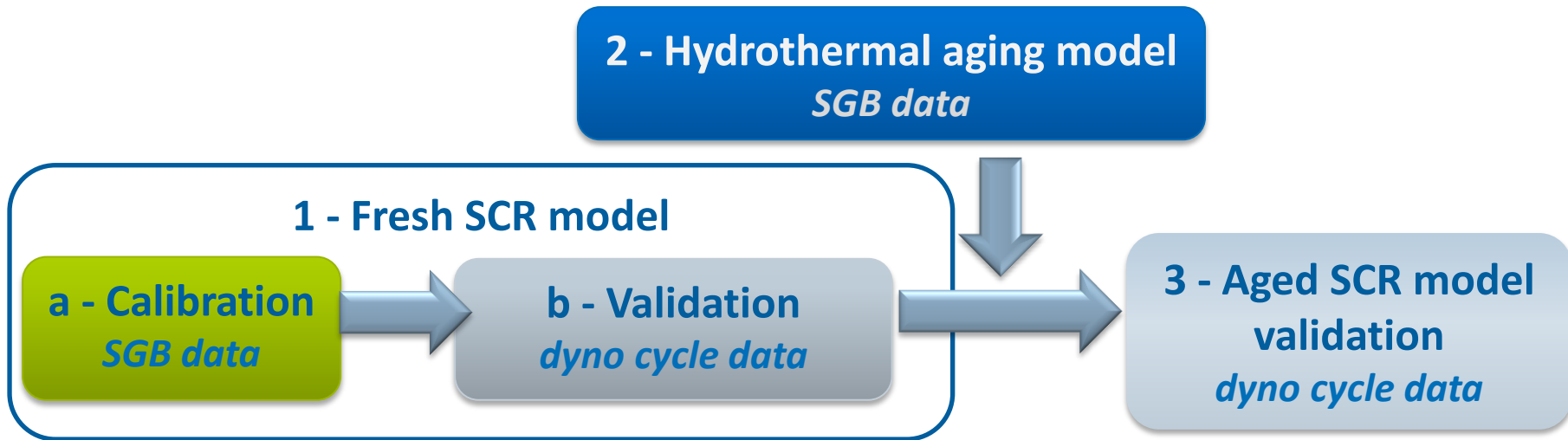
Approach: from fresh to aged SCR model



- **1 - Fresh SCR model set up and validation:**
 - a - SCR model is calibrated on specific Synthetic Gas Bench tests
 - b - SCR model is integrated in a complete exhaust line simulator and validated on chassis dynamometer on test cycles
- **2 - Aging model set up from SGB tests for various aging conditions**
- **3 - Aging model validation on dyno on test cycles using an aged SCR**



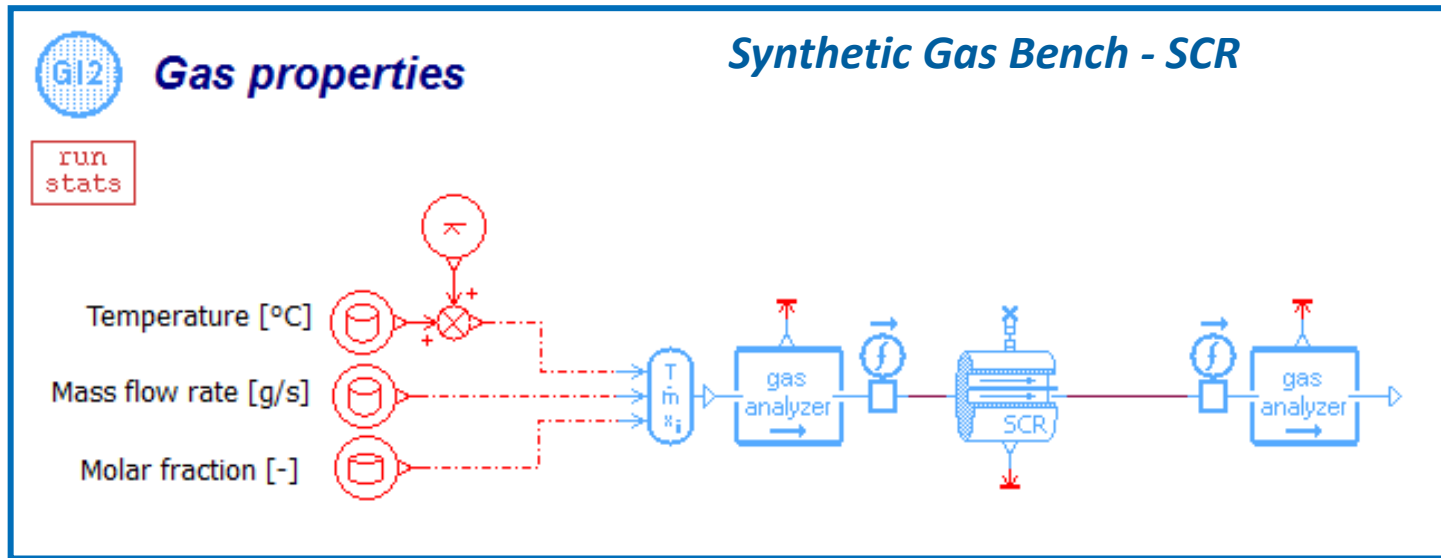
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Fresh SCR model presentation – Reactor scale

- LMS Imagine.Lab Amesim model of SGB SCR using IFP-Exhaust library

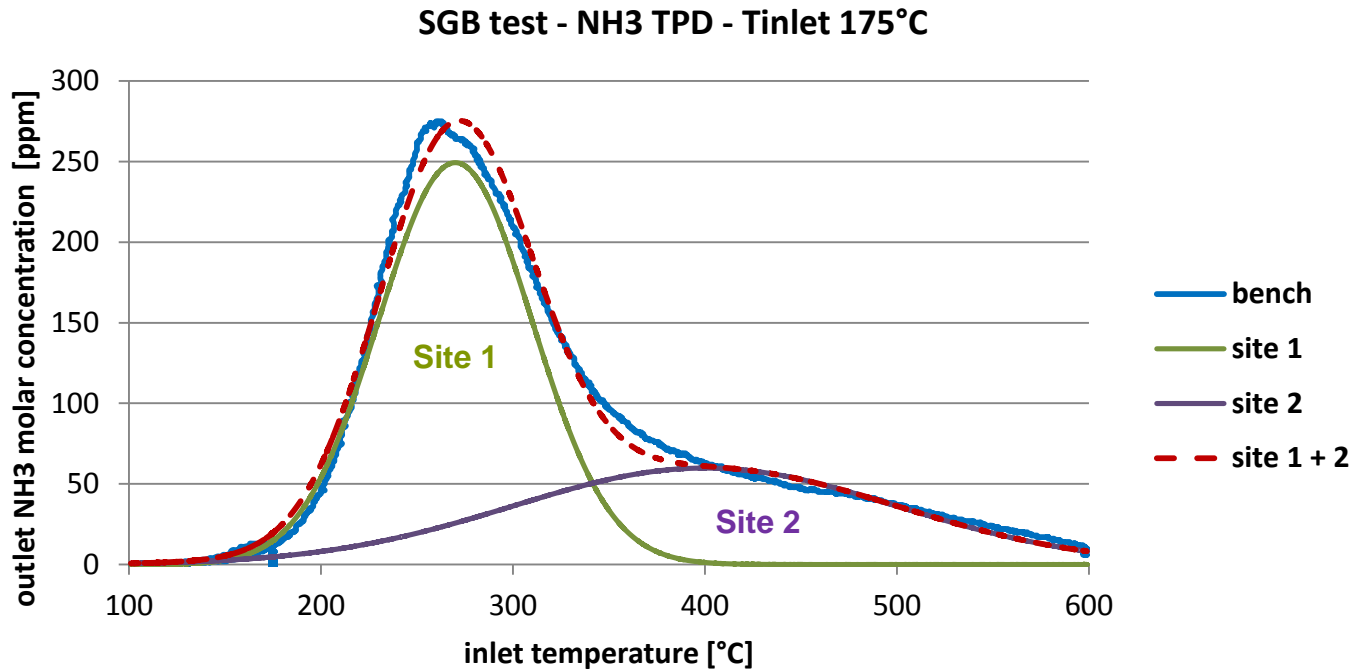


- reactions considered: NH_3 adsorption/desorption, DeNO_x reactions (standard, fast & NO₂ SCR) and NH_3 oxidation
- isolated sample: no thermal exchanges considered
- SCR sample geometry: length = 3 inches, diameter = 2 inches, 350 cpsi



SCR modeling set up for NH₃ storage on SGB test

- From NH₃ TPD observation, two adsorption sites can be considered



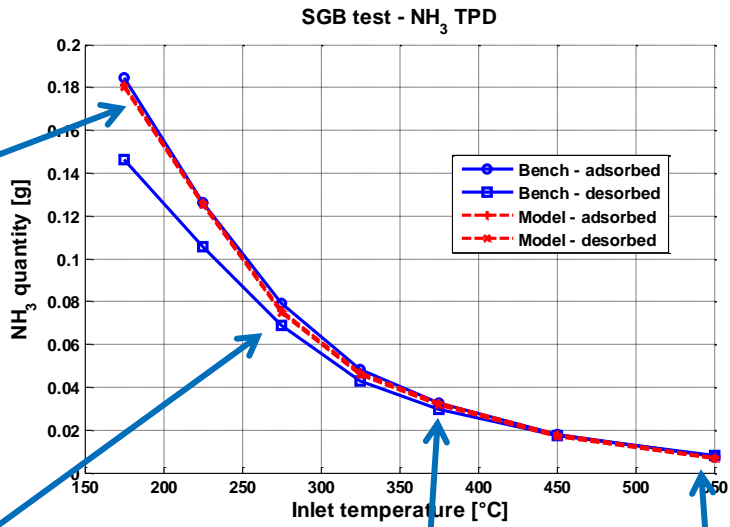
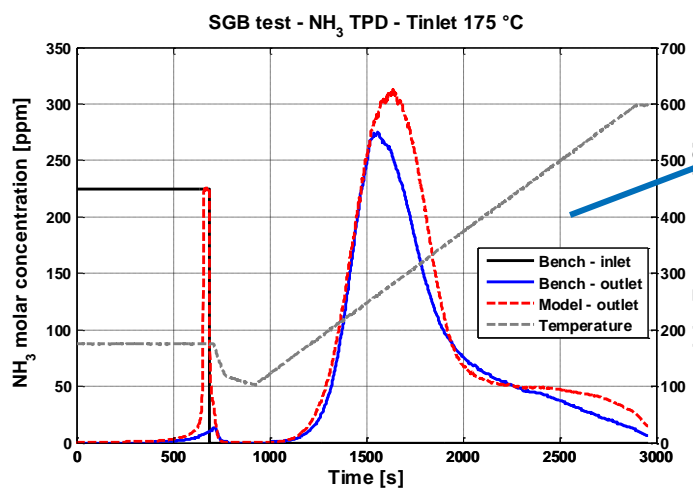
- NH₃ adsorption reaction for each adsorption site:



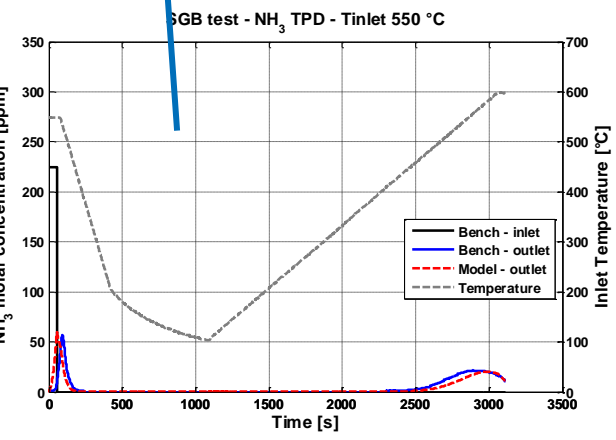
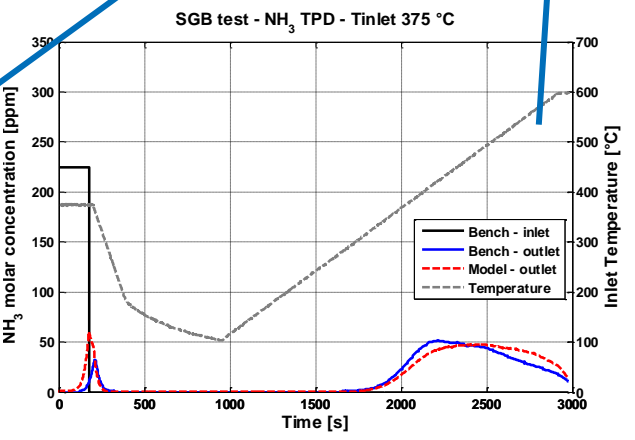
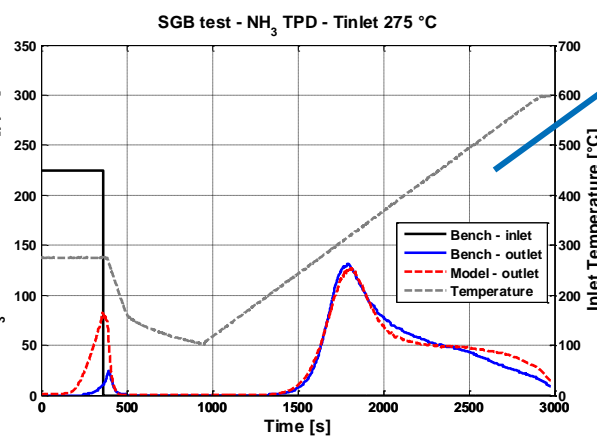
- $$r_{ad} = k_{ad} \times [NH_3] \times (1 - \theta_{NH_3}) - k_{de} \times e^{\frac{Ede \times (1 - \alpha \times \theta_{NH_3})}{R \times T_s}} \times \theta_{NH_3}$$

SCR model calibration for NH₃ storage on SGB test

NH₃ storage calibration is done on NH₃ TPD tests with various inlet temperature



Good results consistency for the overall test data



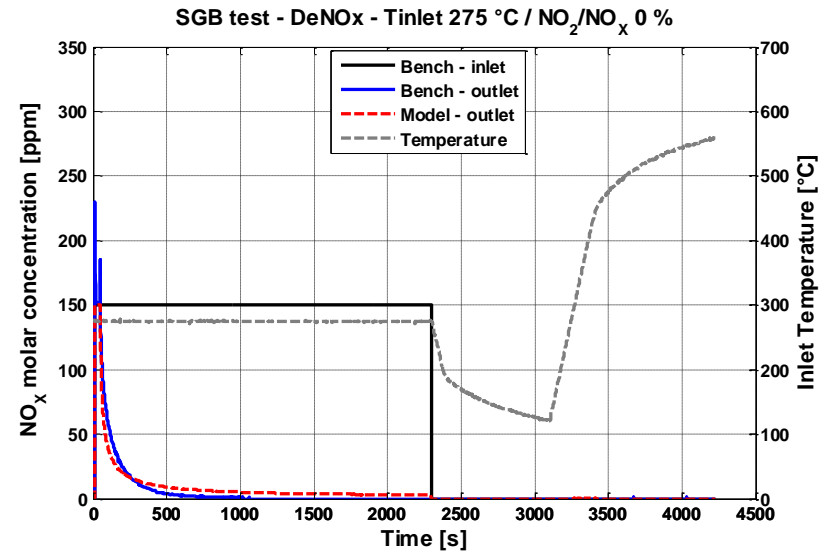
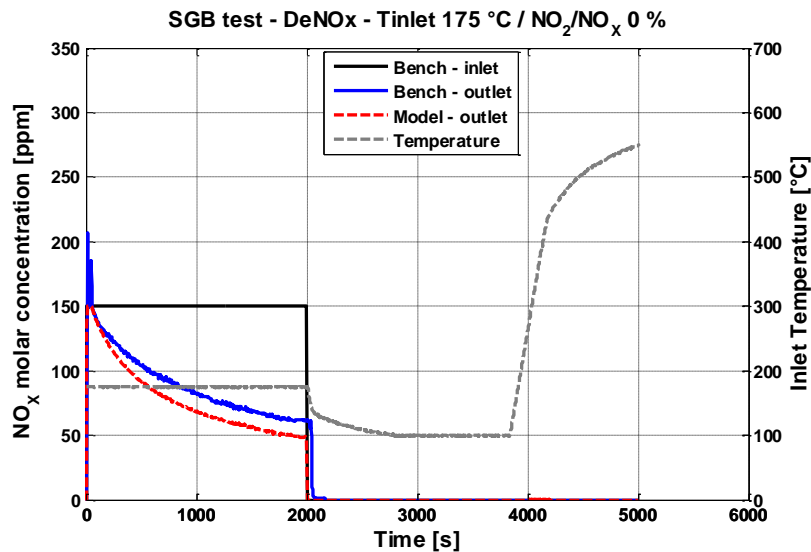
SCR model calibration for DeNOx reaction on SGB test

Standard SCR reaction calibration:



■ $r_{std} = k_{std} \times e^{-\frac{E_{std}}{R \times T_s}} \times [\text{NO}] \times \theta_{\text{NH}_3}$

■ Calibration is done on DeNOx SGB test with an upstream NO_2/NO_x ratio of 0 %



SCR model calibration for DeNOx reaction on SGB test

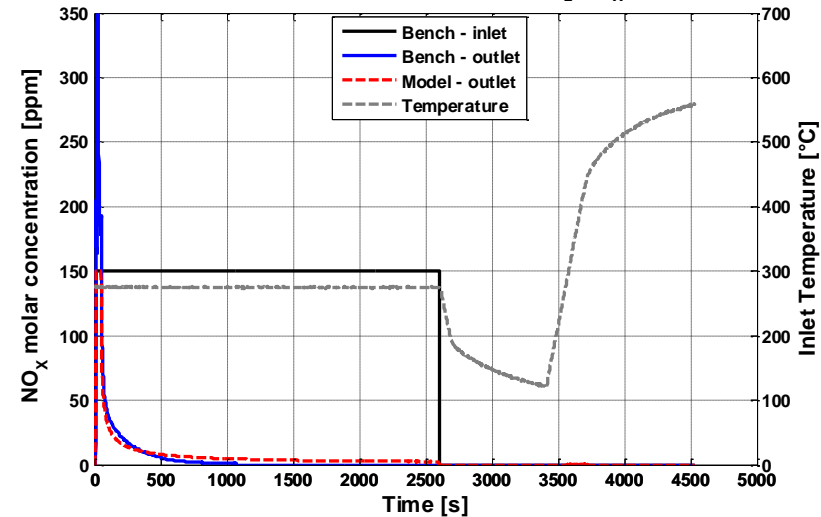
Fast SCR reaction calibration:

- $2 \text{NH}_3\text{-S}_i + \text{NO} + \text{NO}_2 \rightarrow 2 \text{N}_2 + 3 \text{H}_2\text{O} + 2 \text{S}_i$
- $r_{fast} = k_{fast} \times e^{-\frac{E_{fast}}{R \times T_s}} \times [\text{NO}] \times [\text{NO}_2] \times \theta_{\text{NH}_3}$
- calibration is done on DeNOx SGB test with an upstream NO_2/NO_x ratio of 50 %

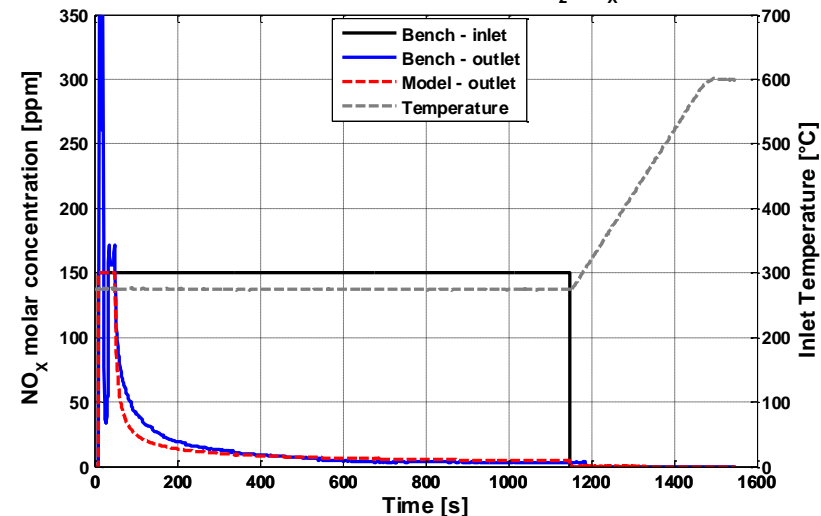
NO_2 SCR reaction calibration:

- $4 \text{NH}_3\text{-S}_i + 3 \text{NO}_2 \rightarrow 5,5 \text{N}_2 + 6 \text{H}_2\text{O} + 4 \text{S}_i$
- $r_{\text{NO}_2} = k_{\text{NO}_2} \times e^{-\frac{E_{\text{NO}_2}}{R \times T_s}} \times [\text{NO}_2] \times \theta_{\text{NH}_3}$
- calibration is done on DeNOx SGB test with an upstream NO_2/NO_x ratio of 100 %

SGB test - DeNOx - Tinlet 275 °C / NO_2/NO_x 50 %



SGB test - DeNOx - Tinlet 275 °C / NO_2/NO_x 100 %



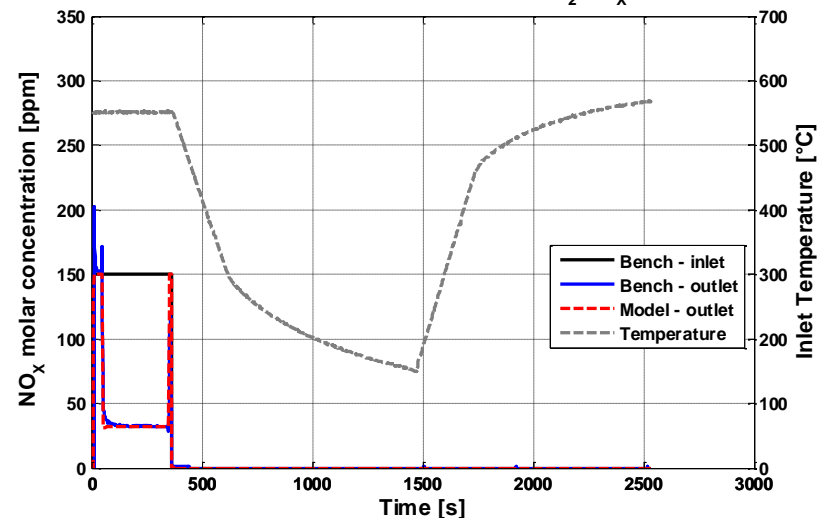
SCR model calibration for DeNOx reaction on SGB test

Ammonia oxidation reaction calibration:

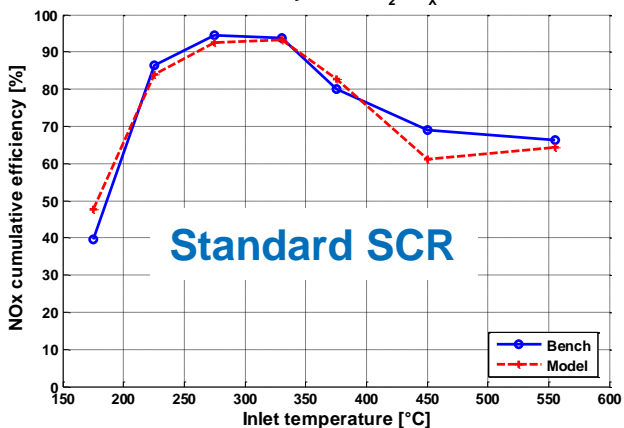
- 4 NH₃-S_i + 3 O₂ → 2 N₂ + 6 H₂O + 4 S_i
- $r_{NH3ox} = k_{NH3ox} \times e^{-\frac{E_{NH3ox}}{R \times T_s}} \times [O_2] \times \theta_{NH3}$
- calibration is done on DeNOx SGB test with an high upstream temperature

This calibration allows to obtain consistent results on the overall test data

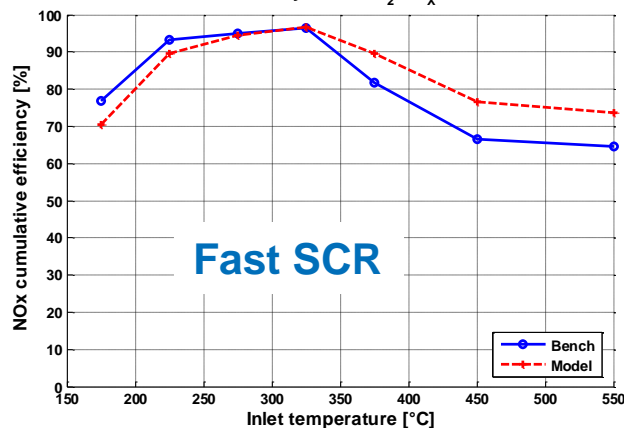
SGB test - DeNOx - Tinlet 555 °C / NO₂/NO_x 0 %



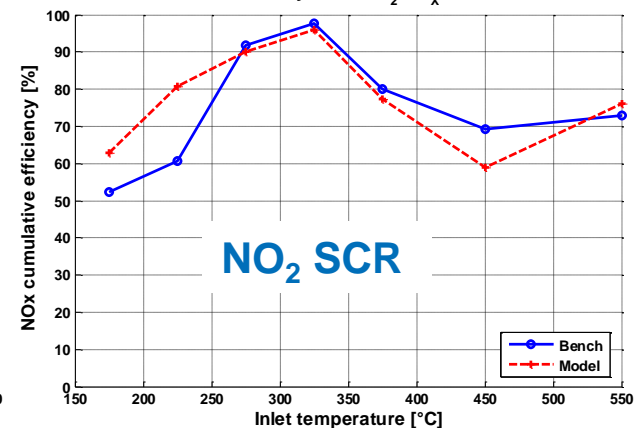
DeNOx efficiency - inlet NO₂/NO_x ratio 0 %



DeNOx efficiency - inlet NO₂/NO_x ratio 50 %

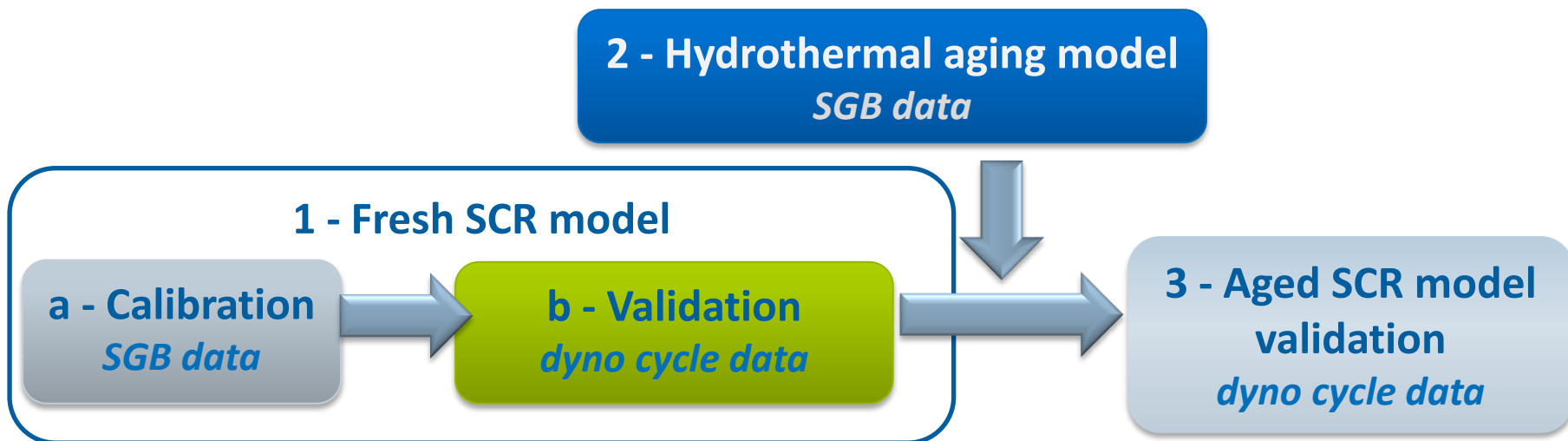


DeNOx efficiency - inlet NO₂/NO_x ratio 100 %





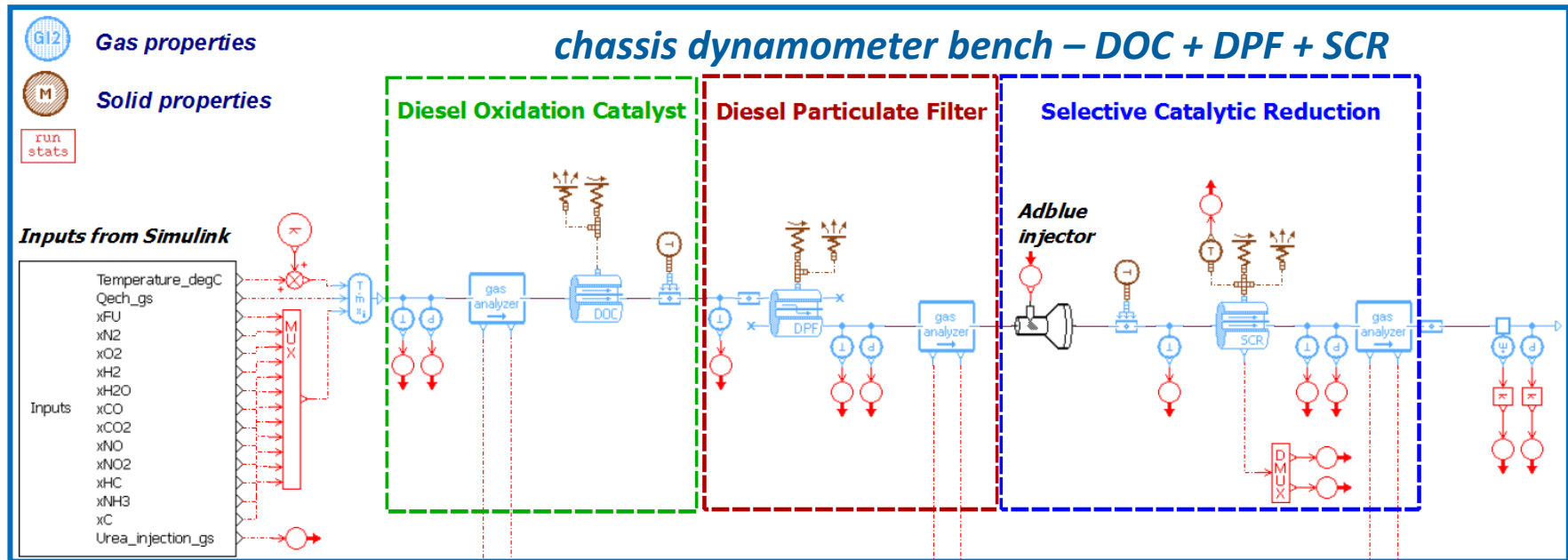
Approach: from fresh to aged SCR model



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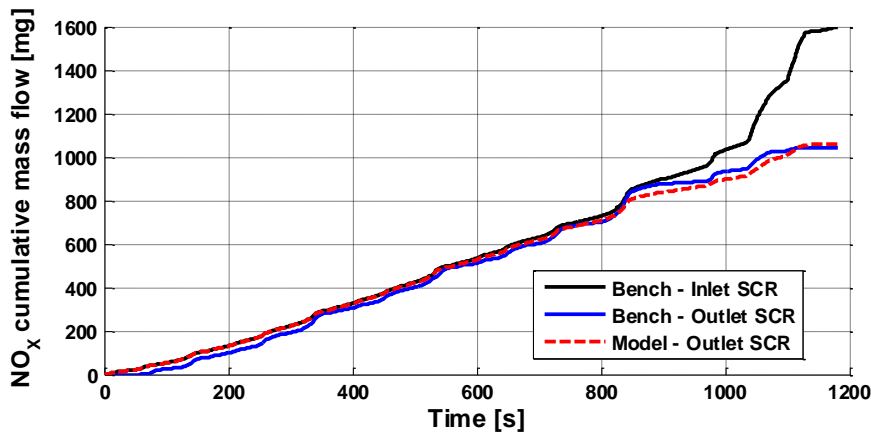
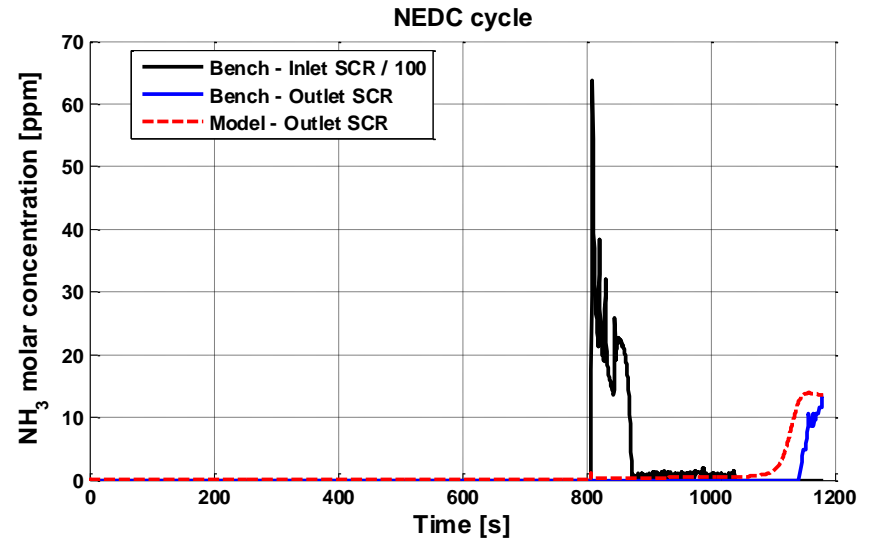
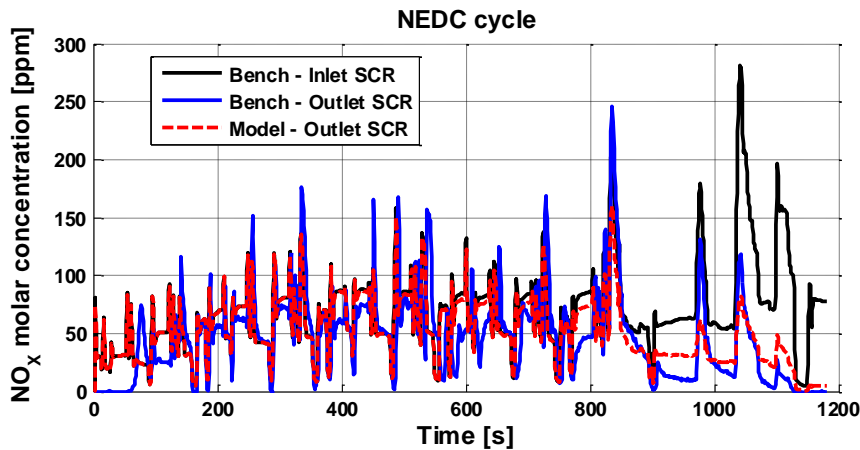
Exhaust line model presentation

- LMS Amesim model of a complete exhaust line composed of a DOC, a DPF and a SCR using IFP-Exhaust library coupled with a Simulink control platform



- Upgrade from SGB models:
 - convective and radiative heat transfer modeled
 - light adjustment of catalysts calibration

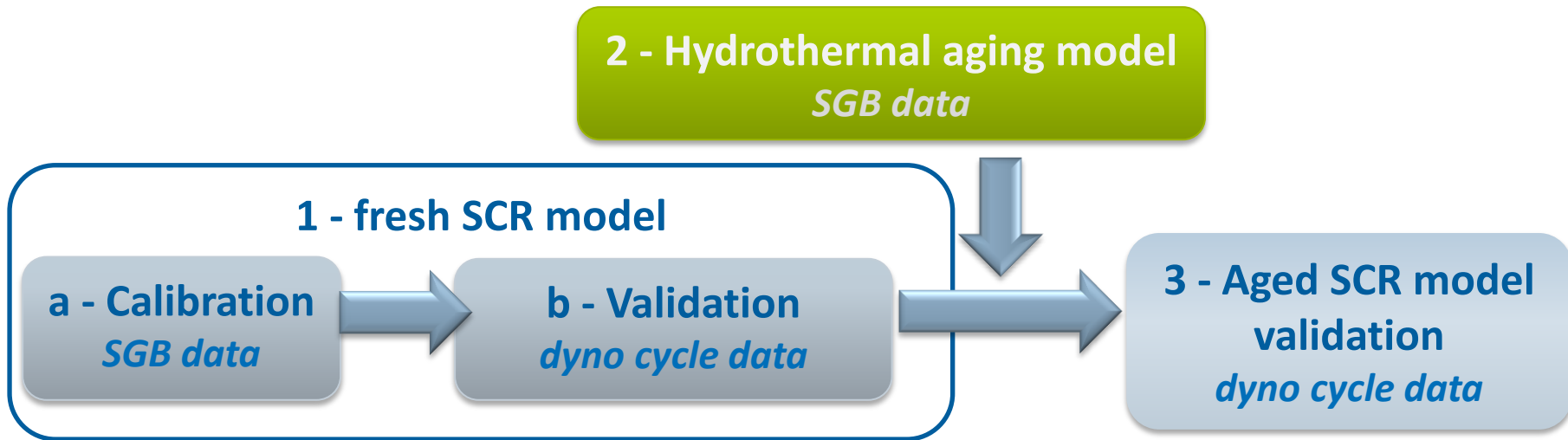
Fresh SCR cycle simulation results on NEDC cycle



- **NO_x reduction is well represented by the model**
- **Almost no NH₃ slip on this cycle with a fresh SCR**

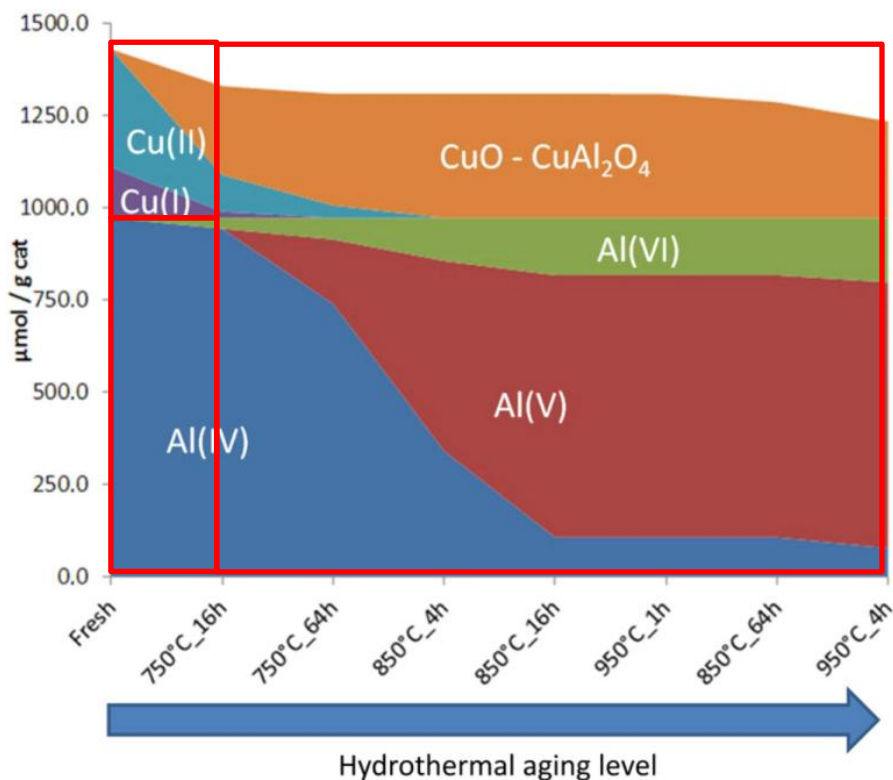


Approach: from fresh to aged SCR model



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Hydrothermal aging effects on SCR catalyst : Cu-Zeolite



■ Catalyst changes with aging:

- at low aging, copper migrates from exchanged positions to form copper oxide aggregates in the zeolite porosity
- during this phase, copper in exchanged positions protects the zeolite structure against dealumination
- with increasing aging, copper is completely removed from exchanged positions and dealumination occurs.

Ref.

« Hydrothermal aging effects on Cu-zeolite NH₃-SCR catalyst »

M. Valdez Lancinha Pereira, A. Nicolle, D. Berthout, IFPEN

[doi:10.1016/j.cattod.2015.03.027](https://doi.org/10.1016/j.cattod.2015.03.027)

Hydrothermal aging effects on SCR catalyst : Cu-Zeolite

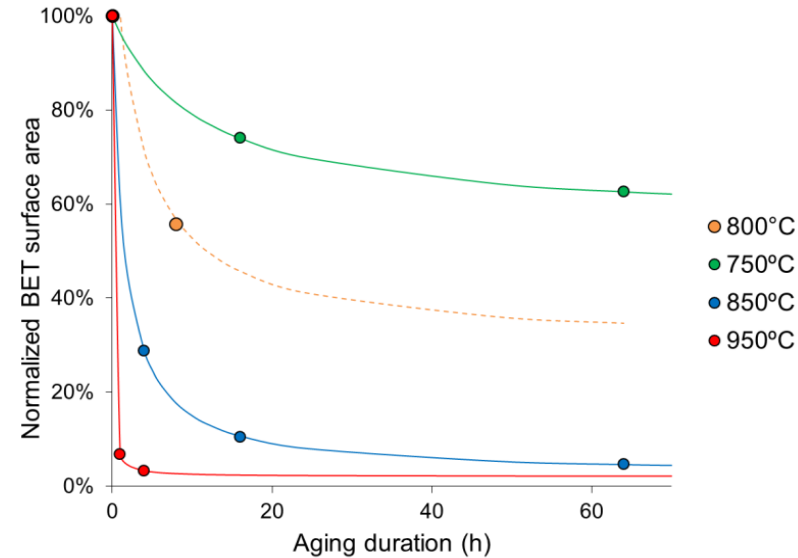
■ Aging model description:

- loss of copper and aluminum ions contributes to a loss of catalytic sites: modeled by a coefficient on active surface and on NH₃ capacities

$$S_{active\ aged} = Coeff_{Cu} \times S_{active\ fresh}$$

$$\Omega_{NH3site1\ aged} = Coeff_{site1} \times \Omega_{NH3site1\ fresh}$$

$$\Omega_{NH3site2\ aged} = Coeff_{site2} \times \Omega_{NH3site2\ fresh}$$

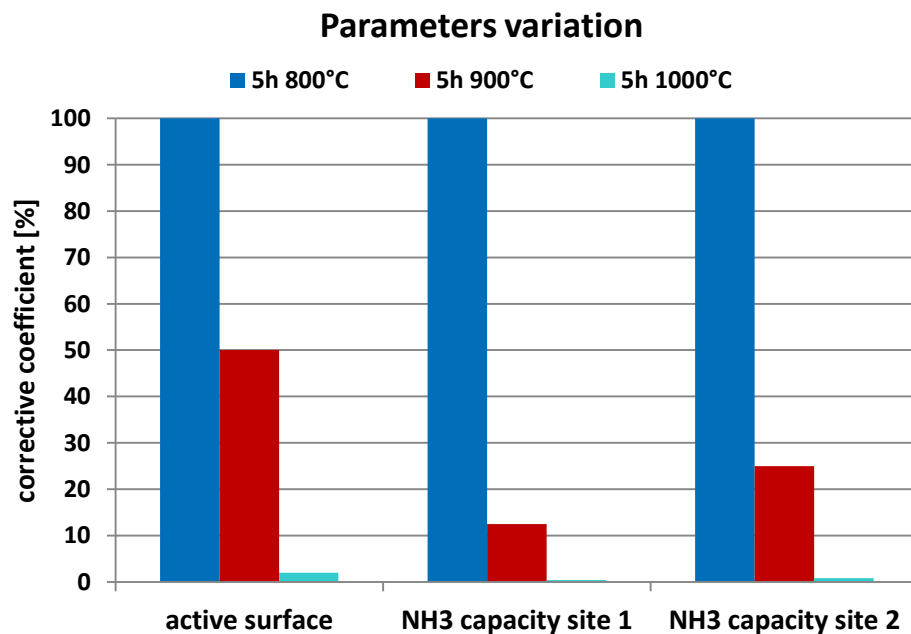
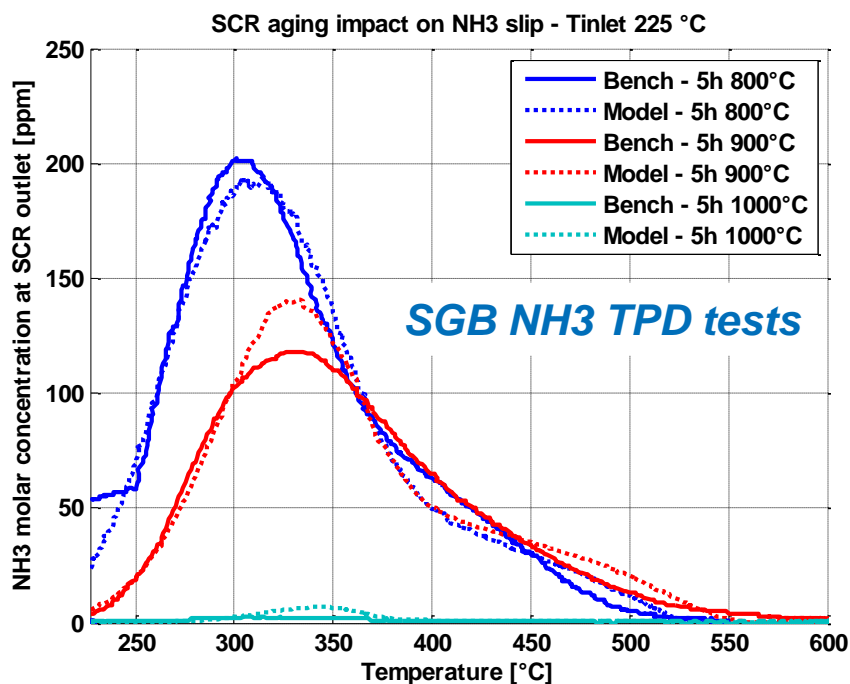


- CuO aggregates promote NH₃ oxidation: modeled by a coefficient on NH₃ oxidation activation energy

$$r_{NH3ox} = k_{NH3ox} \times e^{-\frac{Coeff_{CuO} \times E_{NH3ox}}{R \times T_s}} \times [O_2] \times \theta_{NH3}$$

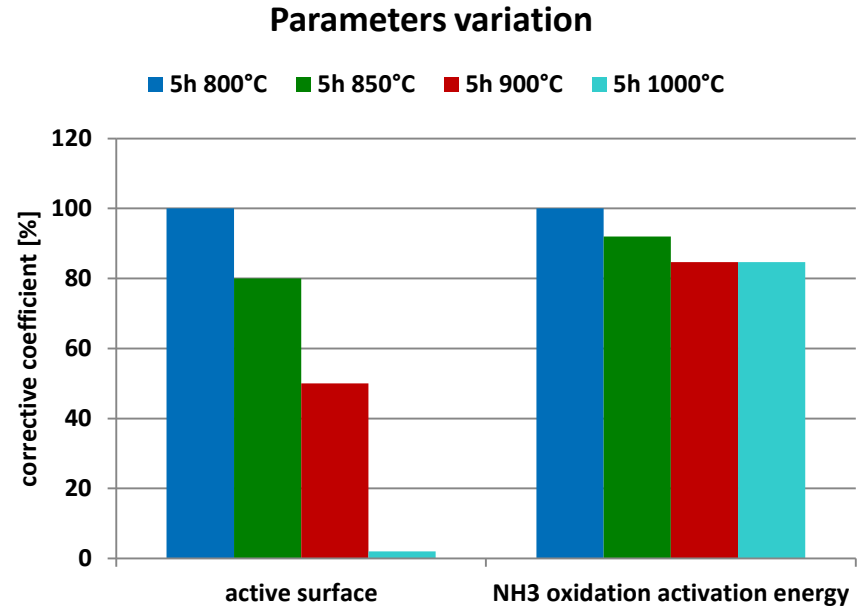
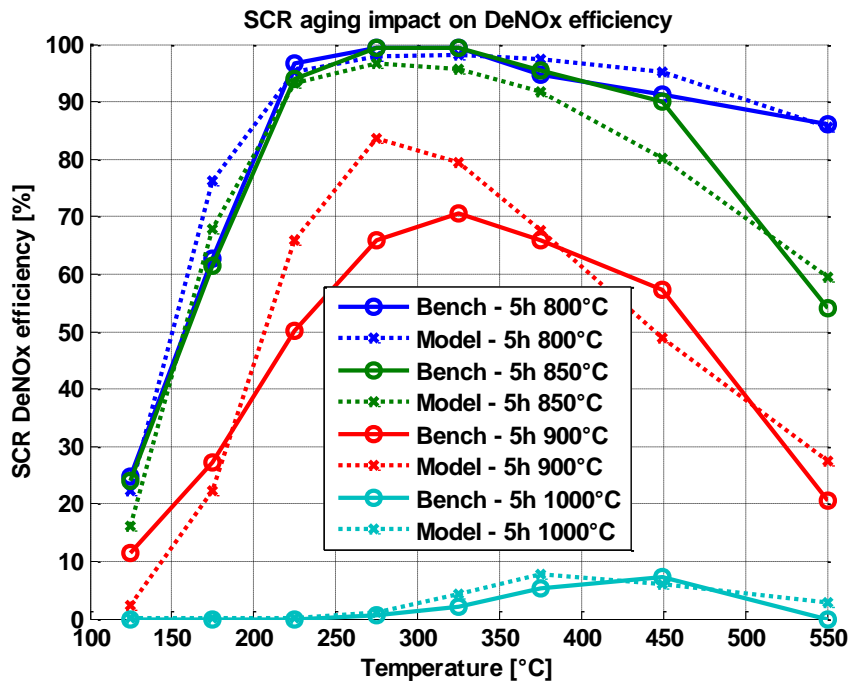
- no other changes in the calibration of the kinetic constants (k_j, E_{a,j}) determined on the fresh sample

Aging impact set up for NH₃ storage - Reactor scale



- NH₃ storage capacity reduction with aging is observed on SGB NH₃ TPD tests
 - hydrothermal impact on NH₃ capacity is higher on “low temperatures” NH₃ storage sites
- Desorption reaction rate reduction due to active surface reduction with hydrothermal aging temperature is well modeled

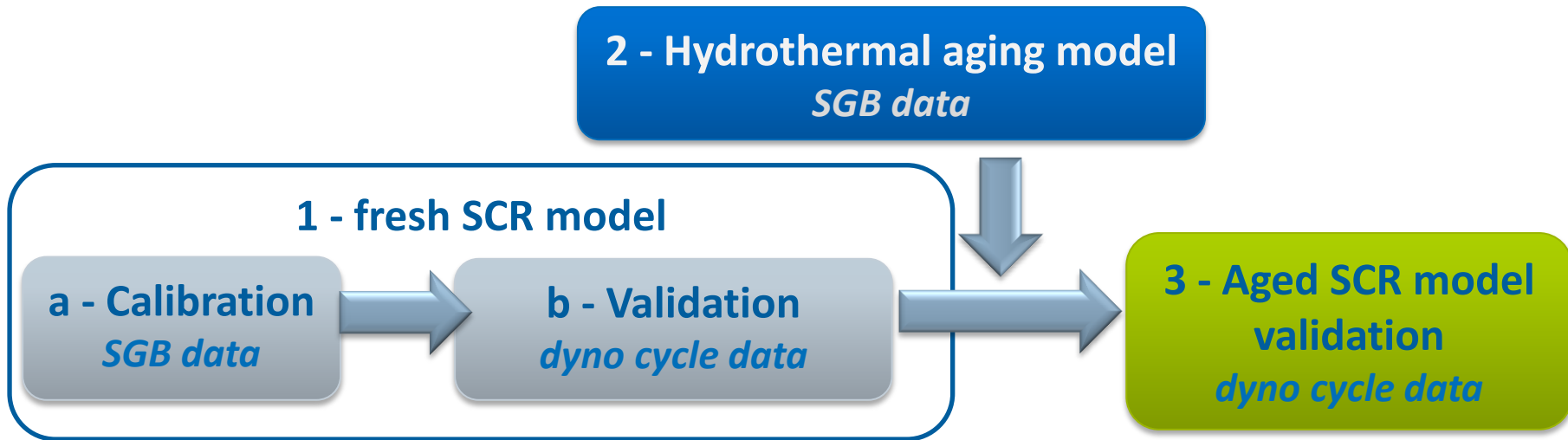
Aging impact set up for DeNOx reaction - Reactor scale



- At low hydrothermal aging temperature, main aging impact is a loss of catalyst efficiency at high temperature due on NH₃ oxidation
 - impact of CuO aggregates
- DeNOx reactions rates reduction is well represented by the reduction of the active surface in the model

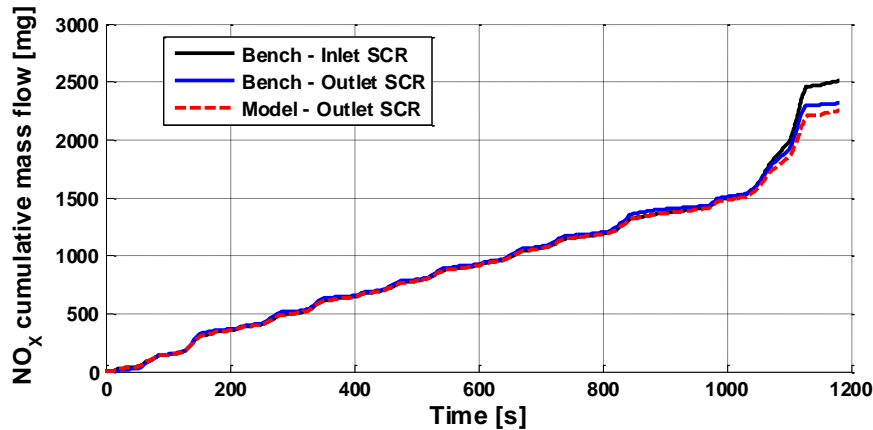
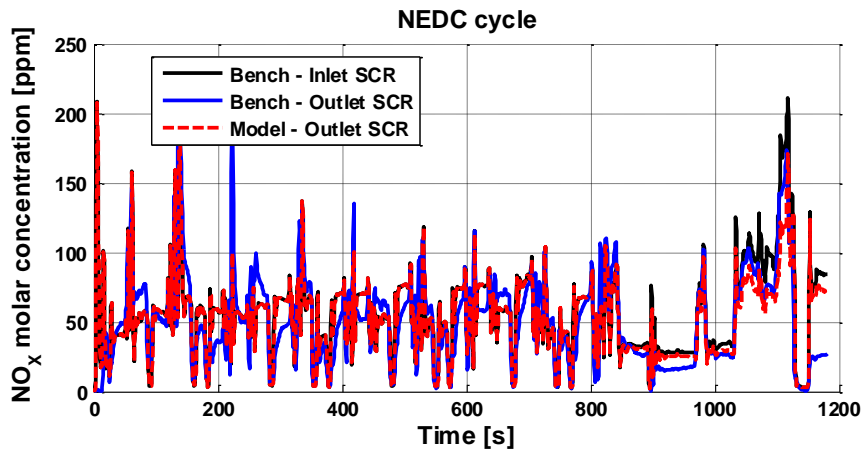


Approach: from fresh to aged SCR model



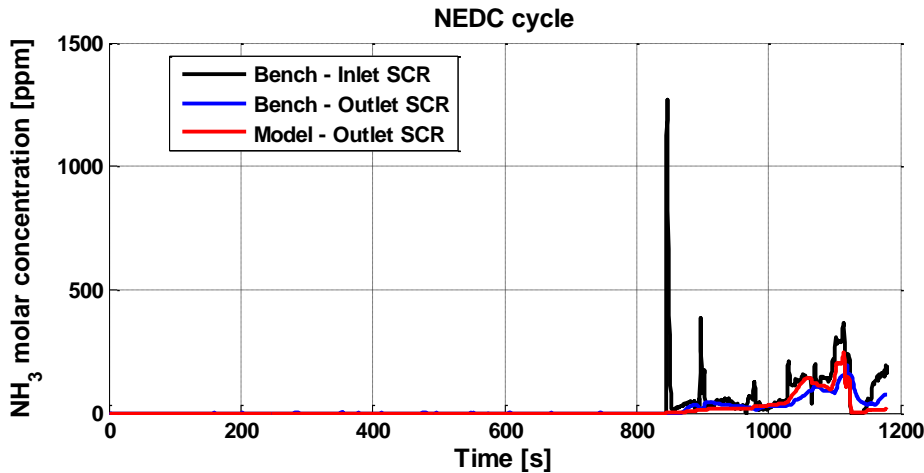
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5h 1000°C aged SCR cycle simulation results - NOx



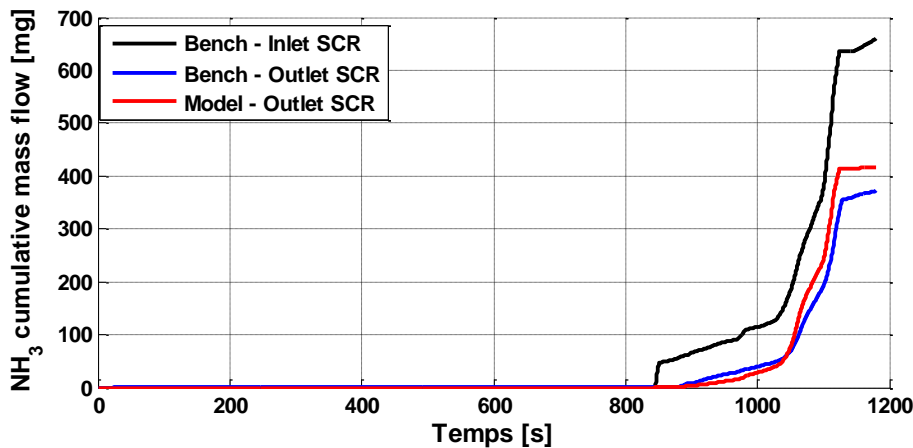
- **NEDC cycle is realized with an aged SCR**
 - hydrothermal aging of 5h at 1 000°C
- **Coefficients from previous aging model are used:**
 - Active surface: 2 %
 - NH₃ capacity of site 1: 0,5 %
 - NH₃ capacity of site 2: 1 %
 - NH₃ oxidation activation energy: 85 %
- **Impact on DeNOx efficiency is well modeled**

5h 1000°C aged SCR cycle simulation results - NH₃



- Impact on NH₃ slip is well modeled : right timing and level of release

- While NH₃ slip for the fresh SCR was almost inexistent, more than half of NH₃ injected quantity is released in the atmosphere for the aged SCR



➔ Injection strategy can be optimized



Summary

- **Simple hydrothermal aging model with reduced parameterization:**
 - one coefficient for active surface reduction
 - two coefficients for NH_3 storage capacities, close to active surface coefficient
 - one coefficient for NH_3 oxidation activation energy
 - **Need only SGB data**
 - **Consistency in results on NO_x and NH_3**

 - **Efficient methodology to model SCR aging impact from a fresh SCR calibration**
- ➔ ***This hydrothermal aging model will be integrated in IFP-Exhaust library for the next LMS Amesim release***



Innovating for energy

For further questions, please contact:

Simon Dosda

simon.dosda@ifpen.fr

Engine and Vehicle Modeling Department

IFP Energies nouvelles

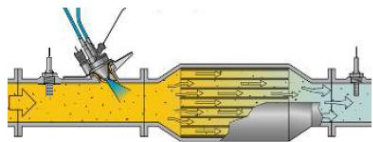
1 et 4 avenue de Bois-Préau

92852 Rueil-Malmaison Cedex - France

www.ifpenergiesnouvelles.com

Back up: IFP-Exhaust Library

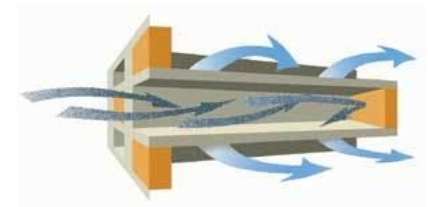
Library dedicated to the **after-treatment devices modeling**
MiL, SiL and HiL capabilities



Catalysts and Filters
(DOC, DPF, LNT, SCR, SCR-F & 3WC)



Pneumatic flow
(pipes, orifices, volumes.)



Heat transfer
(free and forced convection)

Sensors and Sources
(temperature, pressure, species)

