

Evaluation of HC, NO_x and NH₃ Mixed-Potential Sensors in Diesel and Gasoline Engine Exhaust

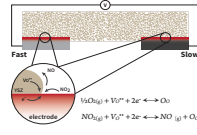
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Background

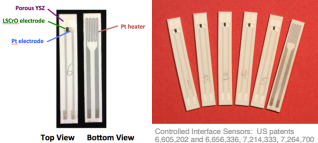
Mixed-Potential Sensors

- The non-equilibrium electric potential (EMF) developed in the presence of oxygen and a reducing/oxidizing gas is called a mixed potential.
- The magnitude of this mixed potential is a function of reducing/oxidizing gas concentrations (CO, HC, NH₃, NO_x).
- The mixed potential is dependent on the kinetics of the electrochemical reactions and is a function of the gas composition, electrode/electrolyte composition & morphology, and the operating temperature.

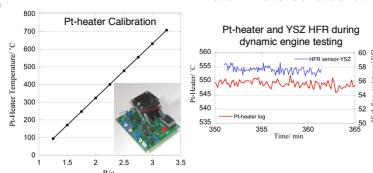


LANL Sensor design: Dense electrodes/porous electrolyte

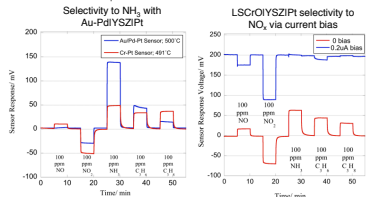
- Minimize heterogeneous catalysis: Minimize diffusion path through catalytically active electrodes.
- Avoid changes in morphology- Control the interface: Electrolyte morphology stable and reproducible.



- Planar design with sensor operating temperature controlled by resistance of Pt-heater
- Heater-board developed with Custom Sensor Solutions controls Pt-heater resistance with control-loop feedback electronics
- Controlling Pt-heater resistance appears adequate for maintaining constant sensor element temperature as shown by constant high frequency resistance of YSZ electrolyte

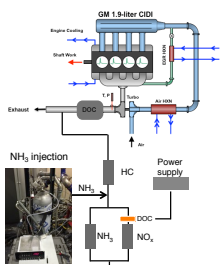


- Exploits differences in O₂ reduction kinetics of electrode materials by using dissimilar electrodes
 - Common platform allows selectivity to be altered by changing electrode materials only
- Interfacial resistance dominates: Sensor response dominated by electrochemical reactions.
 - Allows selectivity to be tuned by operation under a current bias

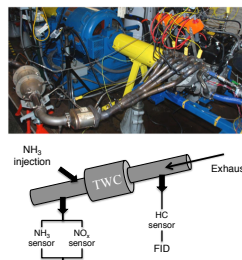


Dynamometer Sensor Testing

GM 1.9-liter CIDI

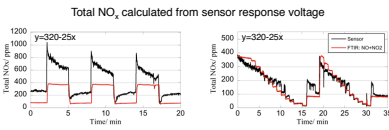
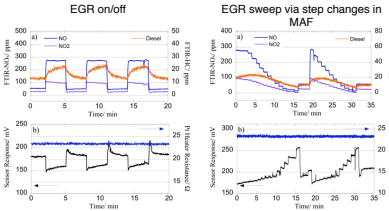


BMW 120i lean gasoline engine



Diesel Exhaust

NO_x Sensor

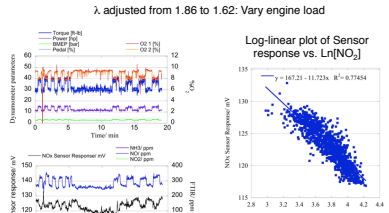


- NO_x sensor qualitatively tracks step changes in NO and NO₂
- Log-linear curves exhibit similar slopes with offset in Y-intercept between different engine experiments as well as laboratory calibration.
- Y-intercept likely correlated to [THC] or other interferences but not in trivial (linear) way

Gasoline Exhaust

NO_x Sensor

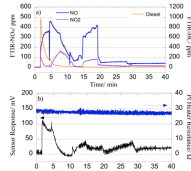
- NO_x sensor response during λ switching, sampling downstream of TWC.



- Raw data with no averaging or signal processing
- NO_x sensor exhibits log-linear correlation to [NO_x]

HC Sensor

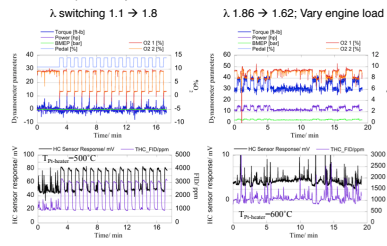
- Data collected during engine start-up showing large variation in HC (Diesel) and total NO_x measured via FTIR.
- Sensor response qualitatively tracking HC content.
- Constant Pt-heater resistance shows that sensor element was maintained at temperature set-point under dynamic operation.



- Calibration curve generated from averages of steady-state sensor response data collected during engine testing.
- Transients in HC content tracked by sensor voltage response.
- When sensor is sampling Engine-out exhaust, response is noisy. Calculated Diesel concentration from sensor response ±40ppm FTIR measurement.
- DOC out sensor signal less noisy, but sensor over predicts Diesel content by as much 50ppm compared to FTIR measurement.
- Explanation for discrepancies not clear at this time.
 - Change in HC speciation as exhaust gas transports to instruments?
 - Non-trivial influence of inter-ferent species?

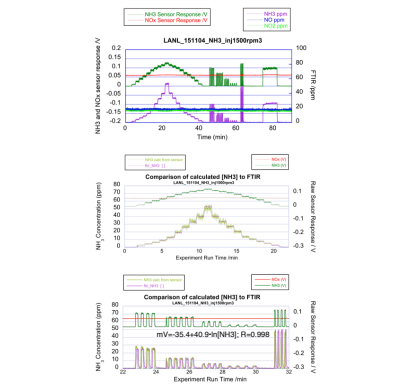
HC Sensor

- HC sensor response during λ switching, sampling upstream TWC. Exhaust from sensor sent to FID for THC analysis.
- Sensor qualitatively tracks THC measured via FID.



NH₃ Sensor

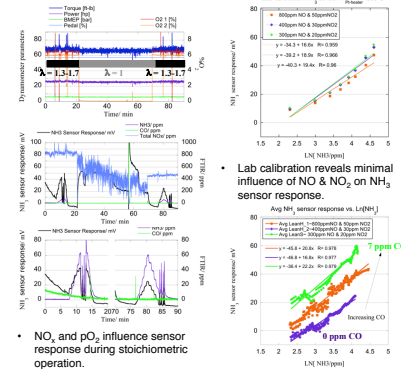
- Steady-state engine operation with NH₃ injected upstream of sensors



- Bench reactor data was used to generate a calibration curve for NH₃.
- NH₃ sensor clearly tracks injected ammonia.
- NO_x sensor (RED line) reflects steady state NO/NO₂ levels during injection of ammonia.

NH₃ Sensor

- Sensor response during lean-homogeneous and stoichiometric engine operation.
- Comparison of lab calibrations to sensor response in exhaust gas.



- Lab calibration reveals minimal influence of NO & NO₂ on NH₃ sensor response.
- Aug NH₃ sensor response vs. Ln[NO_x]
- Baseline shift between varying NH₃ staircases: CO, even at low levels (<10 ppm), is likely the dominant interferer species.

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