

# DAIMLER

## System Simulation of Advanced SCR-Systems

11th CLEERS Workshop

15.03.2008

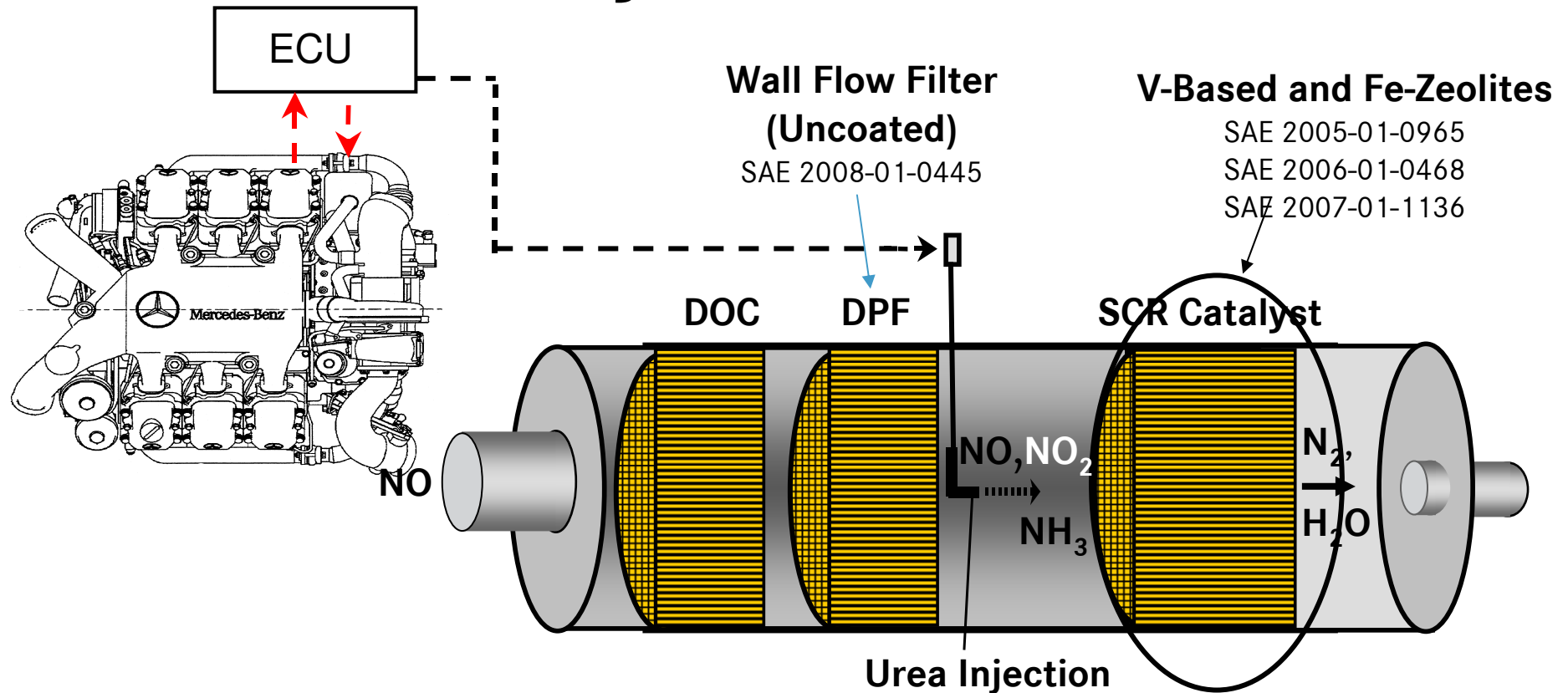
Daniel Chatterjee, Thomas Burkhardt, Stefan Schöffel

and Michel Weibel

# Outline

- Advanced SCR-System Setup (DOC+DPF+SCR)
- Simulation Concept
- Modeling SCR Converters
- Modeling DOC Converters
- DOC: Kinetic Correlations
- Simulation Based System Optimization: DOC Impact on SCR
- Modeling of Urea Processing
- Conclusions

# Advanced SCR-System: DOC+DPF+SCR



- ➔
- $\text{NO}_2$  increases  $\text{NO}_x$  reduction efficiency and complexity of the system.
  - High thermal mass in front of the SCR delays  $\text{NO}_x$  reduction lightoff.
  - Adaptation of the DOC+DPF+SCR-System to the individual application is required.
  - Optimization target: High  $\text{NO}_x$  conversion efficiency without  $\text{NH}_3$  slip.

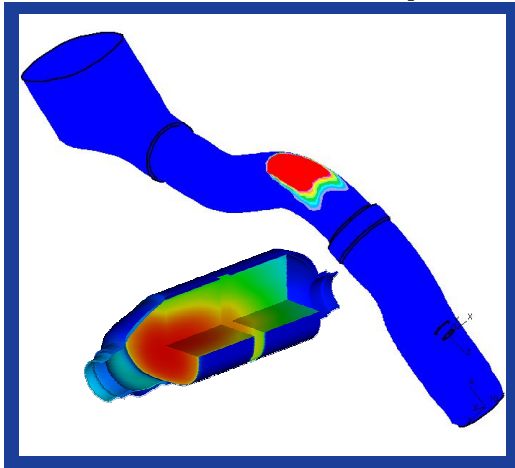
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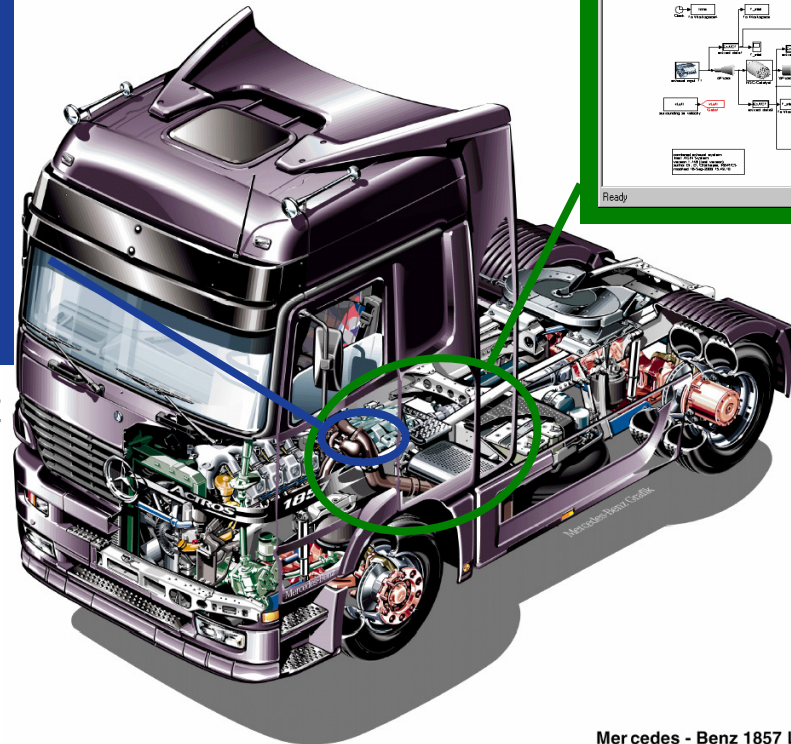
# Exhaust Gas Aftertreatment Simulation

## 3D-Simulation:

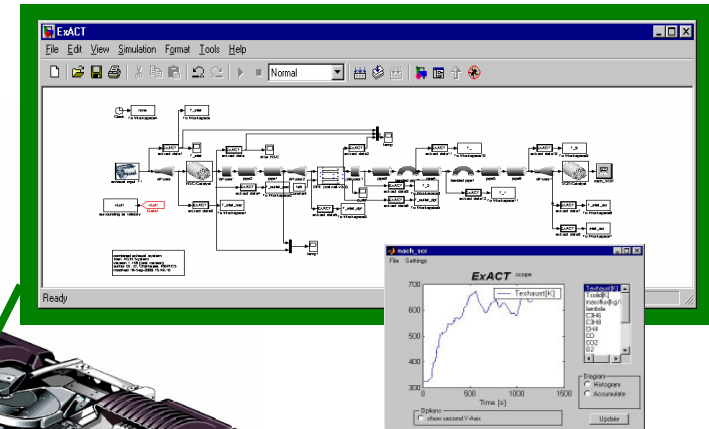
Optimization of Geometry for Individual Components



- Effect of non uniform inlet conditions
- Urea Processing



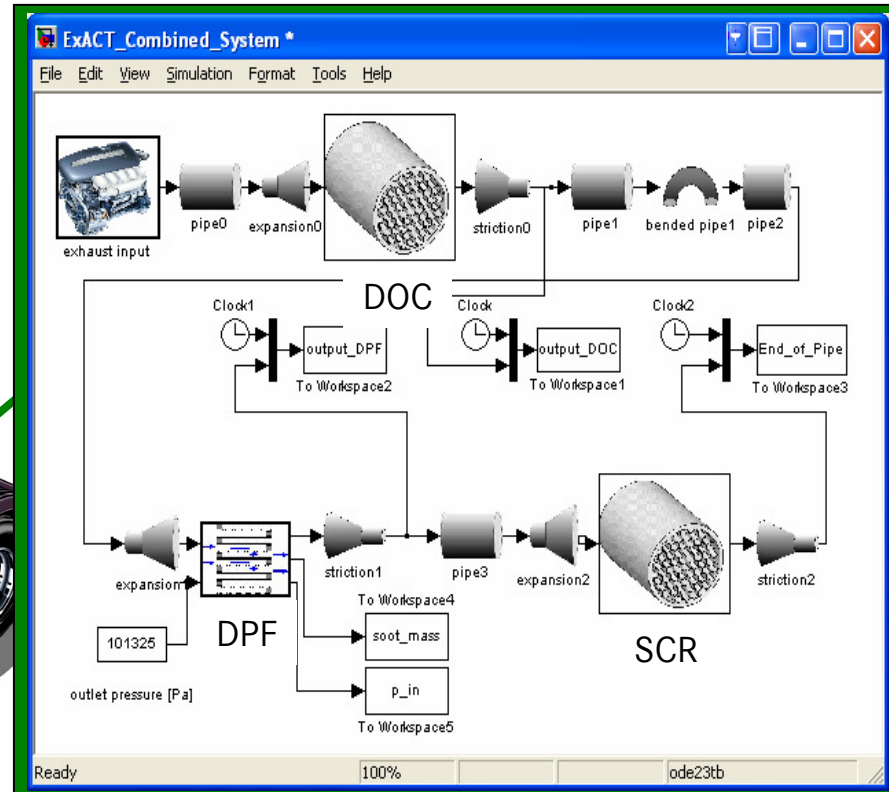
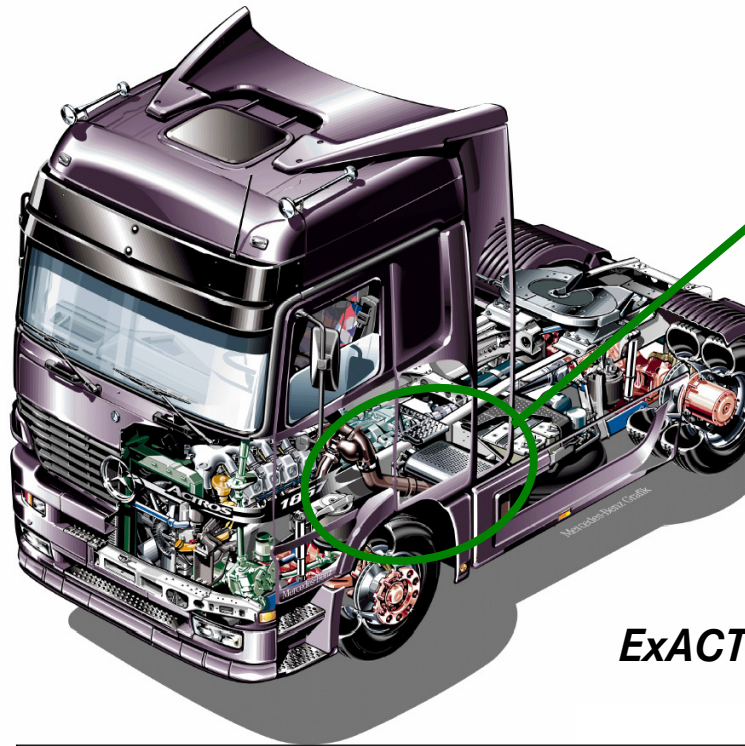
## ExACT: 1D Simulation of Combined Exhaust Aftertreatment Systems



- Testcycle Performance
- System-design
- Operating strategies

Mercedes - Benz 1857 LS

# Simulation Toolbox *ExACT*

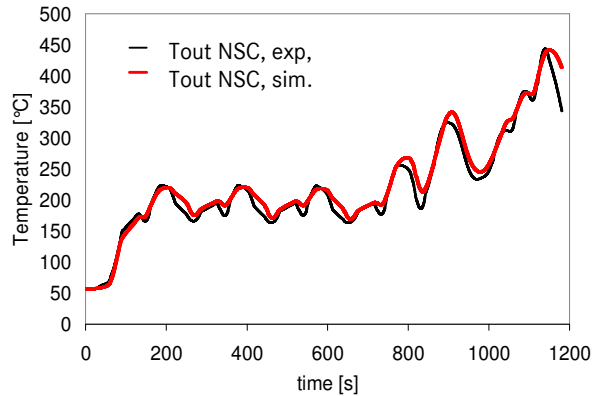


***ExACT*** = ***Ex***haust Gas ***A***ftertreatment ***C***omponents ***T***oolbox

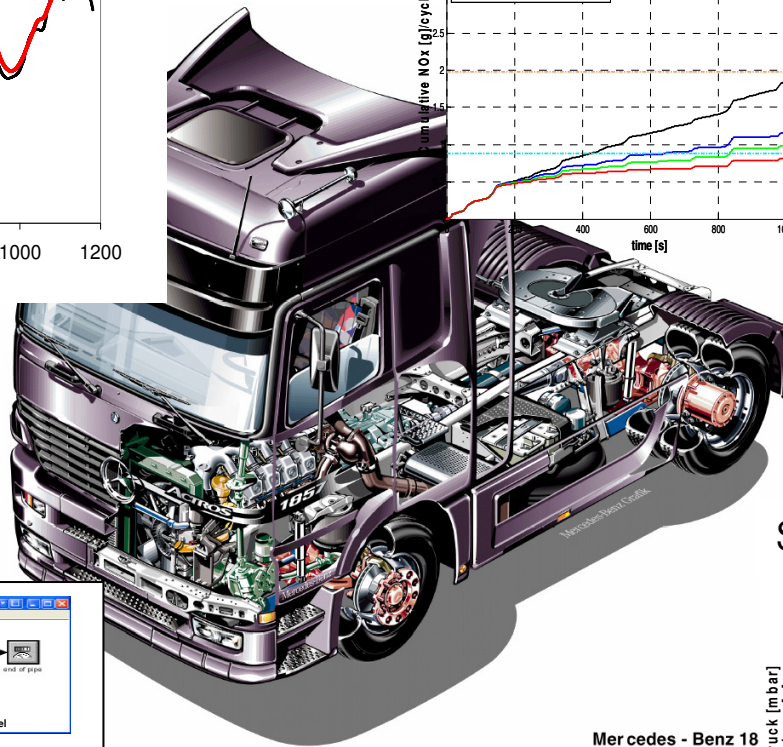
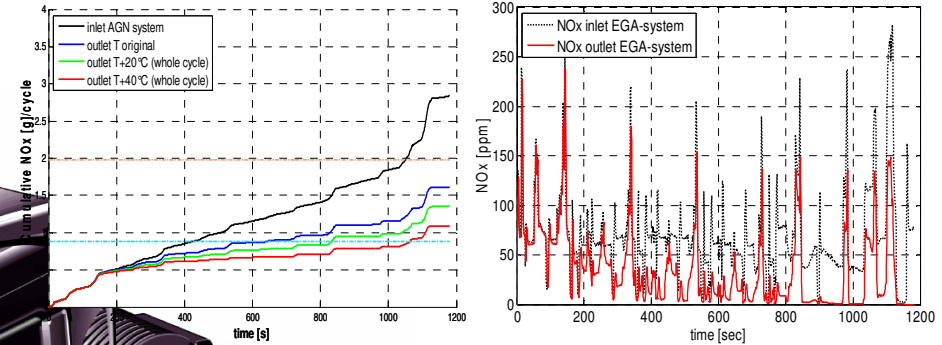
- ExACT toolbox consists of 1D models for SCR, ASC, DOC, DPF, LNT, TWC.
- Model generation by drag&drop.
- Focus: Testcycle simulation, system design, operating strategies.

# ExACT Exhaust Aftertreatment System Modeling: Application Examples

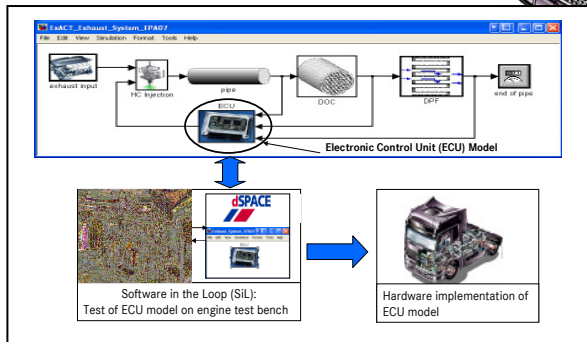
Temperatures



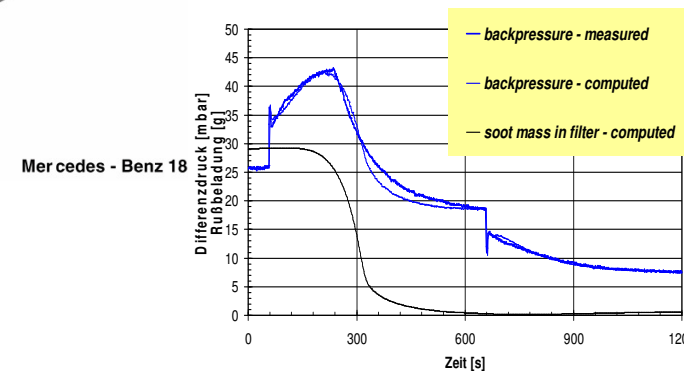
Prediction of Soot, NOx, CO, HC Conversion



ECU-Development „Virtual Testbench“



Soot Loading/Pressure Loss



# Modeling SCR and DOC Catalytic Converters

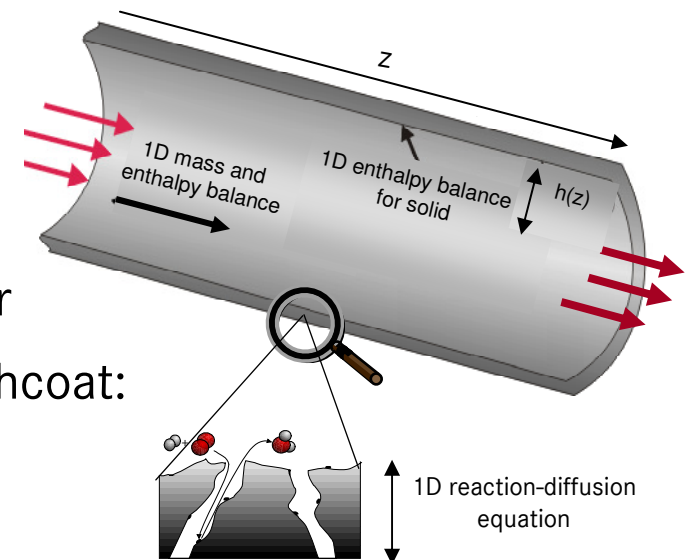
## Transport Processes:

- Modeling of one representative channel.
- 1D mass and enthalpy balances for gas and solid.
- Gas-solid mass and heat transport by means of transfer coefficients,  $h(z)$ .
- SCR :1D reaction-diffusion equation to account for diffusional limitation within the catalytic wall/washcoat:

$$0 = D_{eff,j} \frac{\partial^2 C_j^*}{\partial x^2} + S_W^2 R_j$$

- DOC: Effectiveness factor/Thiele modulus

$$\eta = \frac{\tanh(\phi)}{\phi} \quad \phi = L \sqrt{\frac{R_4}{D_{eff} c_{NO}}}$$



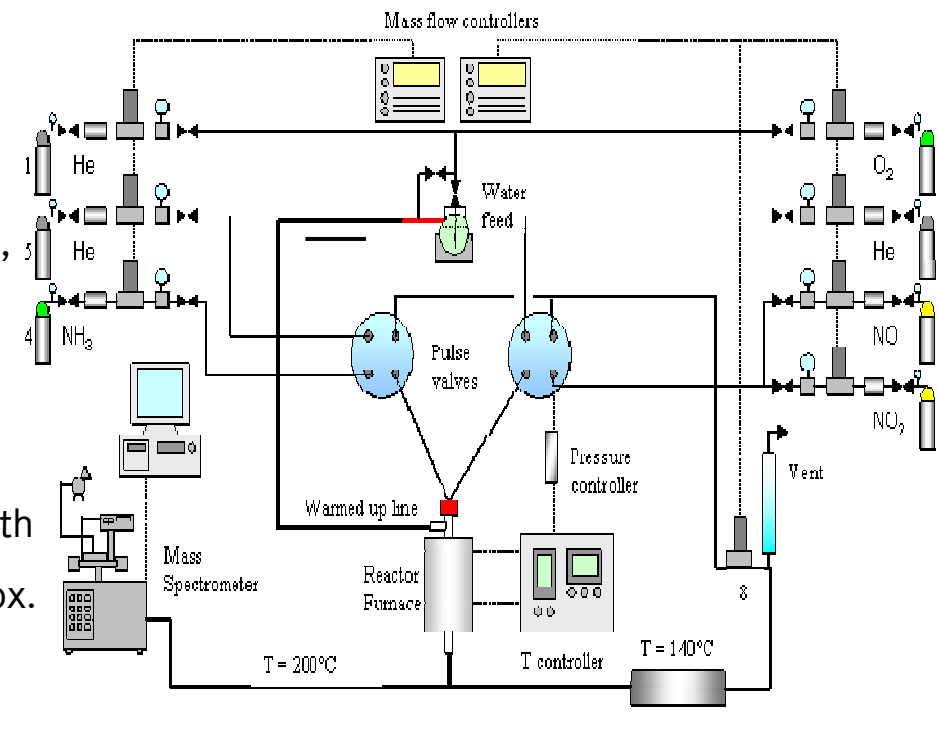


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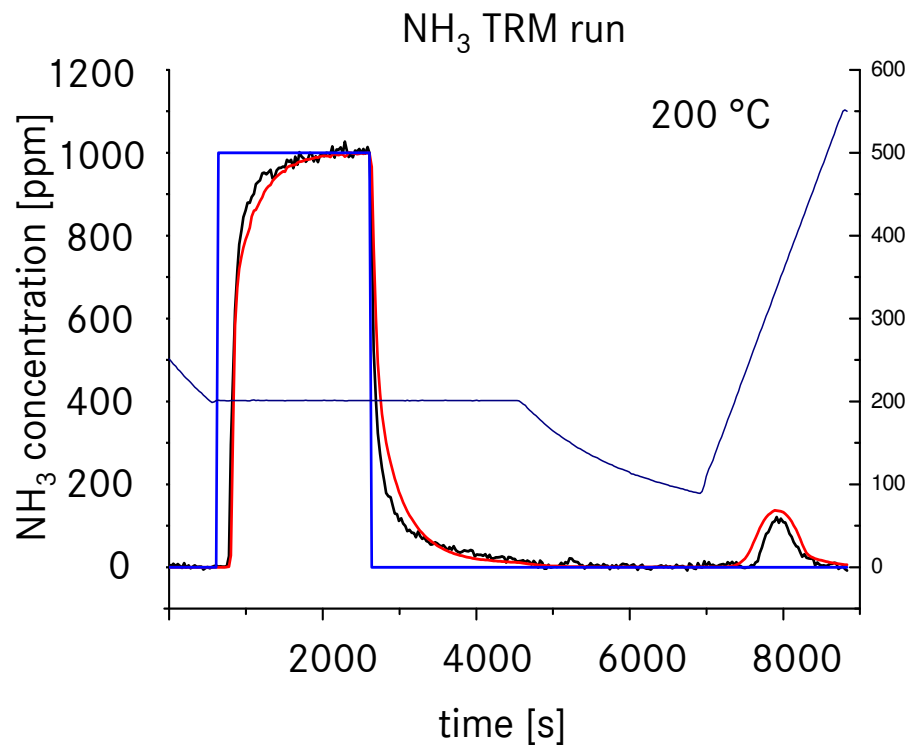
# Modeling SCR-Catalytic Converters: Intrinsic Transient SCR Kinetics

- Catalyst samples taken from commercial SCR catalysts.
- Kinetic investigations with TRM experiments in a flow-microreactor with powdered catalyst.
- Typical concentrations: 1000ppm NO<sub>x</sub> and NH<sub>3</sub>,  $0 \leq \text{NO}_2/\text{NO}_x \leq 1$ , 2% O<sub>2</sub> and 1% H<sub>2</sub>O.
- Temperature range: 50-450°C (NH<sub>3</sub> ads./des., NH<sub>3</sub> ox.), 160-450°C (SCR react.)
- Kinetic parameter estimation of subsystems with the respective TRM runs. (NH<sub>3</sub> ads./des., NH<sub>3</sub> ox. NO-SCR react., NO+NO<sub>2</sub> react.)

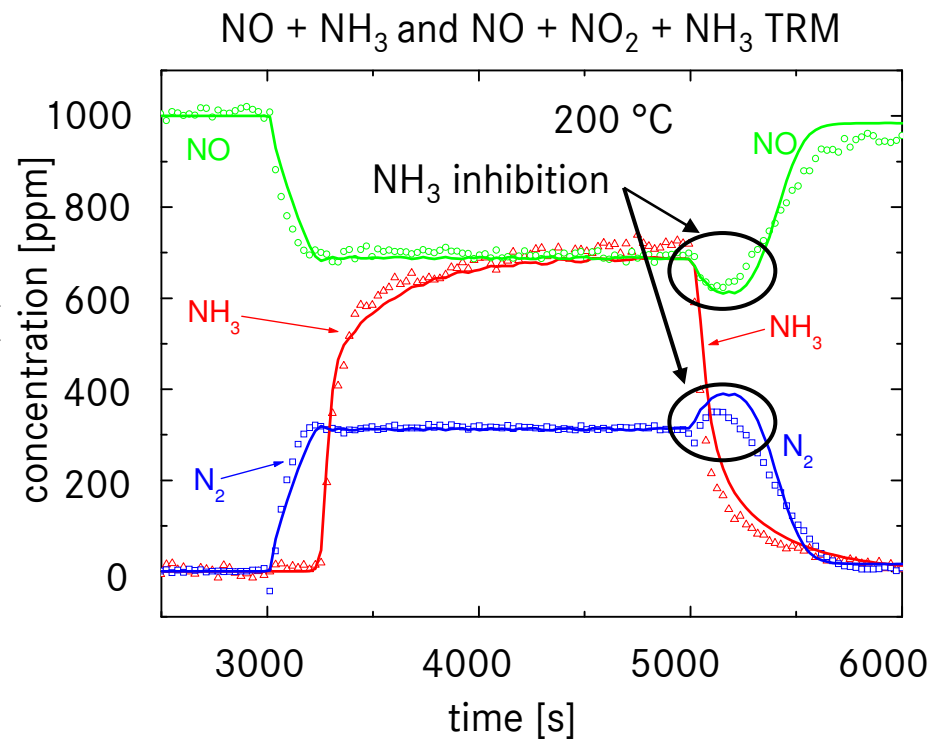


➔ • Experimental set up avoids the influence of transport effects on kinetic measurements.  
 • Sequential fitting of kinetic subsystems minimizes parameter correlations.

# Modeling SCR-Catalytic Converters: Calibration of Reaction Kinetics by TRM Experiments

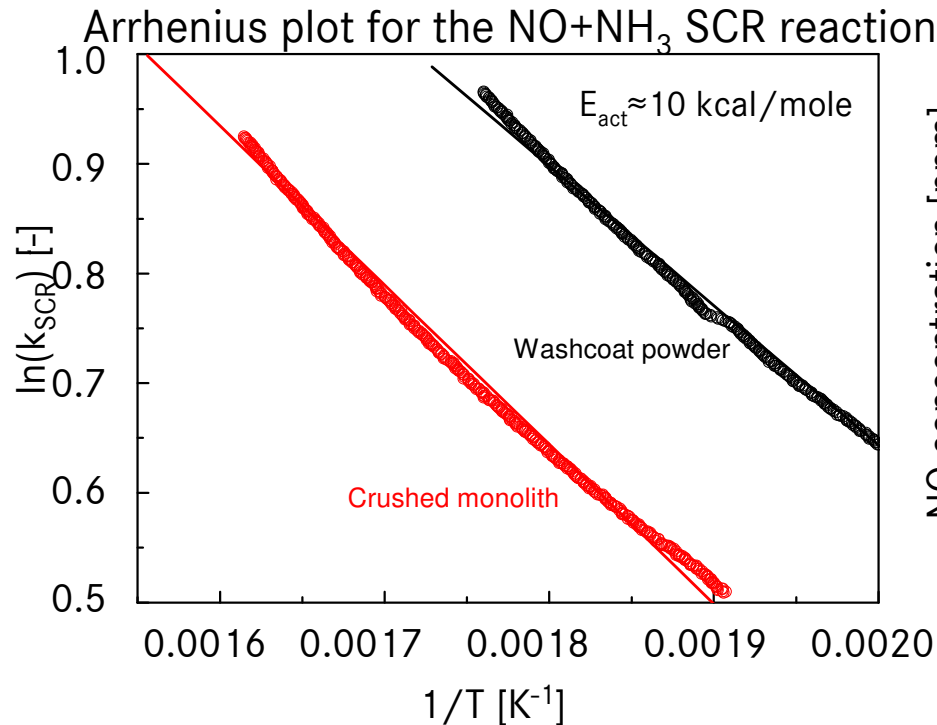


Parameters for NH<sub>3</sub> Adsorption/Desorption,  
NH<sub>3</sub>-Oxidation

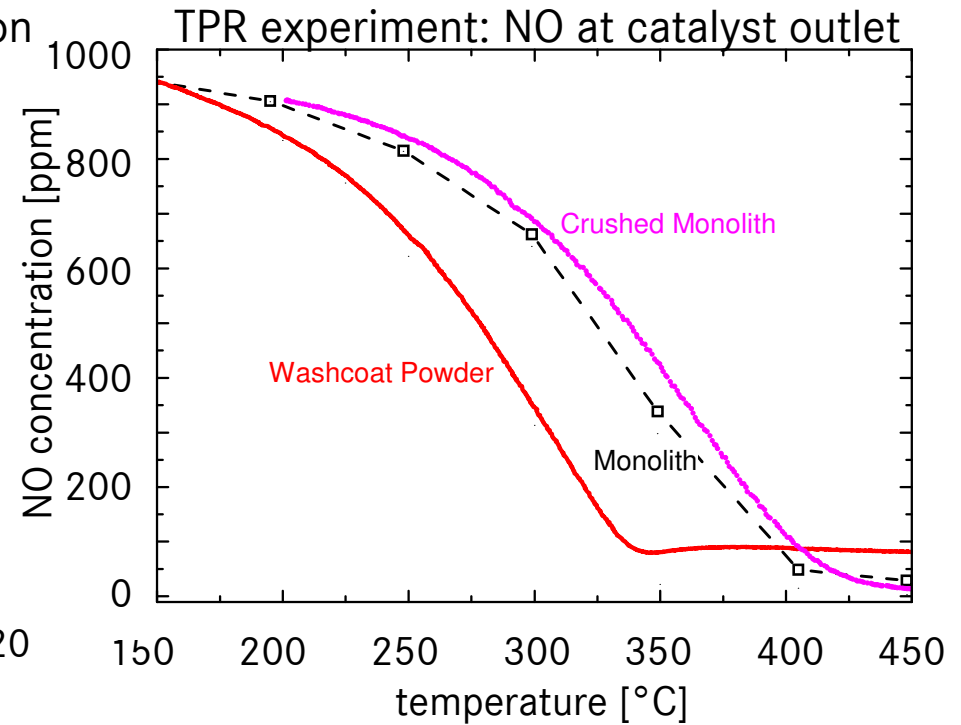


Parameters for NO-SCR Reaction

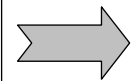
# SCR Kinetic Measurements: Zeolith Washcoat Powder vs. Crushed Monolith



Feed: 1000 ppm NH<sub>3</sub>, 1000 ppm NO, 1% H<sub>2</sub>O, 2% O<sub>2</sub>, balance N<sub>2</sub>,  
**same catalytic active mass** GHSV 170000 Ncm<sup>3</sup>(gh)<sup>-1</sup>

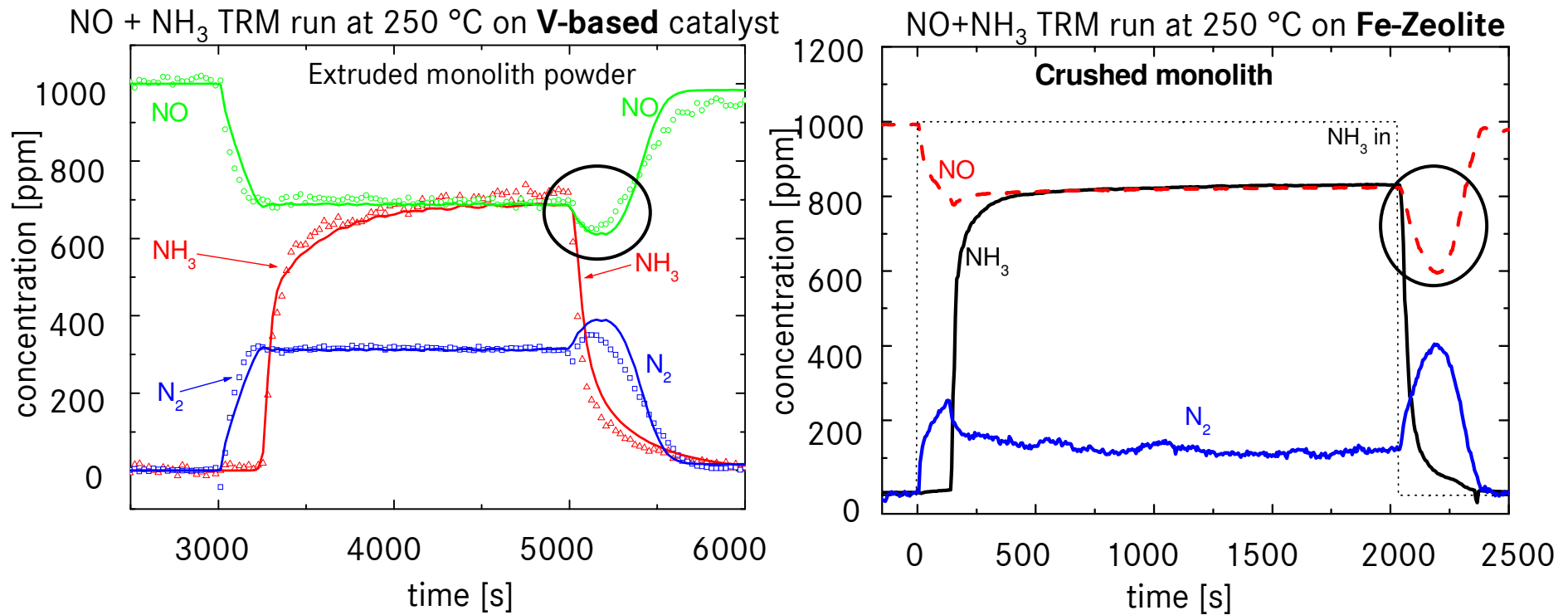


Feed: 1000 ppm NH<sub>3</sub>, 1000 ppm NO, 1% H<sub>2</sub>O, 2% O<sub>2</sub>, balance N<sub>2</sub>,  
**same catalytic active mass** GHSV 177000 Ncm<sup>3</sup>(gh)<sup>-1</sup>



- Washcoat powder reveals a higher SCR activity compared to the monolith.
- NH<sub>3</sub> storage and NH<sub>3</sub> oxidation are more efficient on powder.
- Experiments on crushed monoliths are quantitatively more representative.

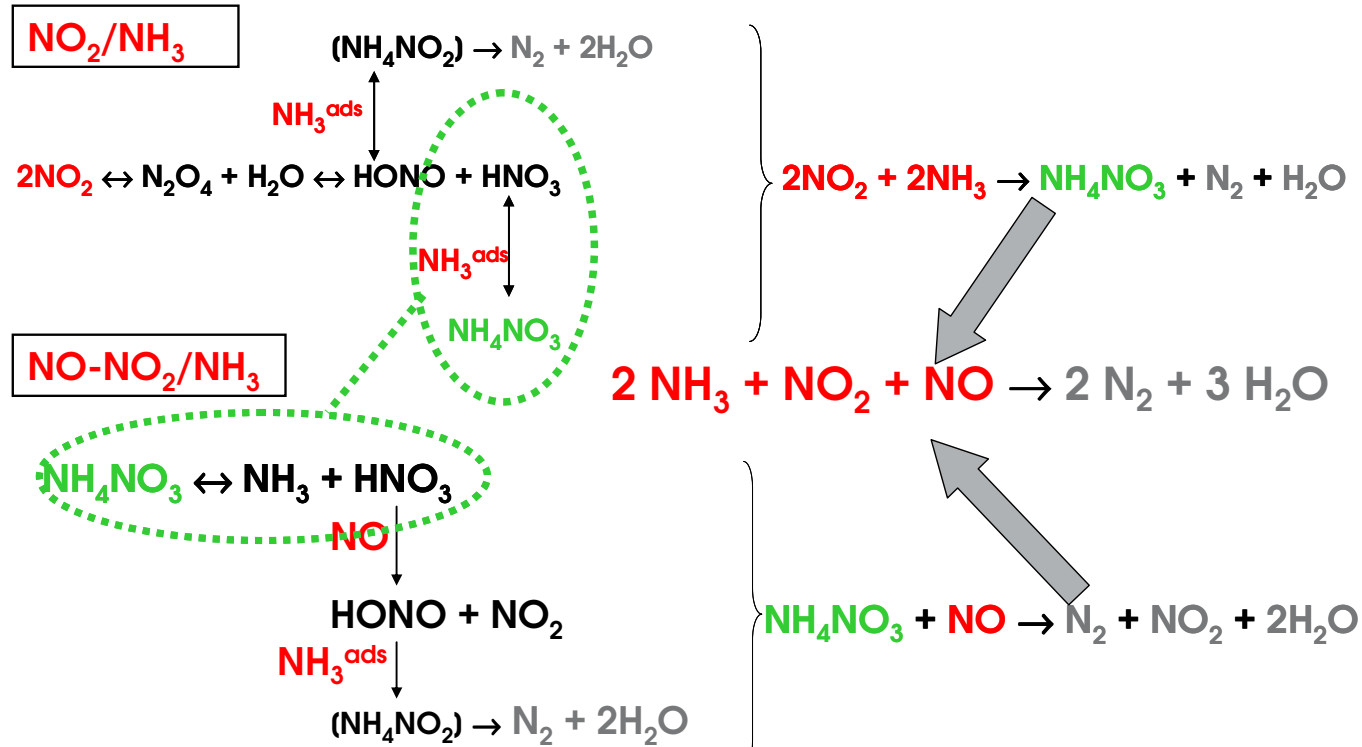
# SCR Kinetic Measurements: Zeolite vs. V-Based Catalyst



- ➔
- Kinetic experiments reveal similar features on V-based and Fe-Zeolite SCR catalysts.
  - NH<sub>3</sub> inhibition effect is more pronounced on Fe-Zeolite.
  - Experimental results suggest the use of similar global reaction rate expressions.

# Modeling SCR-Catalytic Converters:

Proposed Reaction Mechanism for  $\text{NO}_2 + \text{NO} + \text{NH}_3$  on **V-Based** and **Fe-Zeolites** SCR Catalysts:



C. Ciardelli, I. Nova, E. Tronconi, D. Chatterjee, B. Bandl-Konrad, *Chem. Commun.* **23** (2004), 2718,  
 D. Chatterjee, T. Burkhardt, M. Weibel, E. Tronconi, I. Nova, C. Ciardelli, SAE 2006-01-0468,  
 I. Nova, C. Ciardelli, E. Tronconi, D. Chatterjee, B. Bandl-Konrad, *Catal. Today*, 114 (2006), 3,  
 A.Gossale, I. Nova, E. Tronconi, D. Chatterjee, M. Weibel, *J. of Catalysis* (submitted)  
 O. Kröcher, 1<sup>st</sup> Conference MinNO<sub>x</sub>, Feb. 2007 Berlin (Germany).

➔ Data analysis and literature supports the assumption of a similar reaction mechanism on V-based and Fe-Zeolites SCR catalysts.

# Modeling SCR-Catalytic Converters

Chemical Reactions:  $\text{NO} + \text{NH}_3^*$

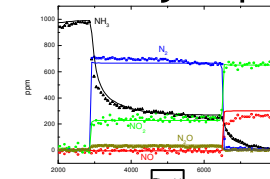
- **NH<sub>3</sub> adsorption:**  $\text{NH}_3 \rightarrow \text{NH}_3^*$
- **NH<sub>3</sub> desorption:**  $\text{NH}_3^* \rightarrow \text{NH}_3$
- **NH<sub>3</sub> oxidation:**  $4\text{NH}_3^* + 3\text{O}_2 \rightarrow 2\text{N}_2 + 6\text{H}_2\text{O}$
- **NO-SCR reaction:**  $4\text{NH}_3^* + 4\text{NO} + \text{O}_2 \rightarrow 4\text{N}_2 + 6\text{H}_2\text{O}$

Chemical Reactions:  $\text{NO} + \text{NO}_2 + \text{NH}_3^{**}$

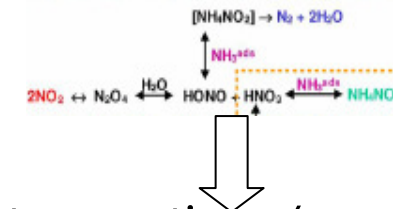
- **Fast-SCR reaction:**  $2\text{NH}_3^* + \text{NO}_2 + \text{NO} \rightarrow 2\text{N}_2 + 3\text{H}_2\text{O}$
- **NO<sub>2</sub>-SCR reaction:**  $4\text{NH}_3^* + 3\text{NO}_2 \rightarrow 3.5\text{N}_2 + 6\text{H}_2\text{O}$
- **N<sub>2</sub>O formation:**  $2\text{NH}_3^* + 2\text{NO}_2 \rightarrow \text{N}_2\text{O} + 3\text{H}_2\text{O} + \text{N}_2$

\* SAE 2005-01-0965, \*\* SAE 2006-01-0468+SAE 2007-01-1136+SAE 2008-01-0867

Microreactor experiments with catalyst powder



Reaction mechanism

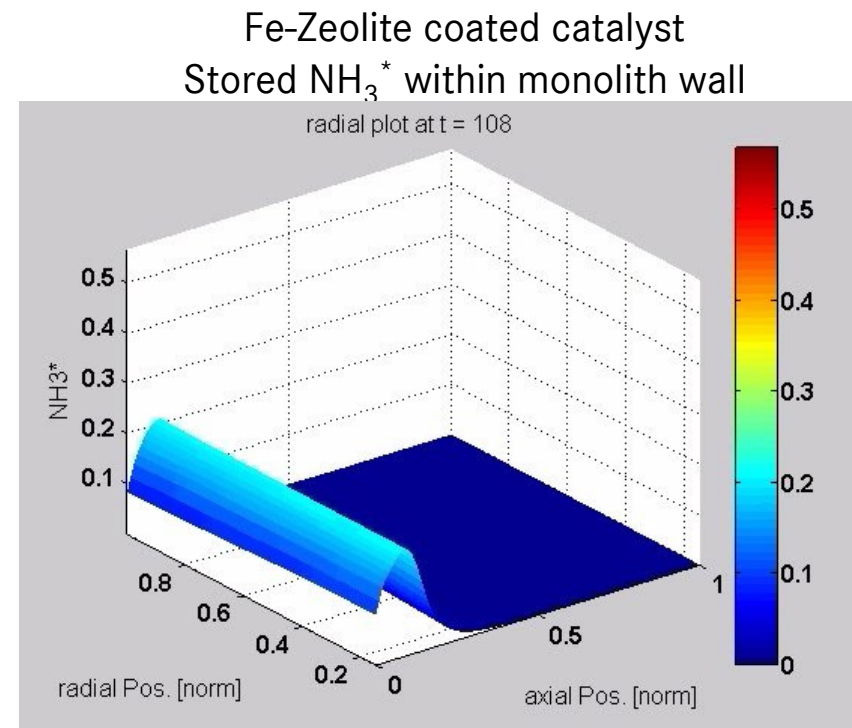
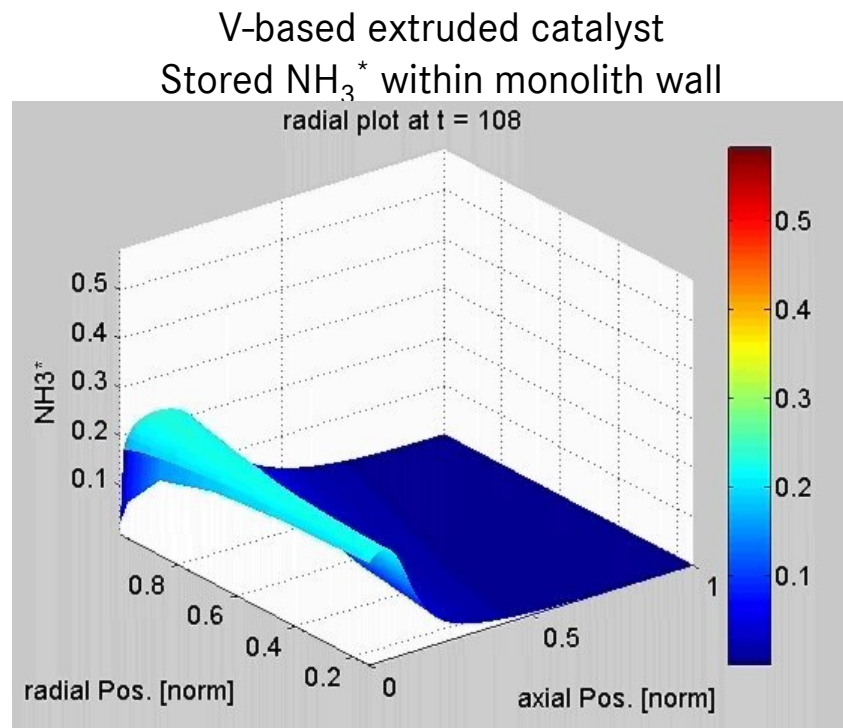


Rate equations/parameter

$$r_{\text{NO}} = k_{\text{NO}}^{\circ} \exp\left(-\frac{E_{\text{NO}}}{RT}\right) \frac{C_{\text{NO}}^{\vartheta}}{1 + K_{\text{LH}} \frac{\vartheta}{1 - \vartheta}} \left(\frac{P_{\text{O}_2}}{0.02}\right)^{\beta}$$

- No NH<sub>3</sub> adsorption/desorption equilibrium is assumed.
- Two-sites L.-H. expression accounts for NH<sub>3</sub> inhibition of the SCR reaction.
- Higher O<sub>2</sub> concentration increases SCR reaction and NH<sub>3</sub> oxidation rates.
- Simplified reaction scheme for the fast SCR reaction ( NH<sub>4</sub>NO<sub>3</sub> formation neglected)

# Modeling SCR-Catalytic Converters: Zeolite vs. V-Based Catalyst: $\text{NH}_3$ Gradients within Catalytic Walls/Washcoat ETC-Simulation



⇒ Washcoat diffusion seems to be of minor importance on Zeolite.

$\alpha=1$ , Catalyst: 18L, 400cpsi

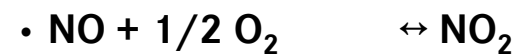
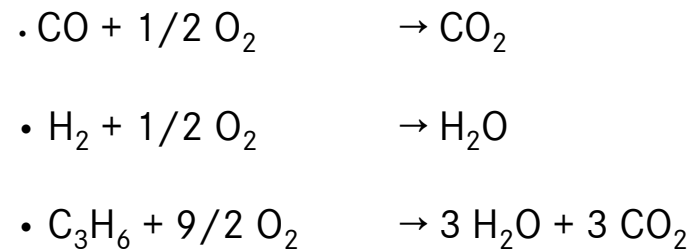


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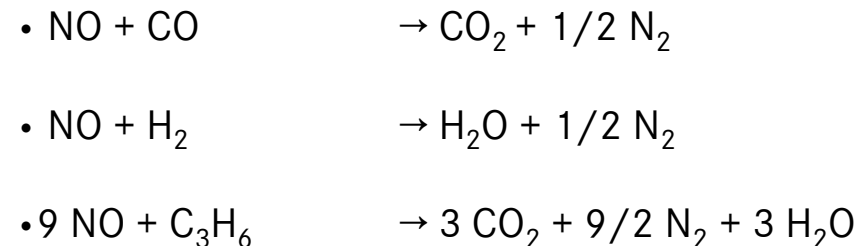
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# DOC Modeling: Reaction Kinetics

## Chemical Reactions: Oxidation Reactions



## Chemical Reactions: NO-Reduction



## Rate Equations\*

$$R_1 = k_1 \frac{y_{\text{CO}} y_{\text{O}_2}}{G_1}$$

$$R_2 = k_2 \frac{y_{\text{H}_2} y_{\text{O}_2}}{G_1}$$

$$R_3 = k_3 \frac{y_{\text{C}_3\text{H}_6} y_{\text{O}_2}}{G_1}$$

$$R_4 = k_9 \left( y_{\text{NO}} y_{\text{O}_2}^{0.5} - \frac{y_{\text{NO}_2}}{K_{v,9}^{\text{eq}}} \right) \frac{1}{G_2}$$

$$R_5 = k_4 \left( y_{\text{CO}} y_{\text{H}_2\text{O}} - \frac{y_{\text{CO}_2} y_{\text{H}_2}}{K_{y,4}^{\text{eq}}} \right)$$

$$G_x = L.H. \text{ Inhibition Term}$$

$$R_6 = k_6 y_{\text{CO}} y_{\text{NO}}^{0.5}$$

$$R_7 = k_7 y_{\text{H}_2} y_{\text{NO}}^{0.5}$$

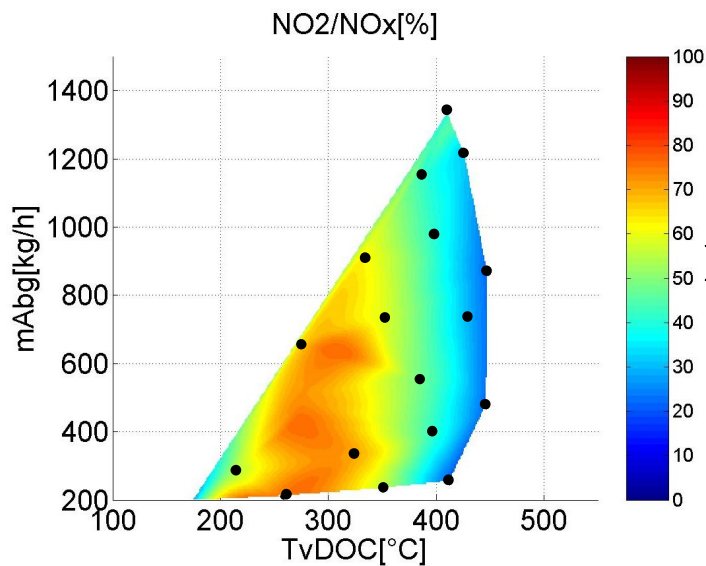
$$R_8 = k_8 y_{\text{C}_3\text{H}_6} y_{\text{NO}}^{0.5}$$



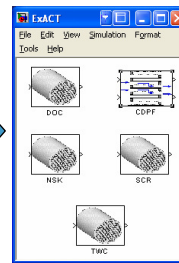
- Global reaction kinetics for CO, H<sub>2</sub>, C<sub>3</sub>H<sub>6</sub> oxidation and NO reduction.
- Backward reaction/thermal equilibrium included for NO-Oxidation
- Inhibition effects are considered.

# DOC Modeling: Model Calibration Strategy

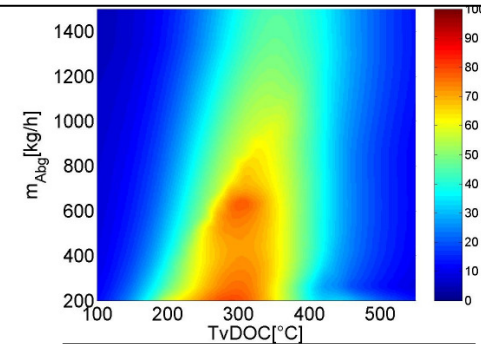
Measurement of Selected Operating Points  
(Engine Test Bench)



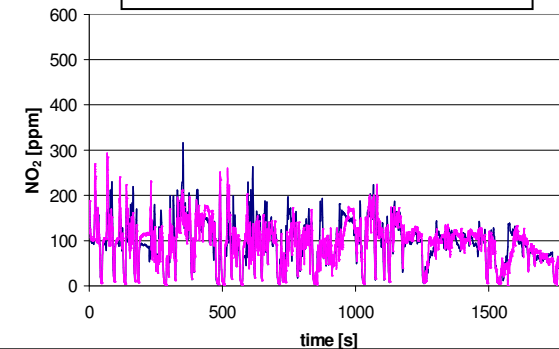
Calibration



Extrapolation to Complete Engine Map

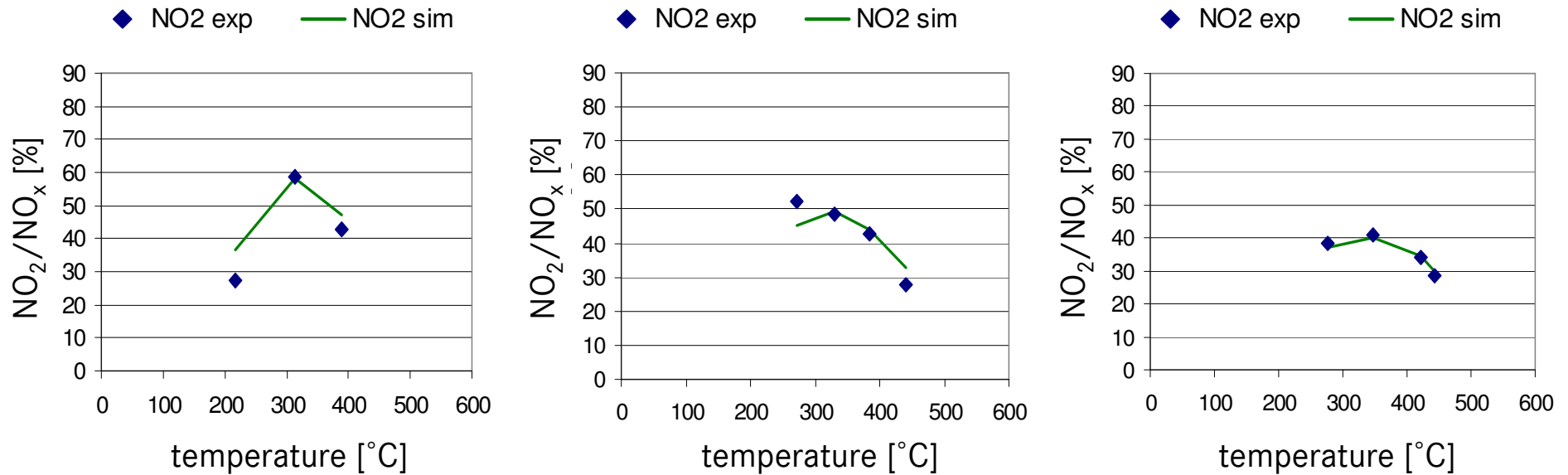


Transient Simulations



- Development of standard calibration procedures.  
(steady state test bench data sufficient for NO-oxidation)
- Model based map generation for ECUs + Assessment of transient performance.

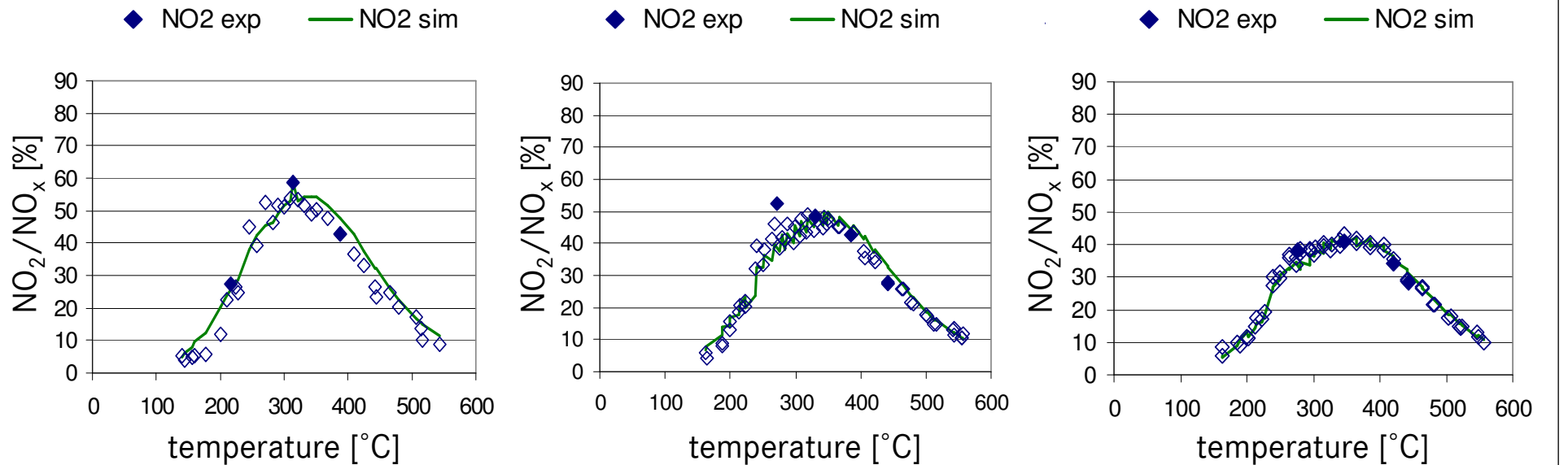
# DOC Modeling: NO-Oxidation – Calibration of Reaction Kinetics



NO<sub>2</sub>-Ratio at approx. 400kg/h m<sub>Exh</sub>    NO<sub>2</sub>-Ratio at approx. 500kg/h m<sub>Exh</sub>    NO<sub>2</sub>-Ratio at approx. 700kg/h m<sub>Exh</sub>  
 Measurements taken from test bench experiments (catalyst 1)

- Objective: Development of a simple calibration procedure
- ➔ • Model calibration to new DOC technology:
- Keep E<sub>act</sub> fixed, only preexponential factor k<sub>0</sub> is changed

# DOC Modeling: NO-Oxidation – Steady State Validation



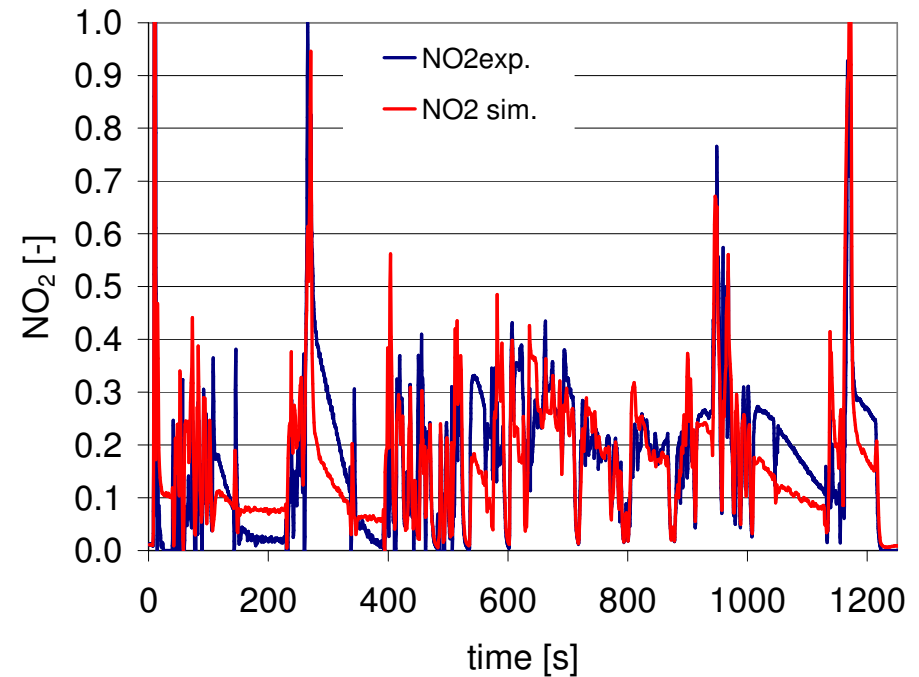
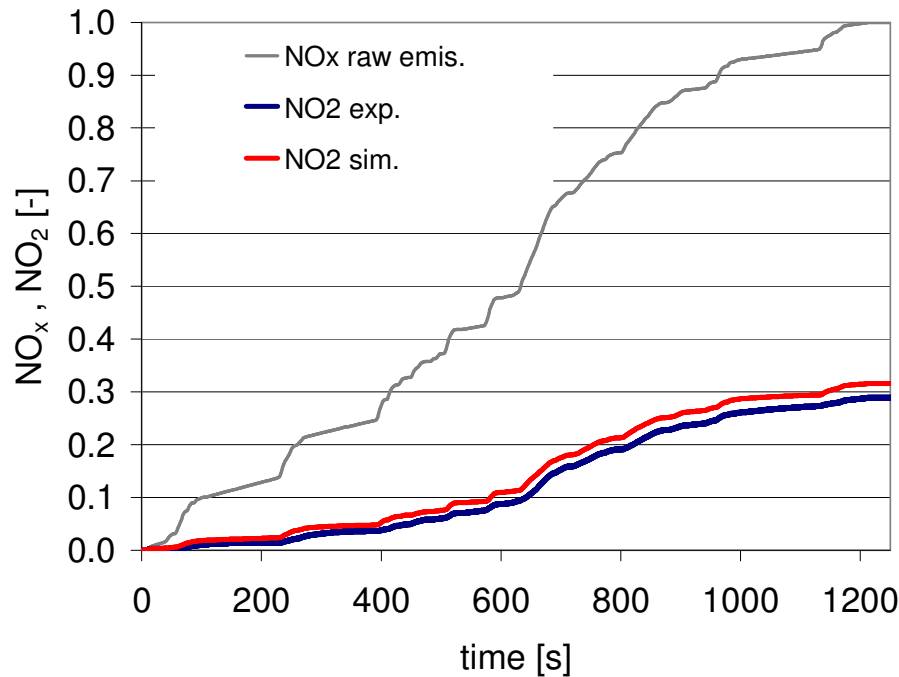
*NO<sub>2</sub>-Ratio at approx. 400kg/h m<sub>Exh</sub>    NO<sub>2</sub>-Ratio at approx. 500kg/h m<sub>Exh</sub>    NO<sub>2</sub>-Ratio at approx. 700kg/h m<sub>Exh</sub>*

Measurements taken from test bench experiments (catalyst 1)

- Only a few measurements at different mass flows are required for calibration.
- Excellent prediction quality for steady state conditions.

# DOC Modeling: NO-Oxidation - Transient Validation

FTP-Testcycle

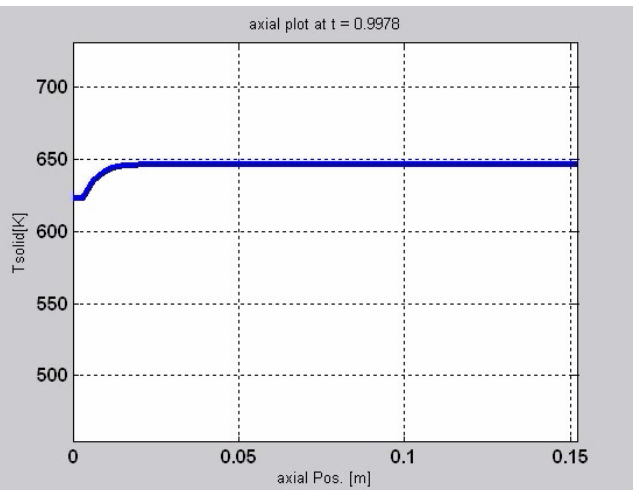


- Excellent prediction quality (error<2.3%) also for transient conditions.
- Deviations at idle speed conditions.

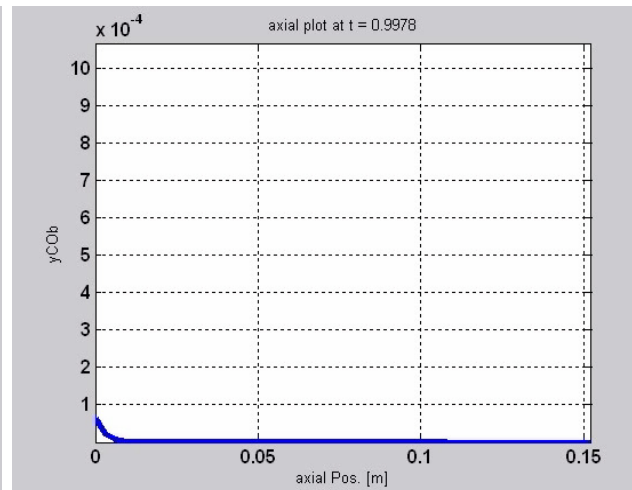
# DOC Modeling: Axial Profiles within DOC Catalyst

ETC Simulation

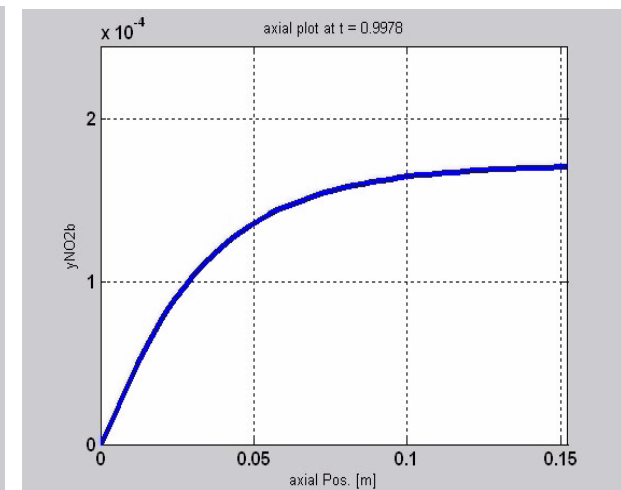
$T_{\text{solid}}$  [K]



CO molefrac. [-]



$\Delta\text{NO}_2$  molefrac. [-]



- Significant axial temperature and concentration gradients within monolith.
- CO/HC-oxidation located at catalyst entrance. NO<sub>2</sub> formation only after CO and HC depletion.
- Slow NO oxidation rate compared to CO and HC oxidation.

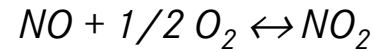
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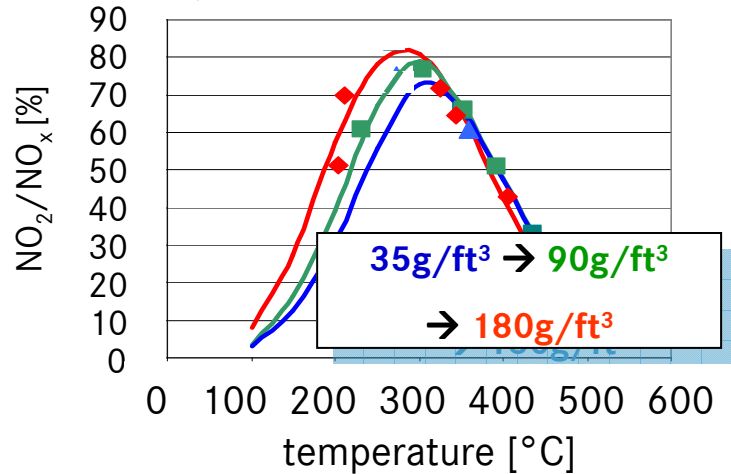


# DOC Modeling: Correlation PGM-Loading – Kinetic Parameters

e.g.: NO-oxidation:



Steady State Measurements:



Parameterization:  $k^0$

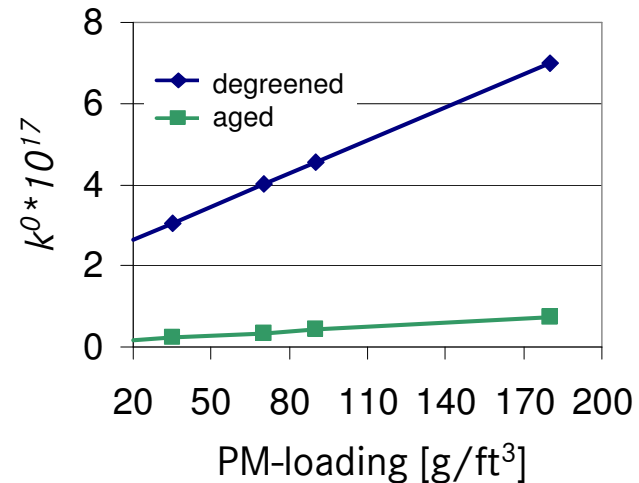
$$R_4 = k_o^0 e^{\frac{E_a}{RT}} \left( y_{NO} y_{O_2}^{0.5} - \frac{y_{NO_2}}{K_{y,9}^{eq}} \right) \frac{1}{G_2}$$

Correlation:

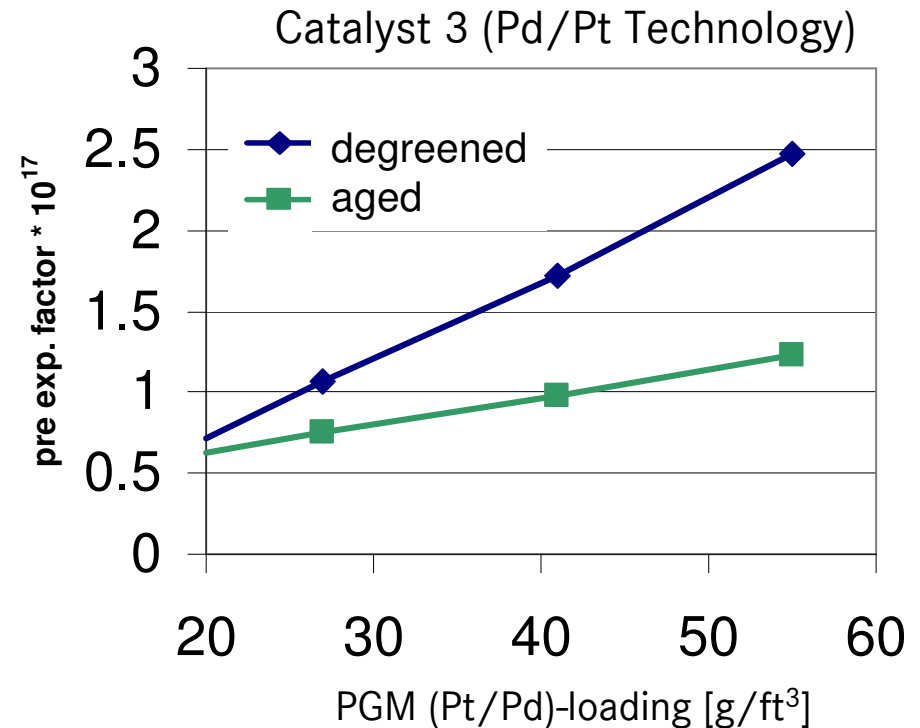
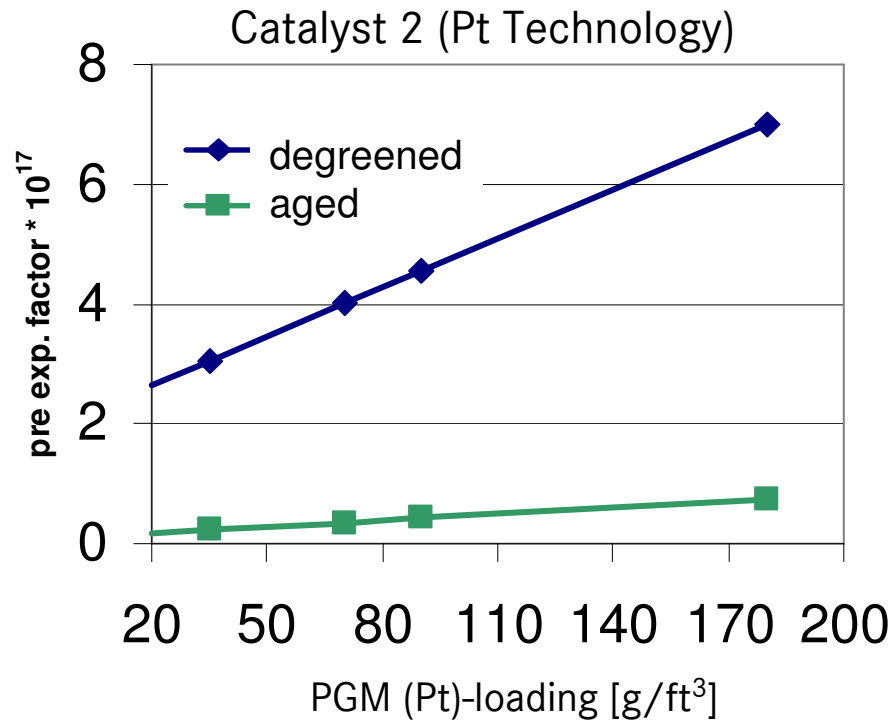
Kinetic-parameter

$$\frac{k^0}{k_{max}^0} = a \rho_{EM}^b + c$$

PM-loading



# DOC Modeling: Kinetic Correlation for NO-Oxidation



Degreened: 600°C, 5h real exhaust, Aged: 700 °C, 200h hydrothermal oven aged

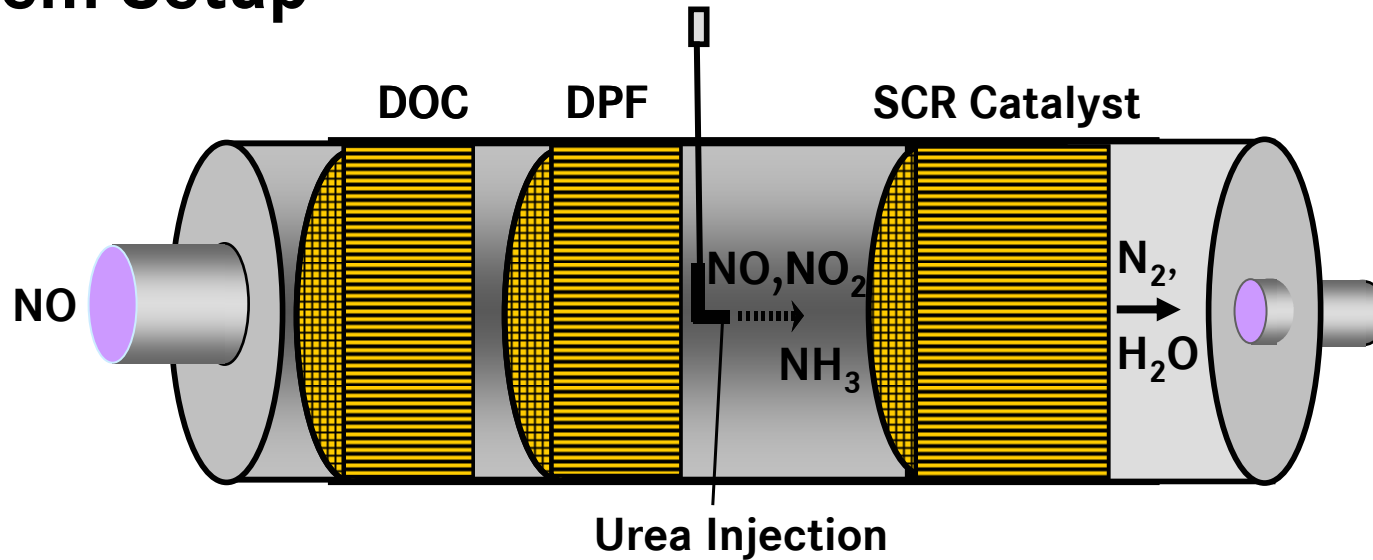


- A linear correlation is revealed within the studied range of PGM loadings.
- Linear correlation is valid also for degreened and aged state.
- Stronger deterioration by aging at higher PGM loading.
- Pd/Pt technology reveals lower intrinsic activity, but more aging resistance.

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# Simulation Study: DOC+DPF+SCR: System Setup



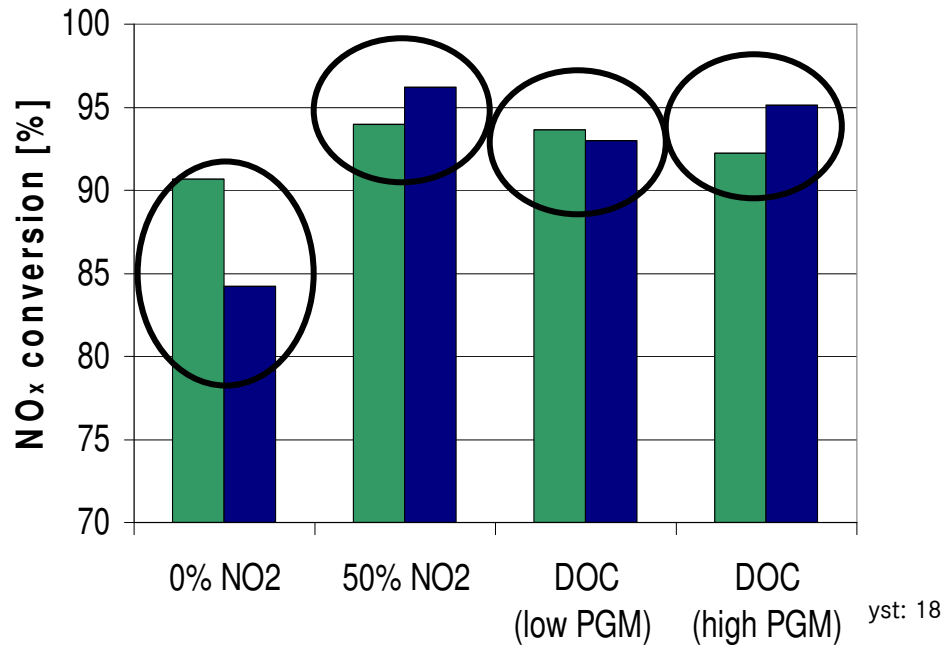
	DOC	DPF	SCR
Noble Metal	Pt/Pd	uncoated	-
Loading [g/ft <sup>3</sup> ]	varied	uncoated	Fe-Zeolite
Volume*	0.5	0.7	1

Design Target 80% NO<sub>x</sub> conversion within the combined FTP  
(1/7 cold FTP + 6/7 hot FTP)

# Model Application: Total NO<sub>x</sub> Conversion Efficiencies: Fe-Zeolite vs. V-Based Catalyst\*

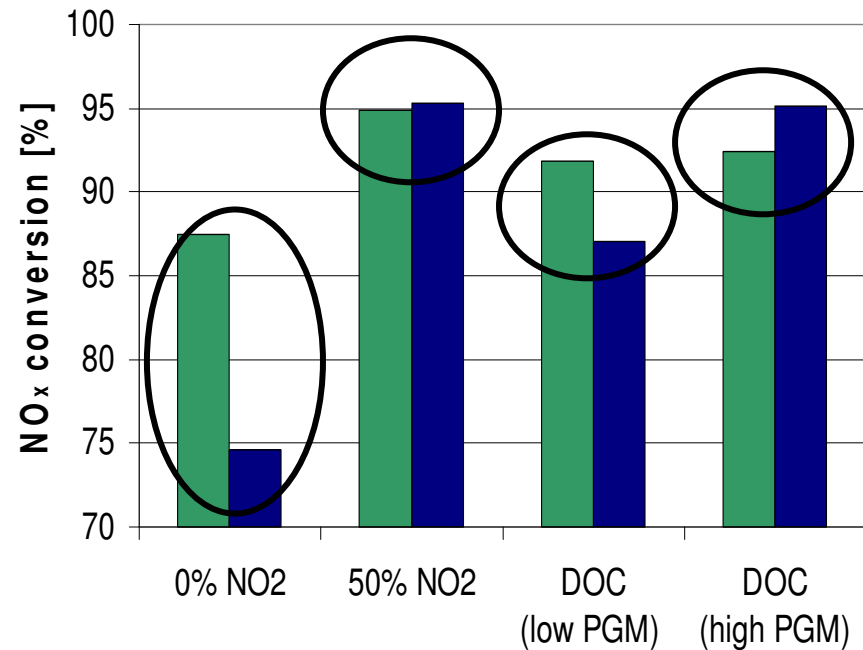
ESC

V-based Zeolite



ETC

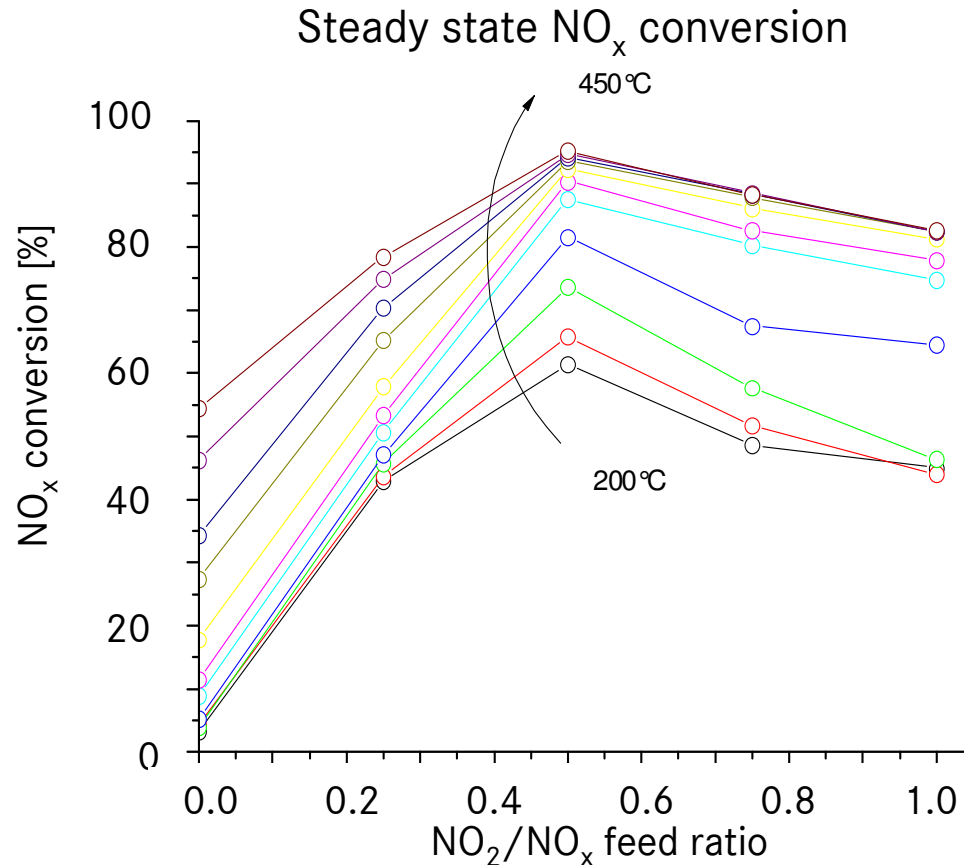
V-based Zeolite



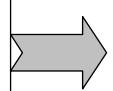
- V-based catalyst has higher efficiency for low NO<sub>2</sub> input concentrations.
- Fe-Zeolite catalyst has higher efficiency for optimized NO<sub>2</sub>/NO<sub>x</sub> ratios.

\* SAE 2007-01-1136

# SCR NO<sub>x</sub> Conversion Efficiency: Influence of NO<sub>2</sub>



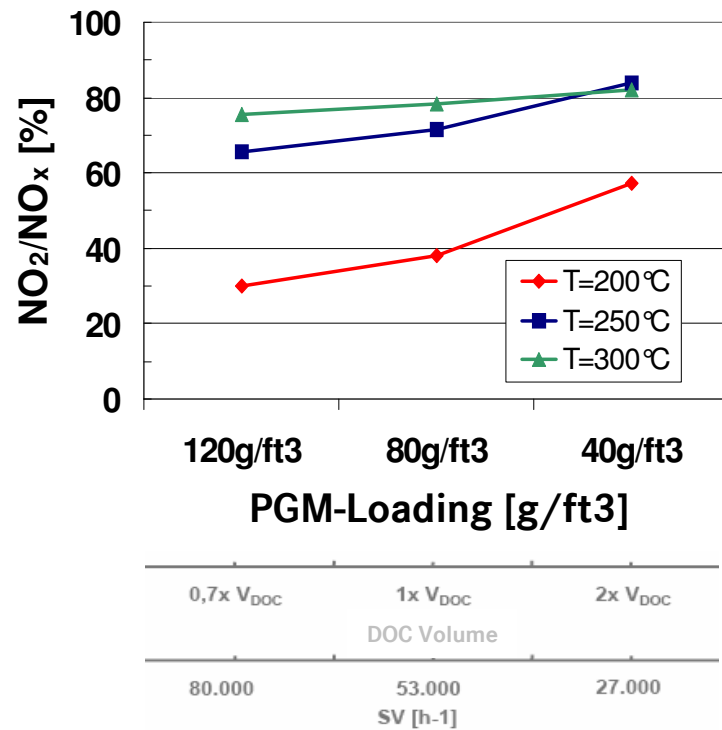
Microreactor sim./exp. with Fe-Zeolite SCR:  
Feed S.V. 172.000 1/h, 1000ppm NO<sub>x</sub> and 1000ppm NH<sub>3</sub>, 2 Vol-% O<sub>2</sub>, 10 Vol-% H<sub>2</sub>O



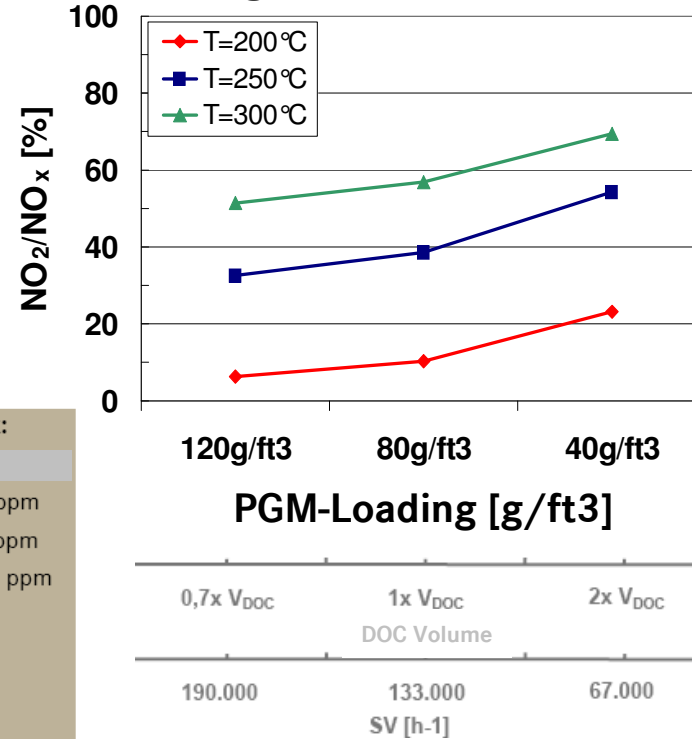
- SCR conversion efficiency and product selectivity is influenced by the NO<sub>2</sub>/NO<sub>x</sub> feed ratio.
- Fe-Zeolites are more sensitive to NO<sub>2</sub>/NO<sub>x</sub> ratios lower than 50%,

# DOC Modeling: Steady State NO-Oxidation PGM-Loading vs. Catalyst Volume – Steady State

Constant Total PGM-Loading



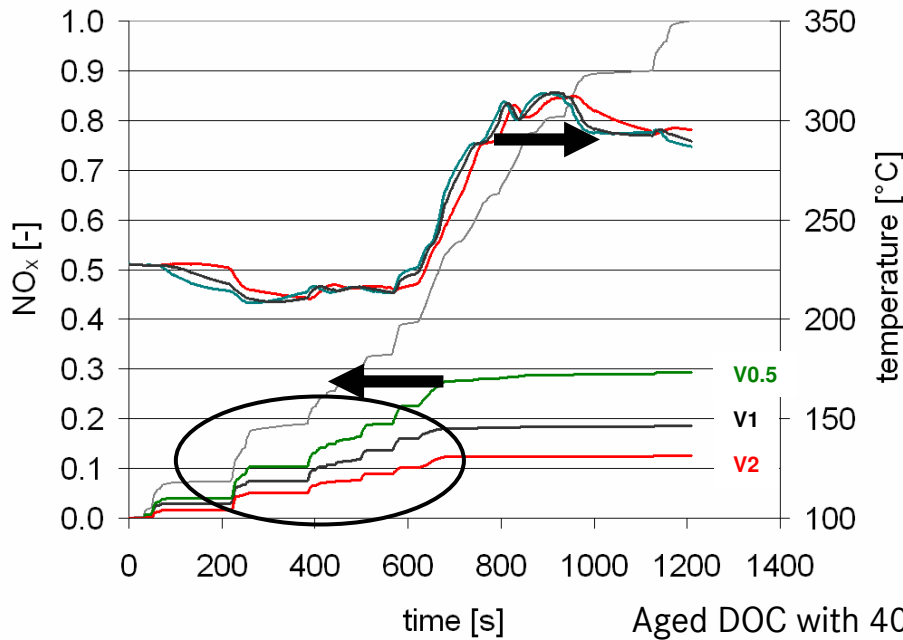
**Gasmatrix:**  
 NO = 200 ppm  
 CO = 100 ppm  
 C3H6 = 20 ppm  
 O2 = 8 %  
 CO2 = 8 %  
 H2O = 8 %  
 N2 = Rest



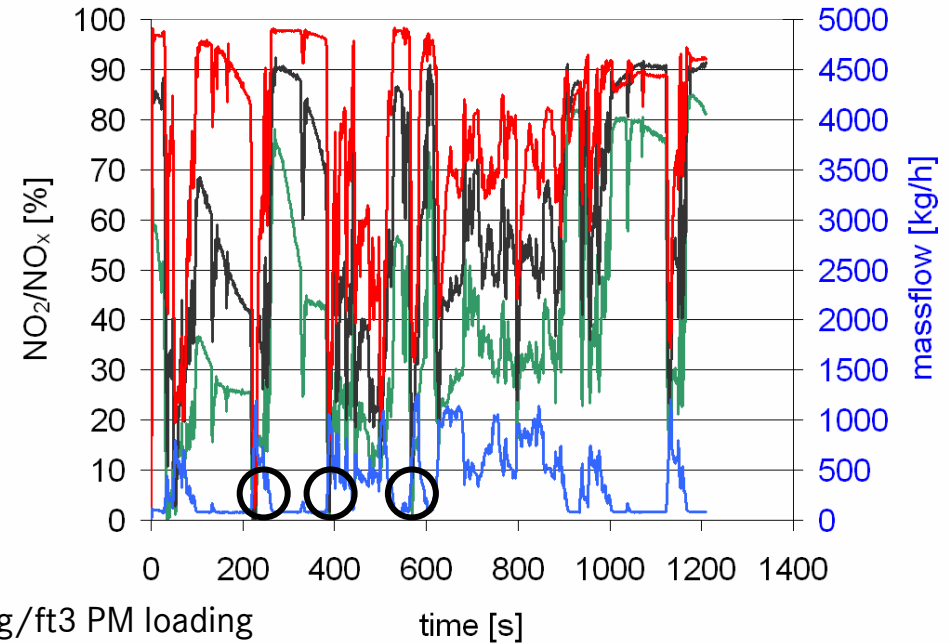
- Simulation reveals higher performance for low specific PGM-loading.
- Conversion efficiency at high S.V. is limited by residence time.

# DOC+DPF+SCR Simulation: DOC Optimization: Detailed Analysis Hot Start FTP

Cumulative NO<sub>x</sub> (End of Pipe) and SCR Temperature



NO<sub>2</sub>/NO<sub>x</sub> at DOC Outlet and Exhaust Mass Flow

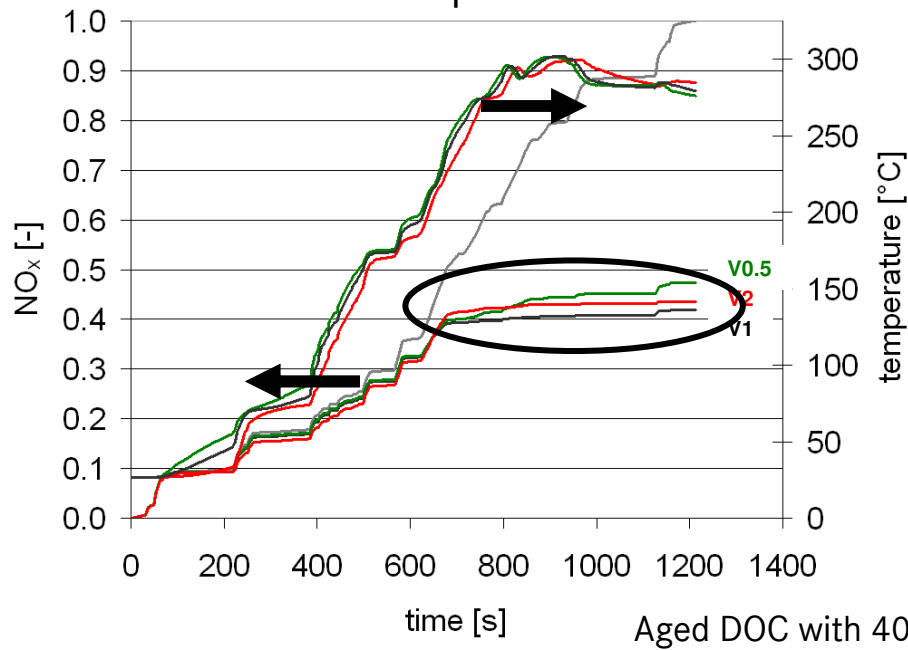


- Total NO<sub>x</sub> conversion determined by first 650s of the FTP.
- Low NO<sub>2</sub>/NO<sub>x</sub> ratios after DOC during acceleration peaks in the first 650s.
- System performance limited by DOC performance under low temperature and high mass flow operating conditions.

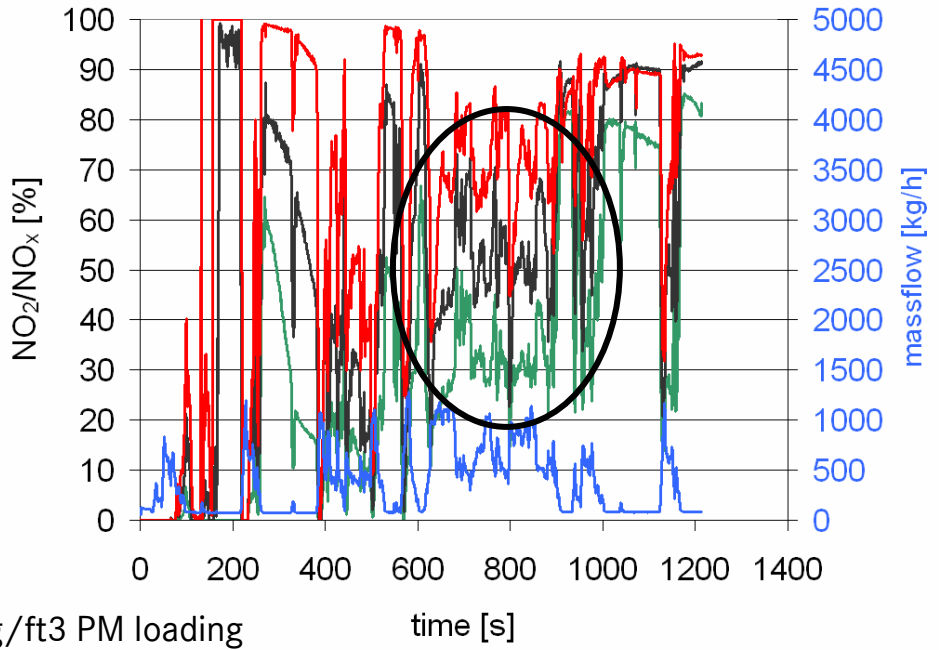


# DOC+DPF+SCR Simulation: DOC Optimization: Detailed Analysis Cold Start FTP

Cumulative  $\text{NO}_x$  (End of Pipe) and SCR Temperature



$\text{NO}_2/\text{NO}_x$  at DOC Outlet and Exhaust Mass Flow

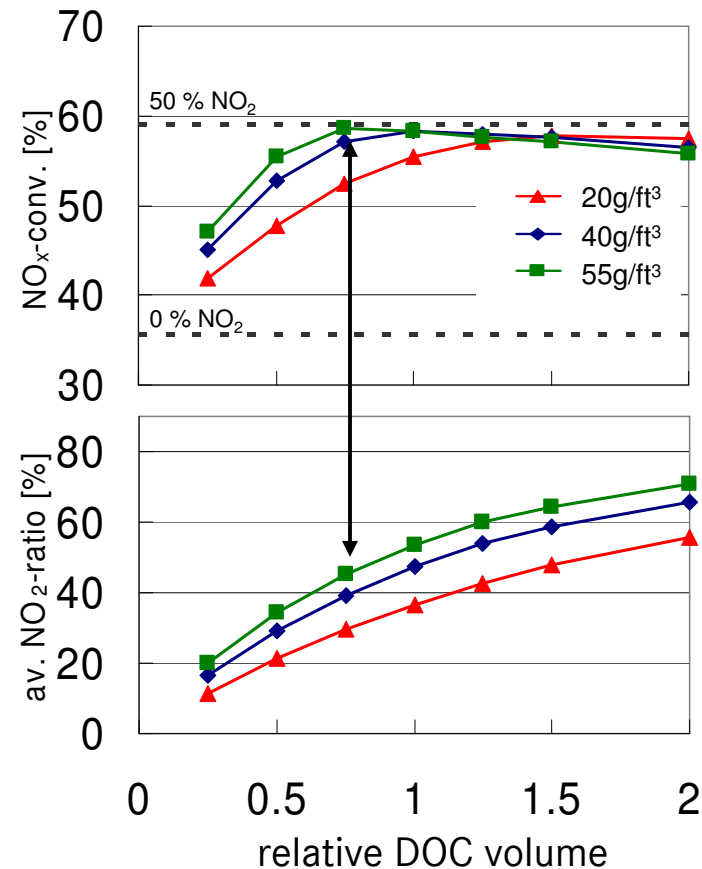
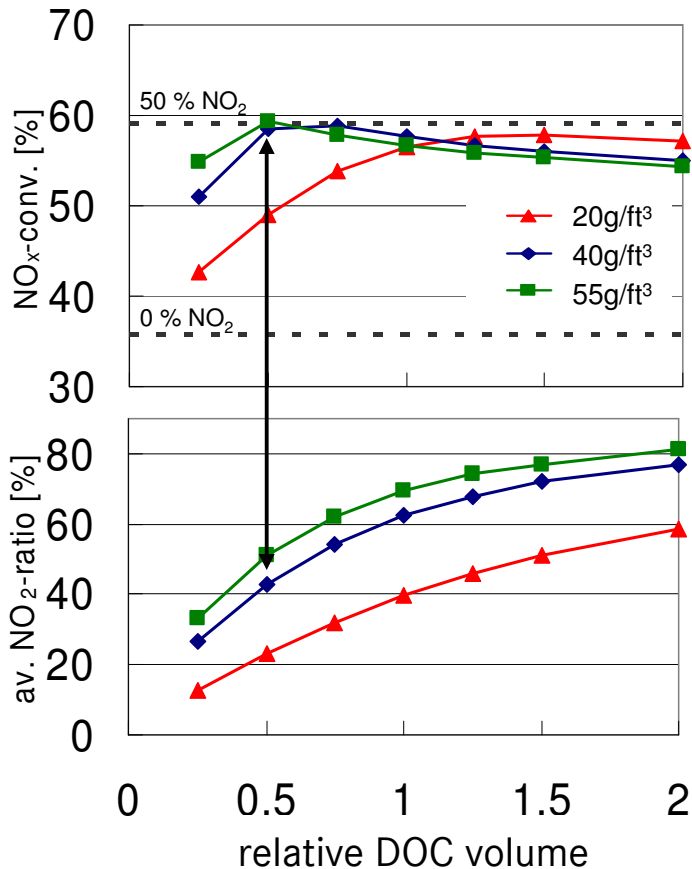


➔ Total  $\text{NO}_x$  conversion determined by the second part of the FTP (600s-1200s)



# DOC+DPF+SCR Simulation: DOC Influence on SCR

Total NO<sub>x</sub> SCR Conversion Efficiencies Within Cold Start FTP  
 DOC degreened DOC aged

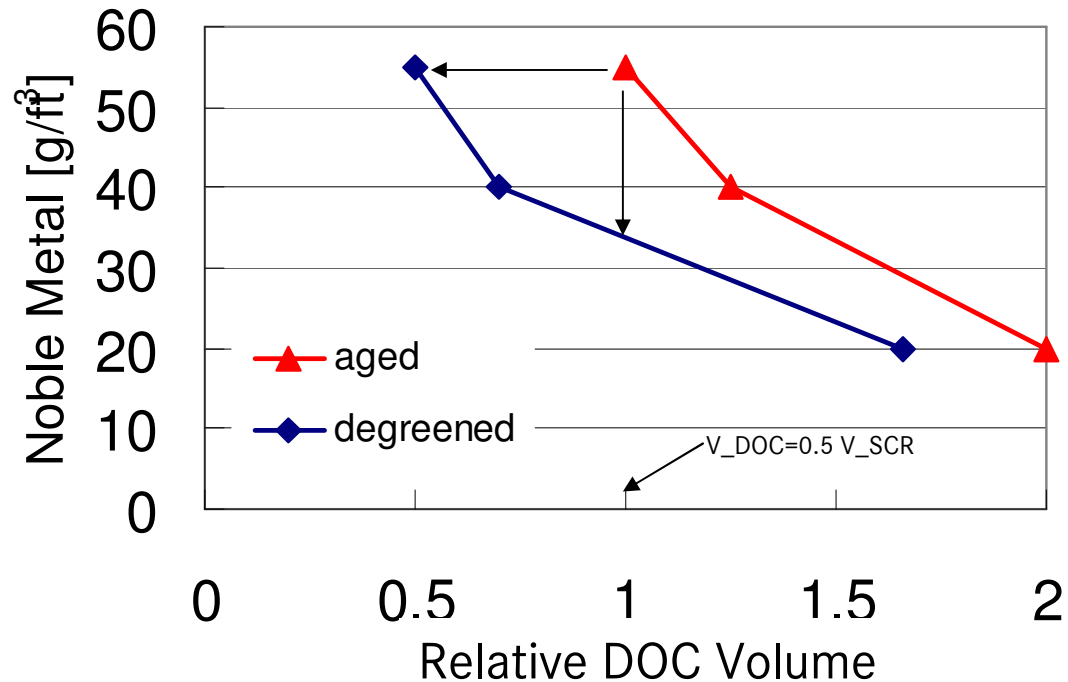


- ➔
- Depending on PGM loading a optimum DOC volume is revealed.
  - Noble metal loading higher than 40 g/ft<sup>3</sup> required for max. efficiency.
  - Significant volume reduction possible of degreened state could be stabilized.

# DOC+DPF+SCR Simulation:

## Impact of DOC Volume vs. Noble Metal Loading on SCR

DOC Parameters for 80% Combined FTP SCR Conversion Efficiency



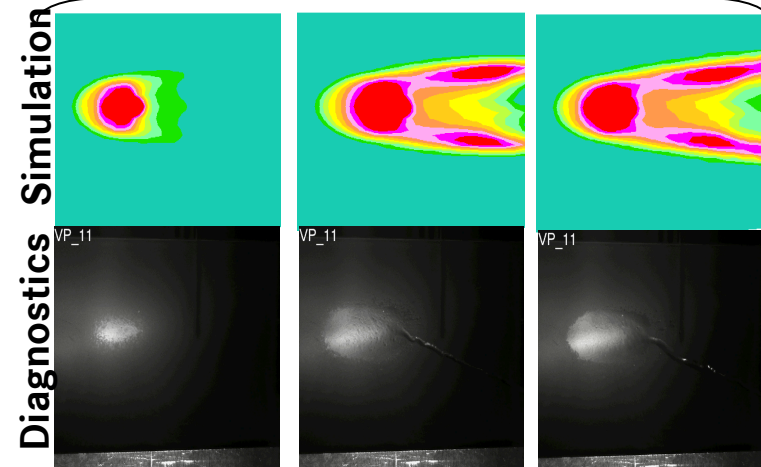
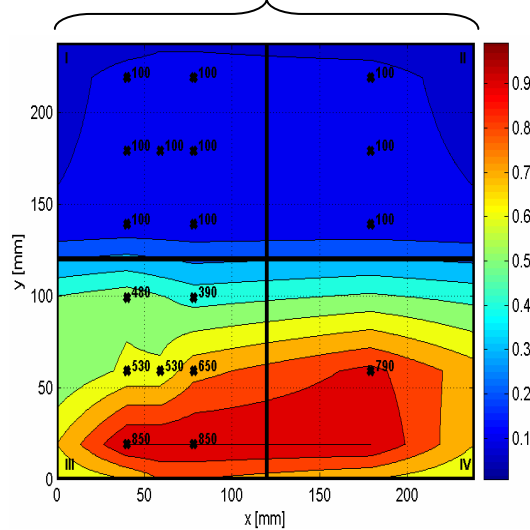
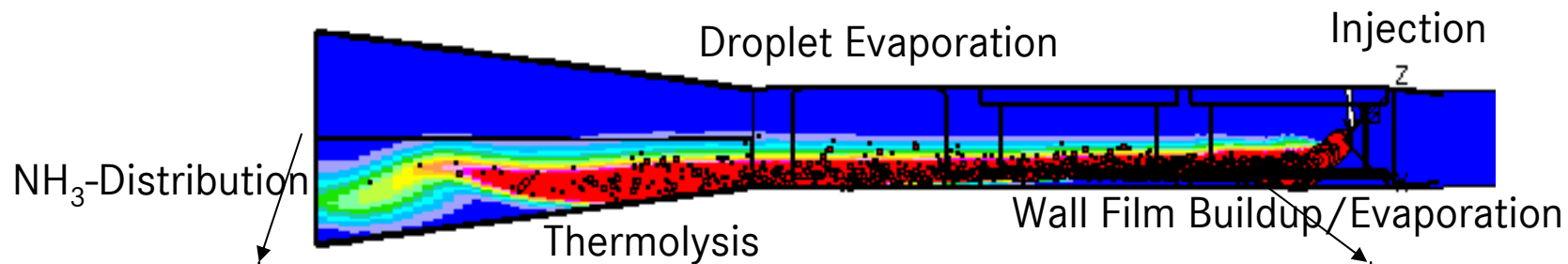
- 55g/ft<sup>3</sup> DOC noble metal loading required with the reference DOC volume at aged state to achieve 80% FTP NO<sub>x</sub> Performance.
- Significant volume and noble metal reduction possible if degreened state could be stabilized.
- Volume Reduction requires significant increase in noble metal loading  
→ Residence time limitation.

## Outline

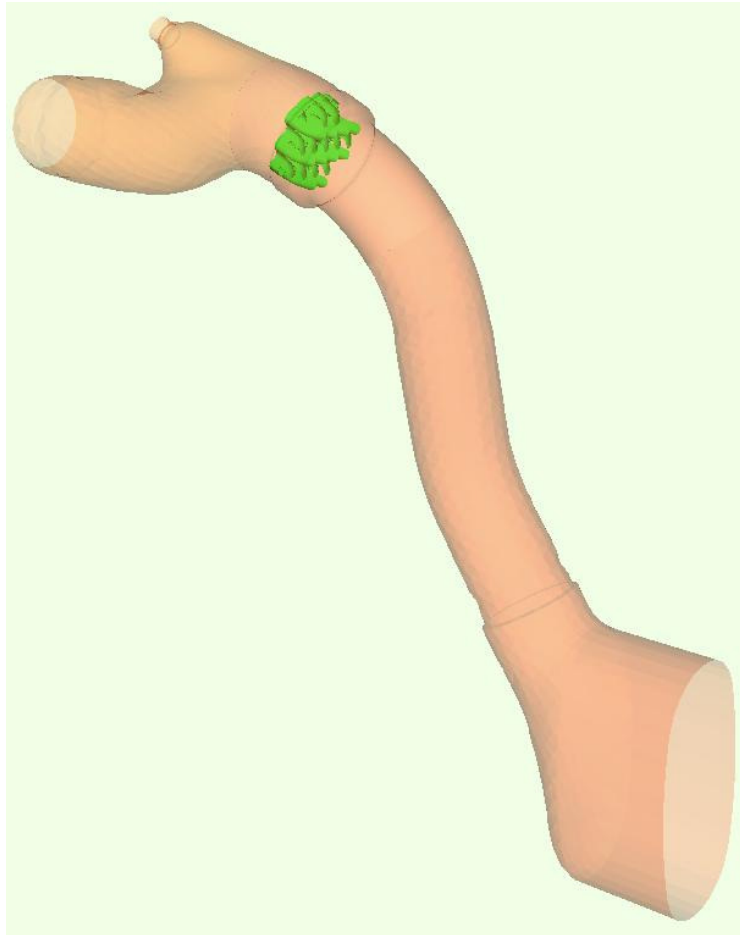
- Advanced SCR-System Setup (DOC+DPF+SCR)
- Simulation Concept
- Modeling SCR Converters
- Modeling DOC Converters
- DOC: Kinetic Correlations
- Simulation Based System Optimization: DOC Impact on SCR
- Modeling of Urea Processing
- Conclusions

# 3D Simulation AdBlue Processing:

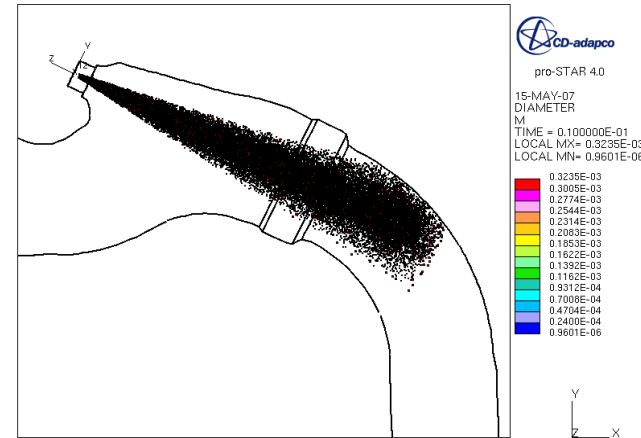
AdBlue Processing Within the Exhaust Line



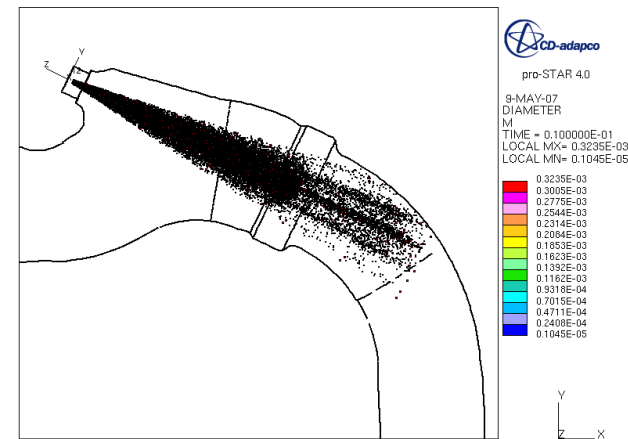
# 3D Simulation AdBlue Processing: Spray Modeling and Influence of Mixing Device



No Mixer



With Mixer

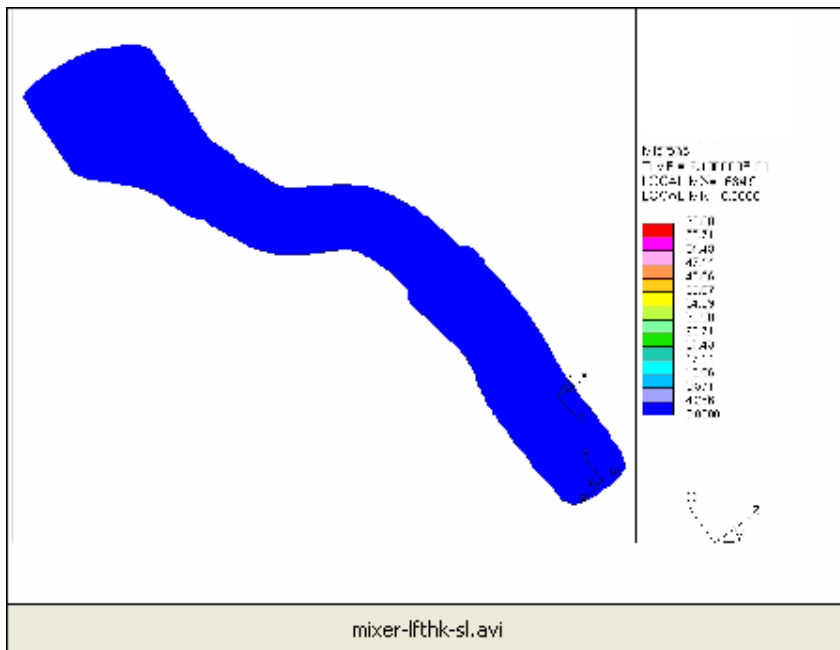


10s after start of injection

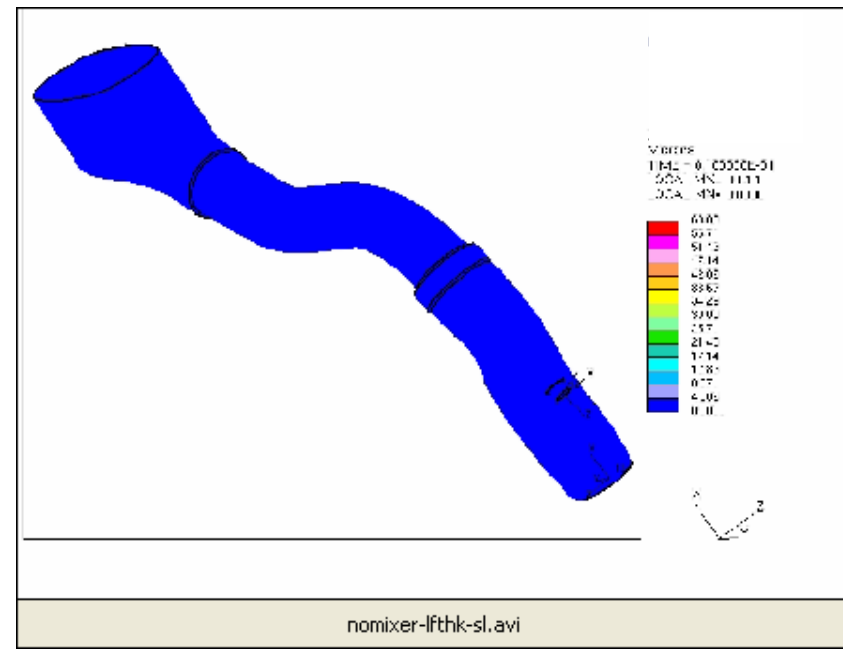
Reduction of wall contact, because of spray interception within mixing device.

# 3D Simulation AdBlue Processing: Mixer Influence on Wall Film Buildup\*

With Mixer



No Mixer



$m_{Exh} = 100\text{kg/h}$ ,  $T_{Exh} = 200^\circ\text{C}$

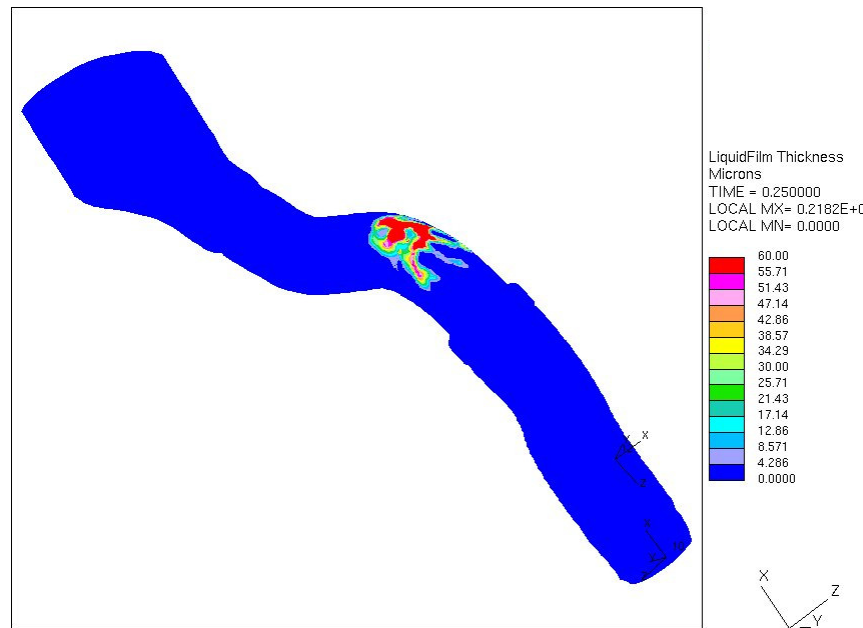
Spray interception by the mixer yields to reduced wall film formation.

\* high AdBlue overdosing

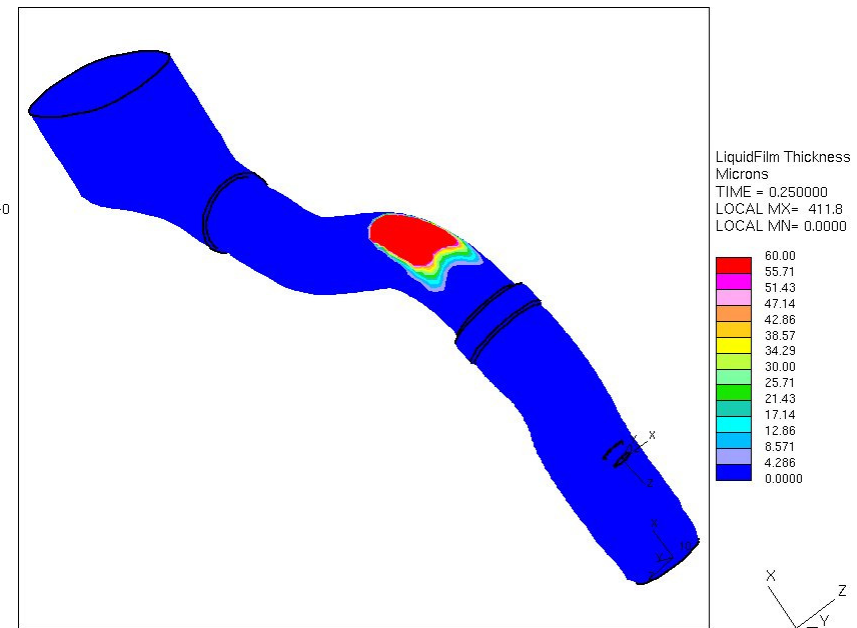


# 3D Simulation AdBlue Processing: Mixer Influence on Wall Film Buildup\*

With Mixer



No Mixer

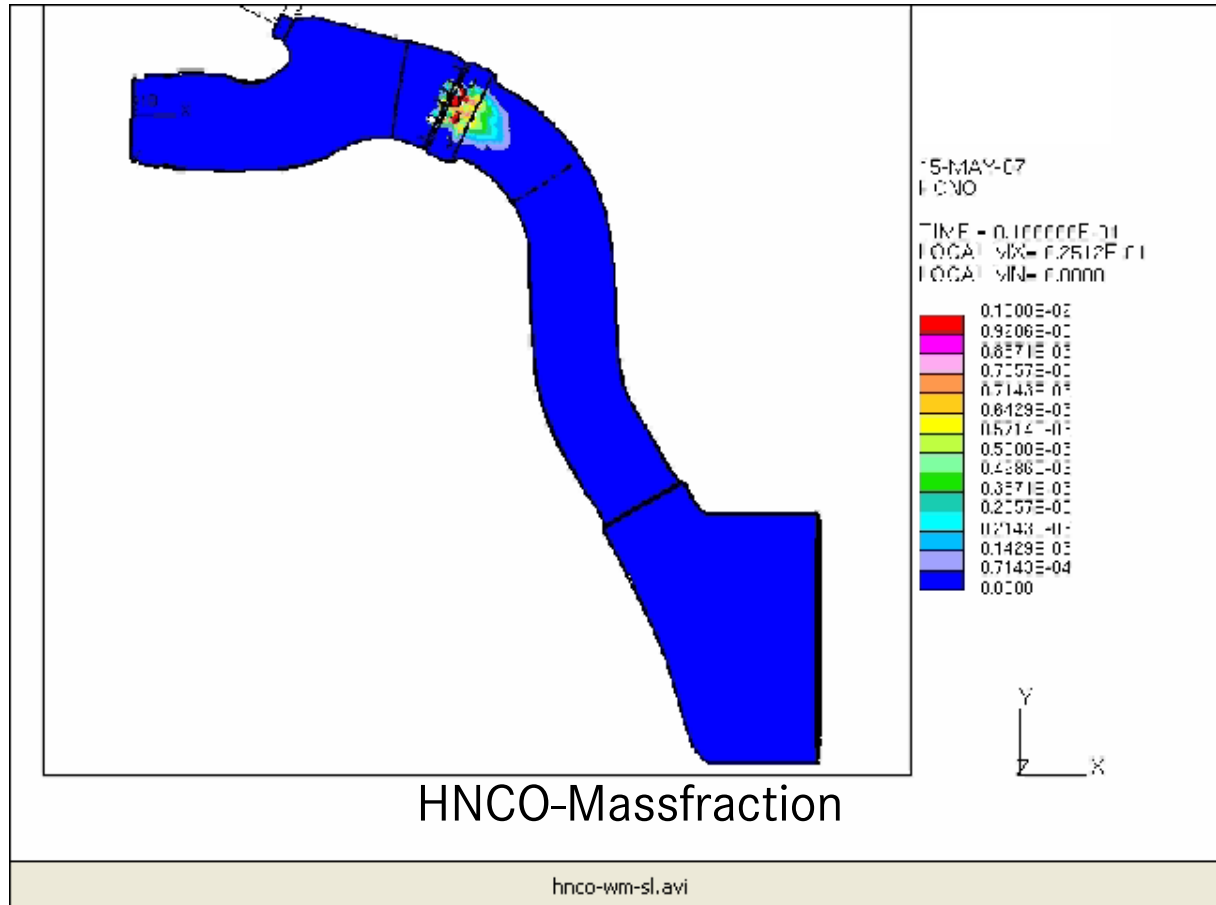


$m_{Exh} = 100\text{kg/h}$ ,  $T_{Exh} = 200^\circ\text{C}$

Spray interception by the mixer yields to reduced wall film formation.

\* high AdBlue overdosing

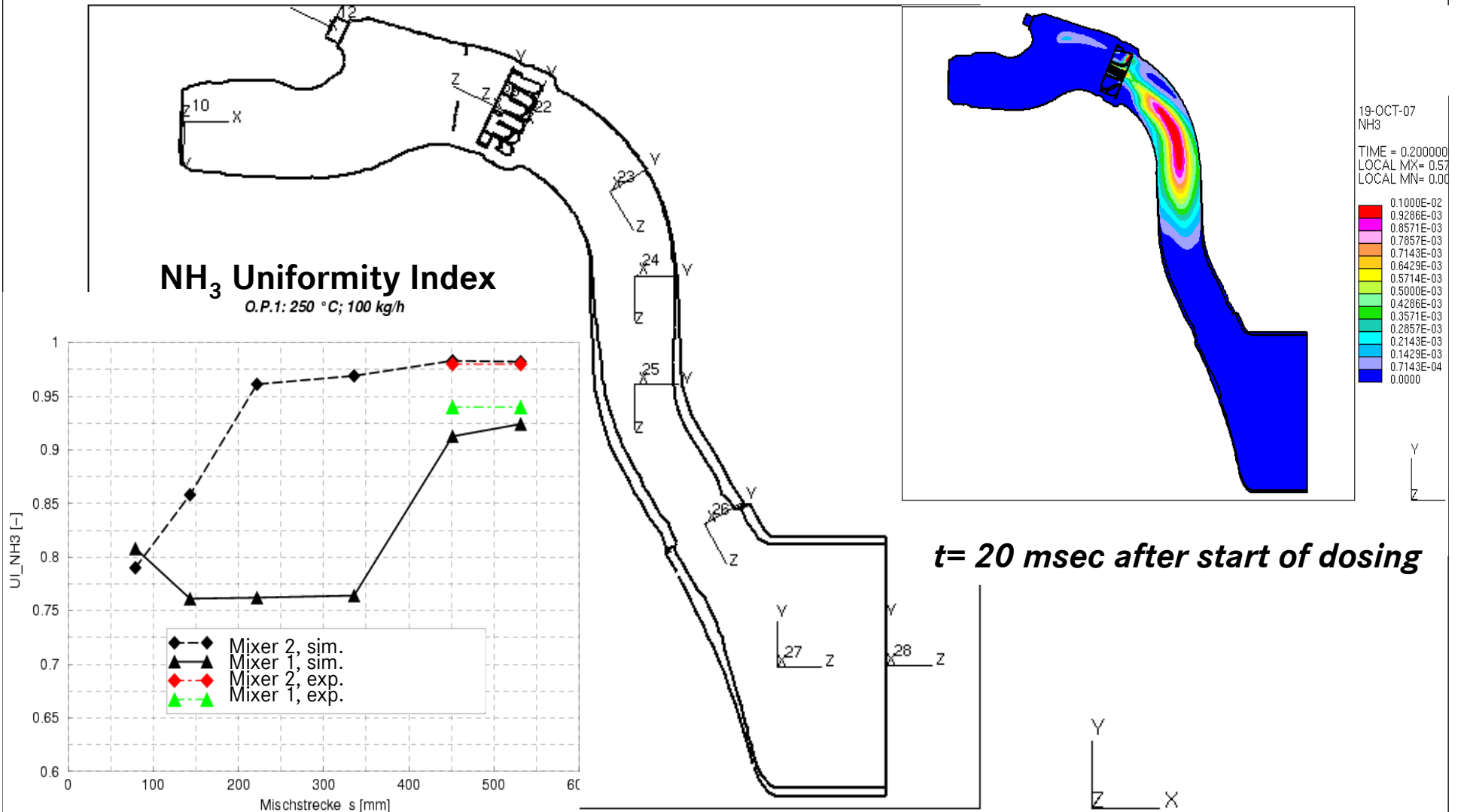
# 3D Simulation AdBlue Processing: Thermolysis/HNCO-Formation



Mixer acts as an efficient evaporation surface within the exhaust flow.

Dr. Schöffel- GR/VPE

# 3D Simulation AdBlue Processing: NH<sub>3</sub> - Distribution



## Outline

- Advanced SCR-System Setup (DOC+DPF+SCR)
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- DOC: Kinetic Correlations
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- **Conclusions**

## Conclusions

- Integrated EGA system modeling has become essential and very efficient tool to assist the design of complex exhaust gas aftertreatment systems.
- Approach: 1D System modeling refined by 3D component modeling.
- Kinetic correlations offers the possibility to include noble metal variations in the system analysis.
- DeNOx efficiency of DOC+DPF+SCR (Fe-Zeolite) systems is limited by the DOC performance at low temperatures and high mass flows.
- New SCR materials may offer the possibility to reduce the critical impact of the DOC performance.
- DOC aging leads to significant “over sizing” of the system.
- Model extensions regarding aging are necessary.

## Acknowledgments



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# DAIMLER

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