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# EGR Catalyst for Cooler Fouling Reduction

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*CLEERS Workshop  
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# Acknowledgements

- Jimi Tjong and his dynamometer group at the Ford Canada Essex Engine Plant ran the cooler fouling engine tests
- Oak Ridge National Lab performed some of the analyses shown – Scott Sluder, John Storey, Sam Lewis
- Johnson Matthey Environmental Catalysts and Technologies provided the catalysts used in these experiments



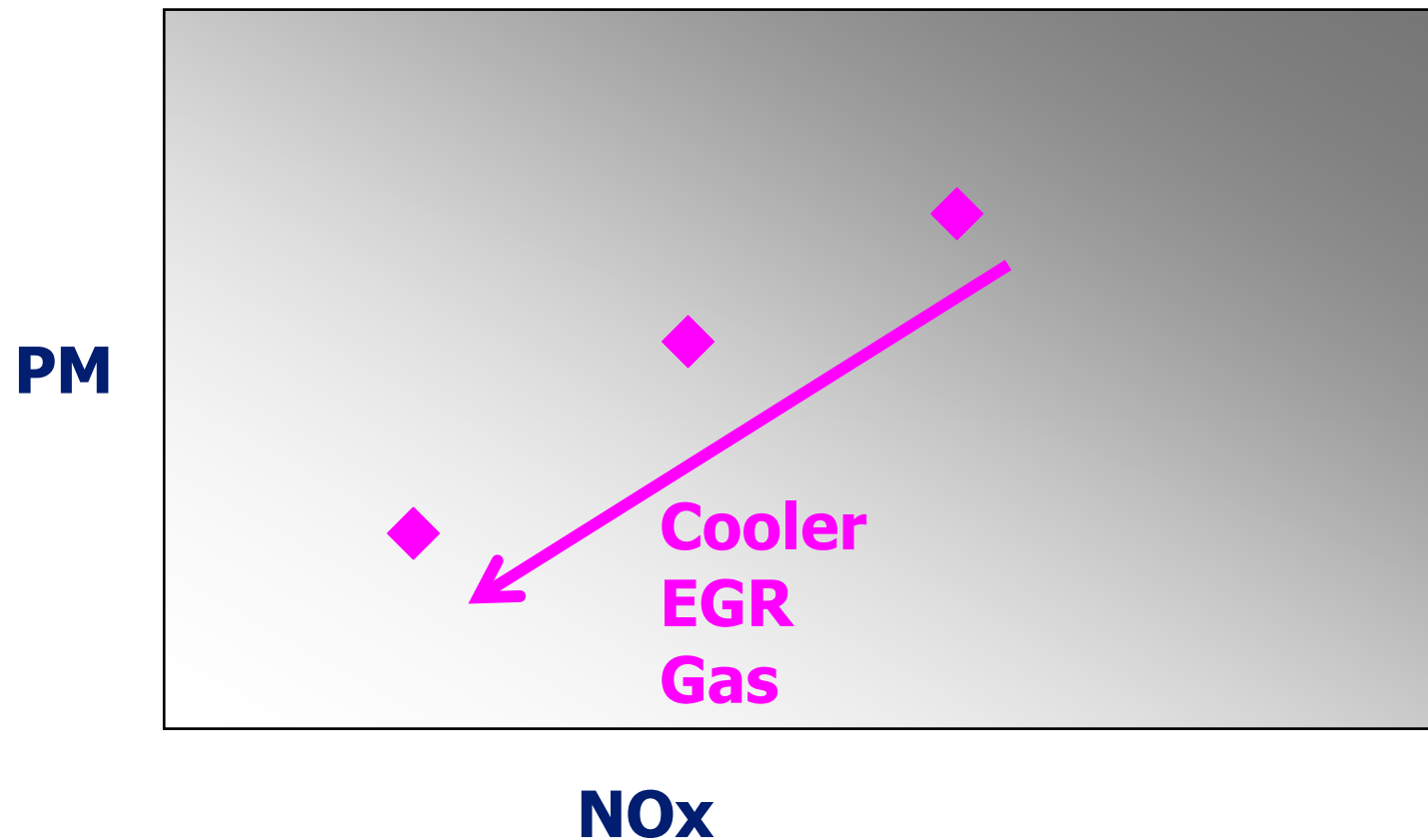
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## Increased EGR Cooling required

- Future emission standards: lower NO<sub>x</sub>, PM
- Reduced charge temperature helps (see next slide)
  - EGR is cooled
  - Bigger coolers
  - Lower temperatures



# Improved EGR Cooling Reduces NOx and PM



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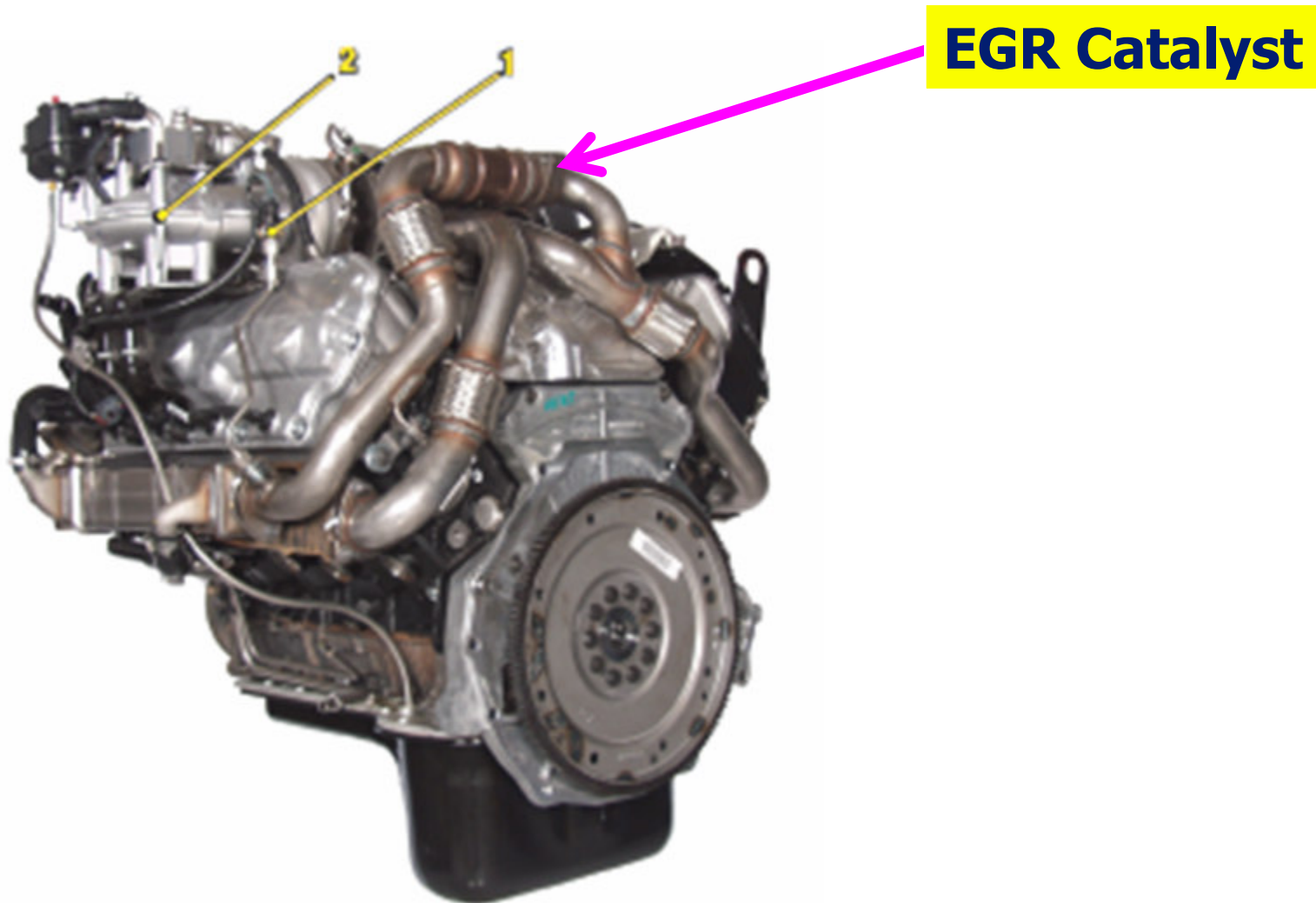
# Deposits

- Cool surfaces exposed to gas collect deposits:
  - Soot - thermophoresis
  - Hydrocarbons - condensation
  - Partially oxidized and pyrolyzed HC
  - Acids – sulfuric, nitric, formic, acetic
- Deposit concerns are worse when
  - Wall temperatures are low
  - “Heavy Wet PM” – more likely at low-NO<sub>x</sub> calibrations
- Likely to get worse with future calibrations for very low NO<sub>x</sub> levels!



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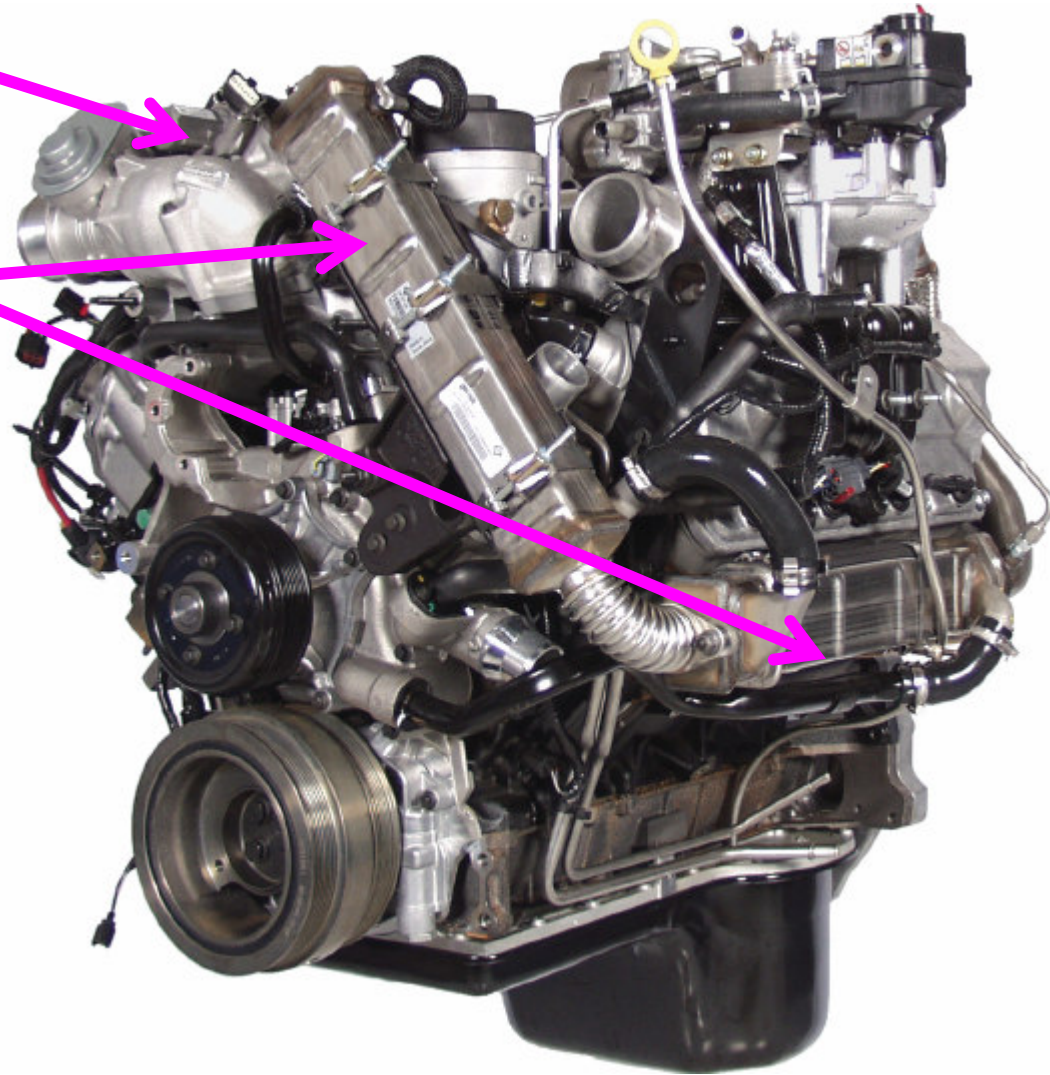
## Test Engine – Rear View



## Test Engine – Left Front View

**EGR Valve**

**EGR Coolers**



## Engine Dynamometer Cooler Fouling Test Cycle

<i>Mode</i>	<i>RPM</i>	<i>Load ft-lb</i>	<i>T<sub>inlet</sub> °C</i>	<i>Space Velocity khr<sup>-1</sup></i>
Idle	700	50	150	127
A25	2100	166	250	503
HSV	2300	300	350	1147

- **Two hours at each point in order**
- **Repeat until effectiveness stabilizes**





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# Response Variables

- Effectiveness

$$\varepsilon = \frac{q_{\text{actual}}}{q_{\text{max-theoretical}}} = \frac{m_{\text{exh}} C_{p,\text{exh}} (T_{\text{exh},i} - T_{\text{exh},o})}{m_{\text{exh}} C_{p,\text{exh}} (T_{\text{exh},i} - T_{\text{coolant},o})} = \frac{(T_{\text{gas in}} - T_{\text{gas out}})}{(T_{\text{gas in}} - T_{\text{coolant in}})}$$

- ECAT HC conversion efficiency

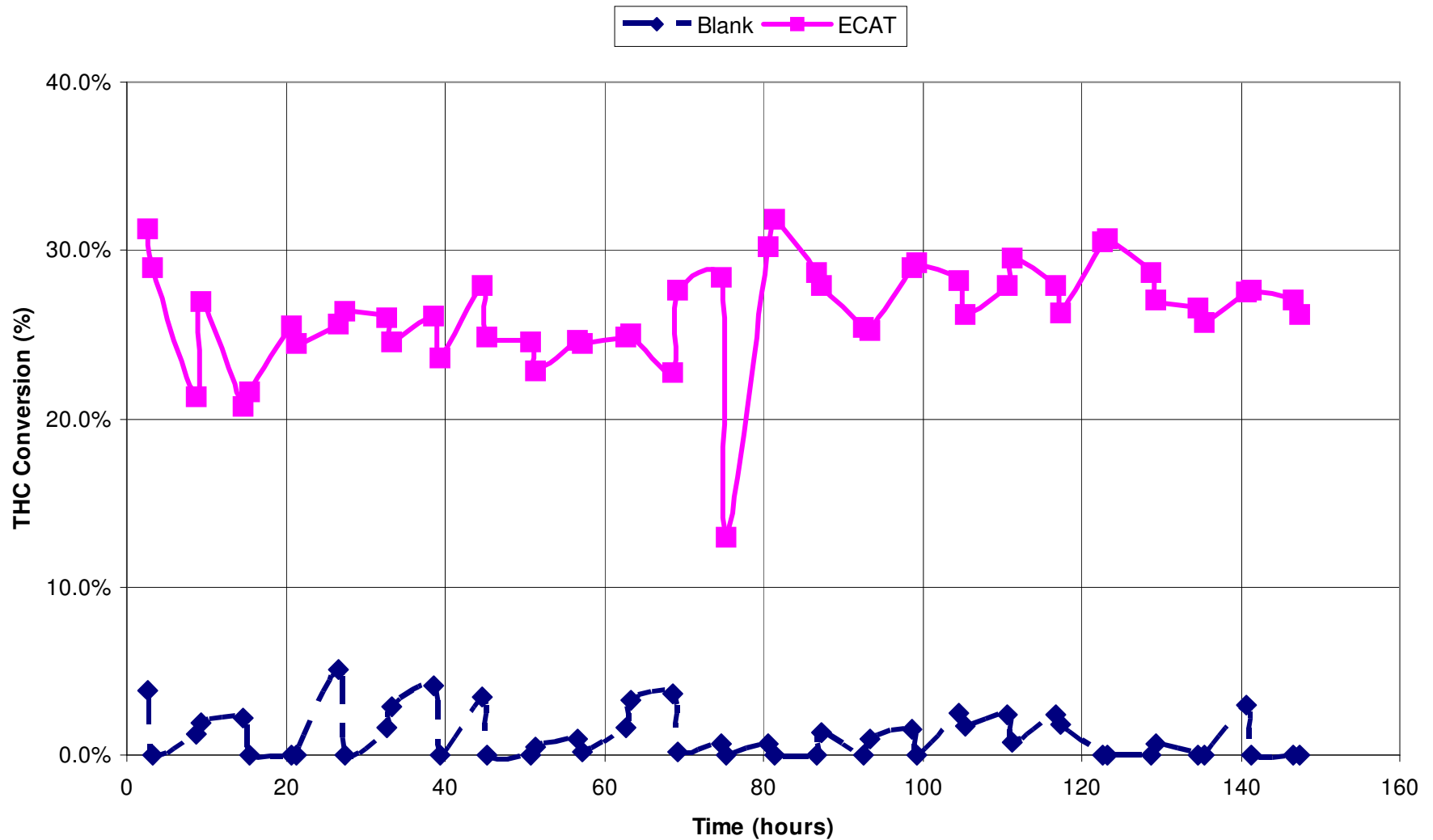
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## Fouling Test – ECAT Effect

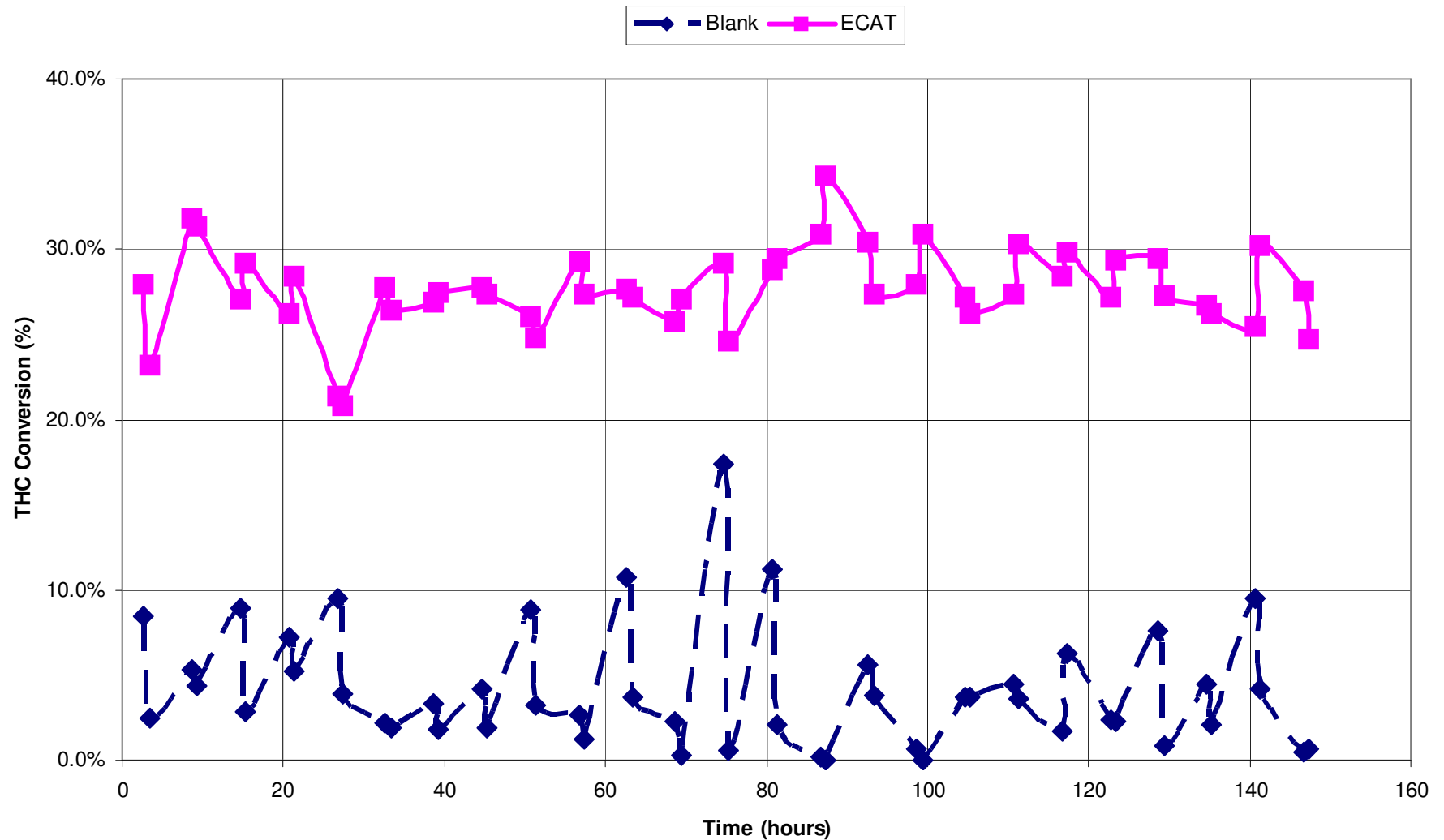
- Cooler fouling test run with ECAT, and with blank ECAT
- ECAT
  - 200 cpsi metallic substrate
  - 60 mm diameter by 90 mm long, 0.24 L volume
  - Oxidation catalyst formulation
  - Compared to coated monolith without PGM
  - Samples prepared for Ford by Johnson Matthey
- Test fuel
  - Canadian market 2005-2006 fuel
  - ~400 ppm sulfur
  - 25-30% aromatics



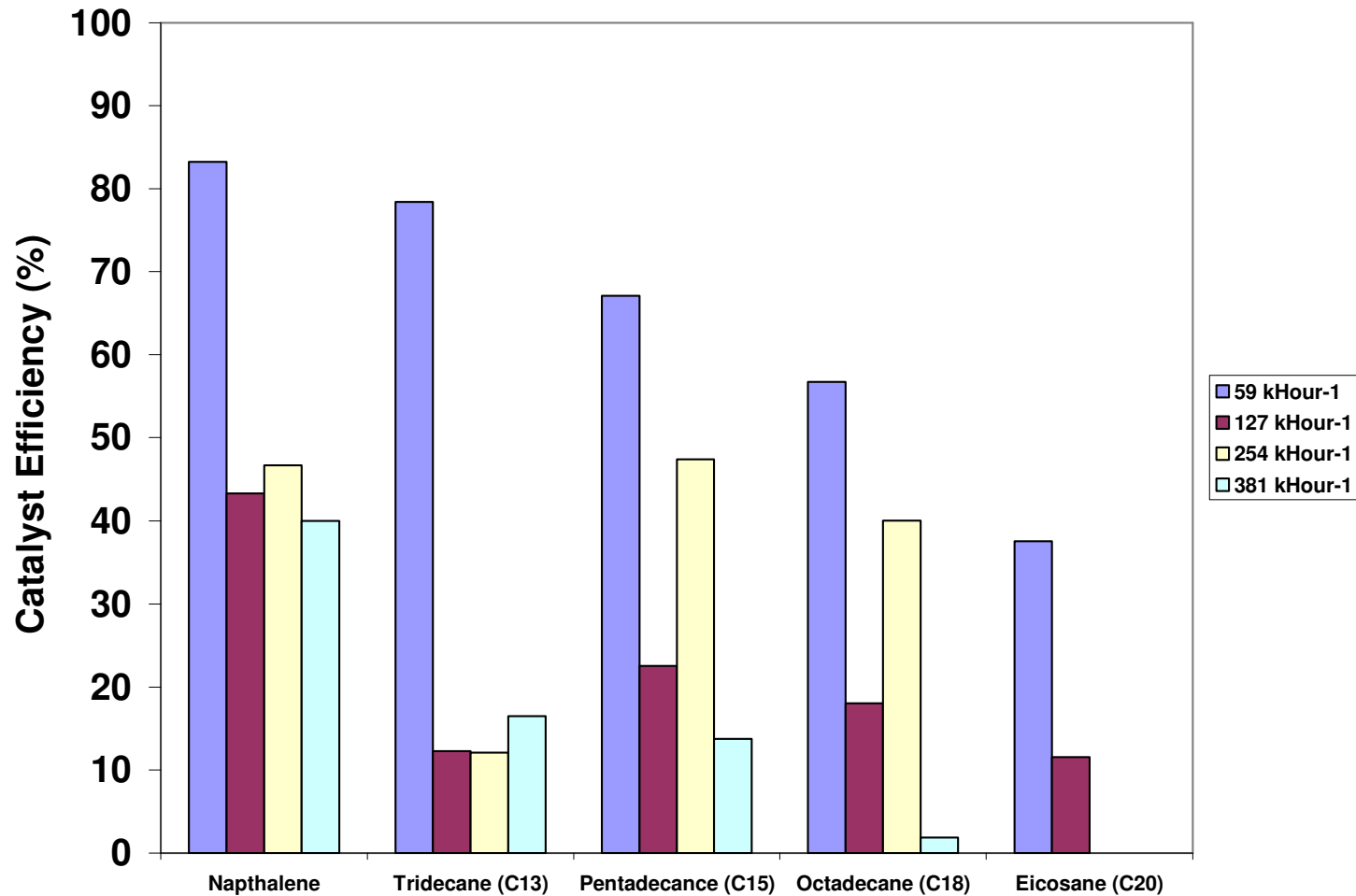
# ECAT HC Conversion – A25



# ECAT HC Conversion – HSV



## ECAT Conversion for Different Species and Space Velocity

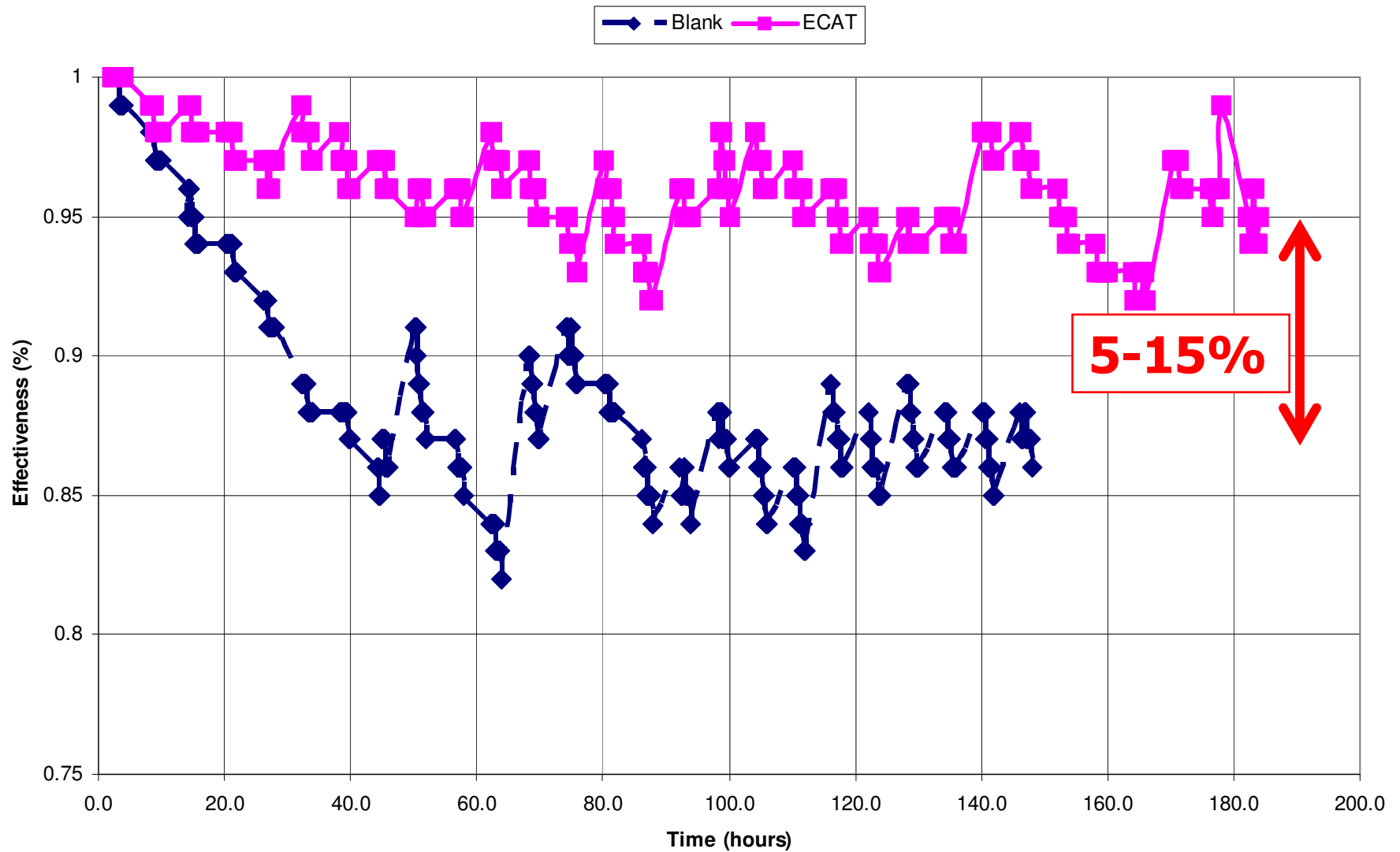


- Reference catalyst and engine
- Testing at ORNL



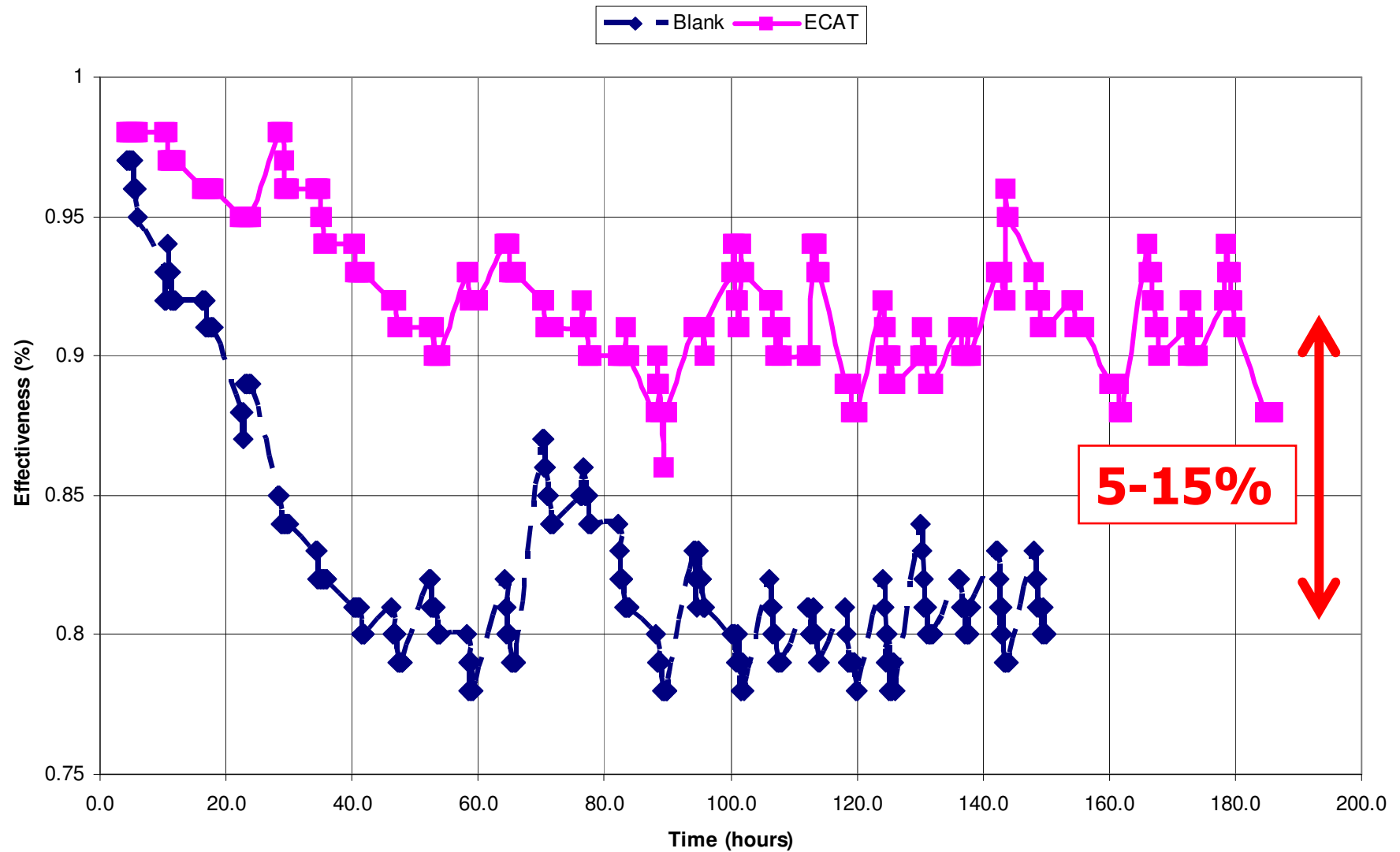
# ECAT Effect on Effectiveness – A25

EGR Cooler System Effectiveness @ A25



# ECAT Effect on Effectiveness - HSV

EGR Cooler System Effectiveness @ HSV



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# Cooler Deposit Analyses

- Following a fouling test with ECAT
- Deposits were analyzed at ORNL



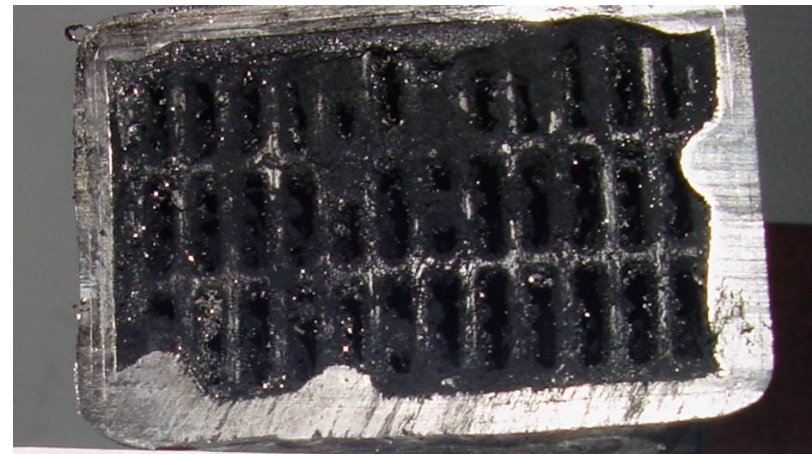


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# Deposits



**Cooler 1 Inlet**

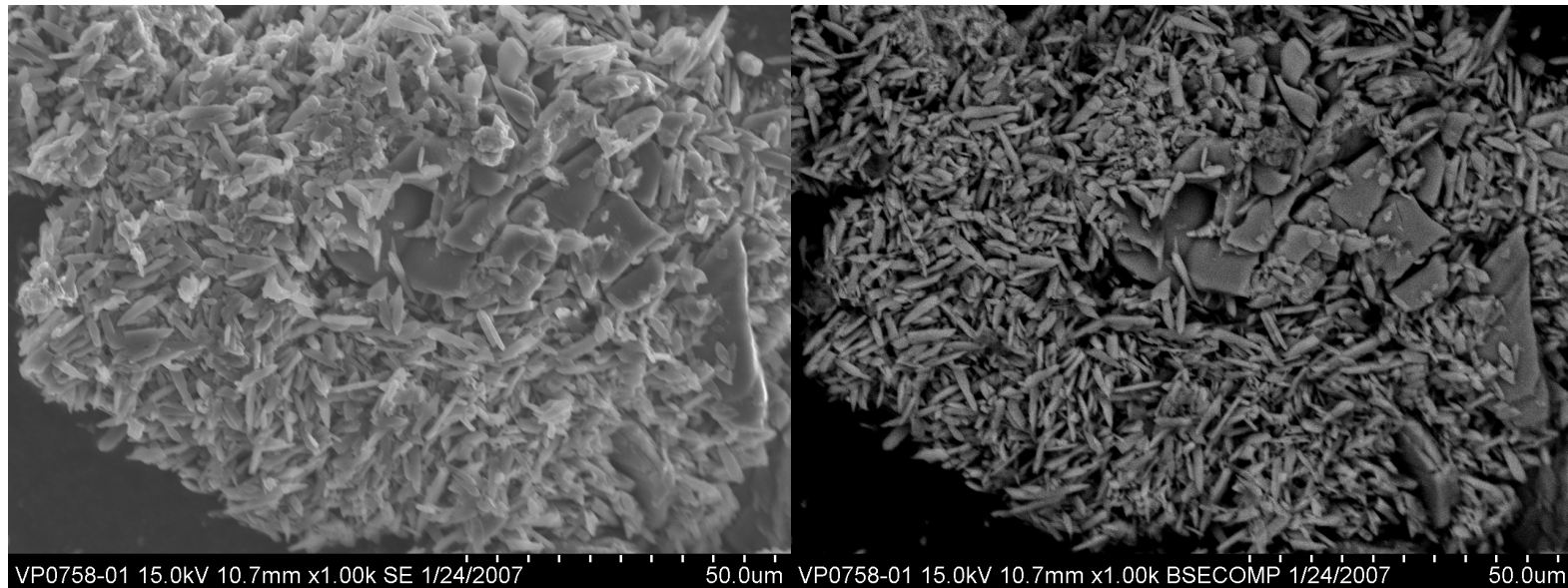


**Cooler 2 Inlet**

# Electron Microscopy of "Ash" Particles Showed Significant Sulfate Fraction

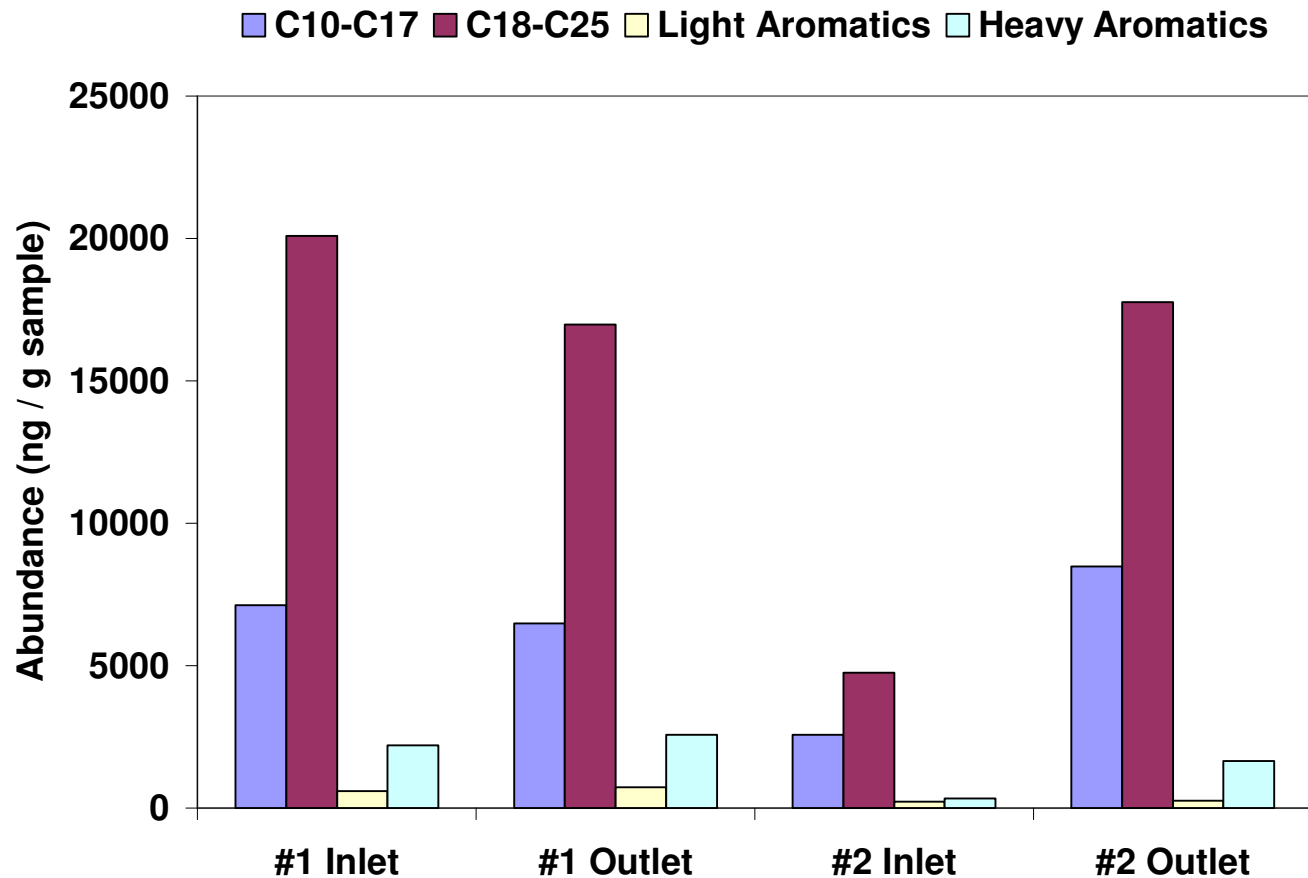
Secondary Electron Image

Back-Scattered Electron image

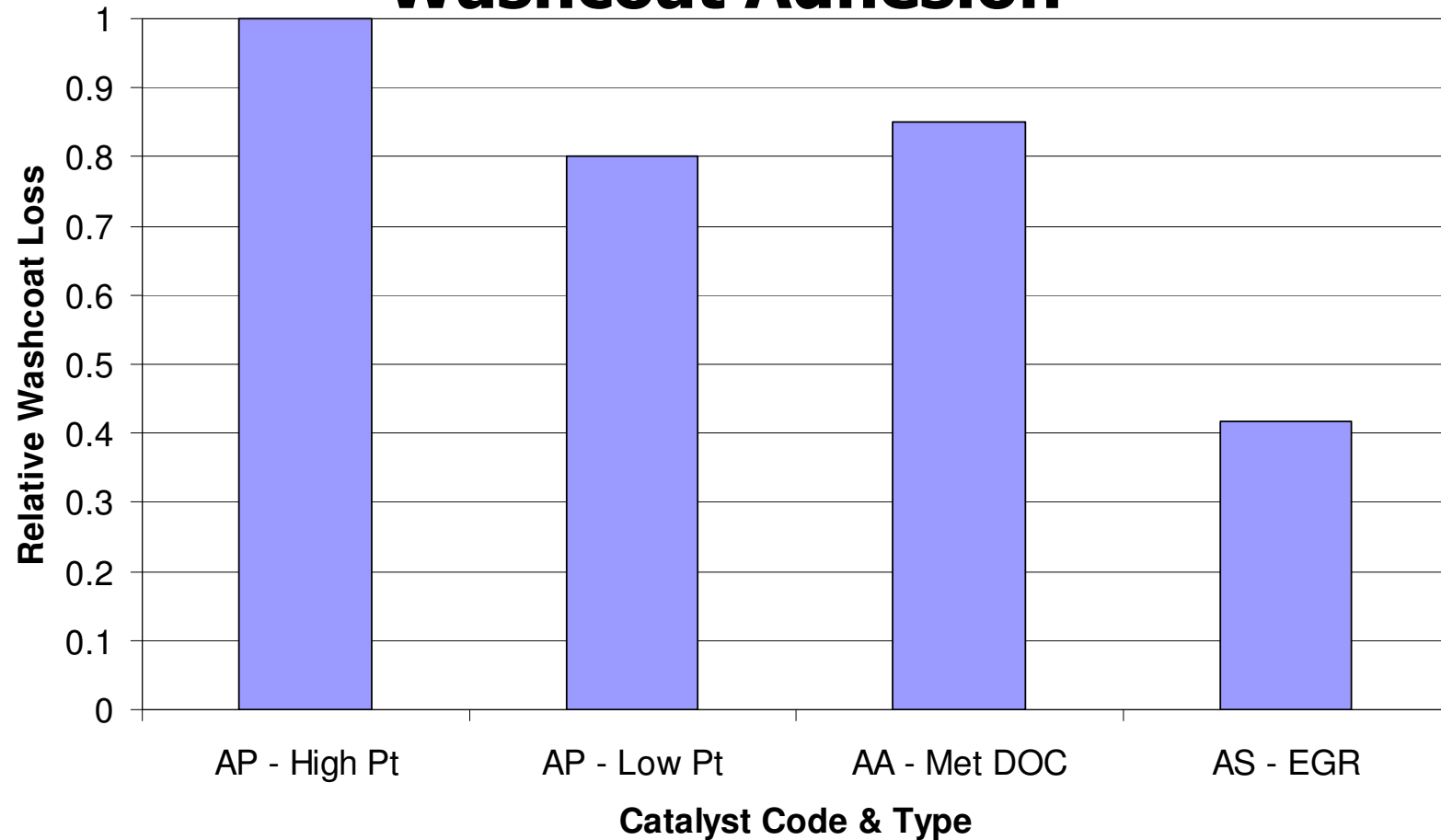


- A significant percentage of the soot was composed of a sulfate phase shown here. The sulfate appeared like grains of rice approximately 5 to 10 microns long and 2 microns wide.
- Consistent with oxidation of fuel sulfur by ECAT; also consistent with fuel sulfur content.

## Chemical Extraction and Analysis Showed that Deposit HCs were Dominated by the Heavy Fraction.



## Washcoat Adhesion



- Washcoat loss puts powder into engine
- Development improved adhesion
  - "AS" versus earlier designs



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# Conclusions

- An ECAT has reduced the rate of EGR cooler fouling
- This can be accomplished with a remarkably high SV
- Washcoat adhesion improved



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# Thanks For Your Attention.....

- Questions?

