

Distributed Impact of Sulfation on LNT Catalyst Reactions

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Focus is on Spatial Nature of LNT Sulfation

Background and Motivation:

- **Sulfation details of strong interest to OEMs**
 - Feedback from CLEERS Workshops and Focus Meetings
 - Recent CLEERS priority polling
 - ORNL CRADA interests
- **Previous engine-based sulfation studies w/ high-S diesel show plug-like loss of NO_x capacity**
 - Other works suggest both distinct and distributed S-front

Goals:

- **Measure spatial nature of sulfation – plug or distributed**
- **Evaluate impact of sulfation on nitrogen selectivity**
- **Identify sulfur species involved in desulfation**

Experimental Approach

- **Catalyst: Umicore GDI LNT (CLEERS Reference)**
 - 7/8" x 3" x 625cpsi core, CLEERS degreened
 - Further degreened before starting S/DeS study (NSR short cycling at 550 C for 5 h)

- **Measurements & Analytical Techniques Applied:**

Technique	Location	Species	Probe Size
Chemi.	Cat. Out	NO, NO _x	n/a
FTIR	Cat. Out	NH ₃ , N ₂ O, NO ₂	n/a
SpaciMS	Intra-cat.	H ₂ , NO _x , H ₂ S, SO ₂ , COS	150- μ m O.D.
Thermocouple	Intra-cat.	T	250- μ m O.D.

- **Procedure:**
 1. Performance evaluation (short NSR & OSC)
 2. 1st S dosing, continuous S dosing w/ short NSR cycling
 3. Continue short NSR w/o S dosing for 30 min (SO₂ purging)
 4. Performance evaluation (short NSR & OSC)
 5. 2nd S dosing, repeat 2 thru 4
 6. Desulfation by TPR (325 to 700 C at 5 C/min)
 7. Repeat 1 thru 6

Conditions for LNT Bench Performance Evaluation ⁴

T=325 °C, SV=30000 h⁻¹ (total flow=13354 sccm)

NSR Short Cycling

Environment	Time	Gas Composition					
		NO	O ₂	H ₂	H ₂ O	CO ₂	N ₂
Lean (storage)	60 s	300 ppm	10%	0%	5%	5%	Bal
Rich (regeneration)	5 s	0 ppm	0%	3.4%	5%	5%	Bal

OSC Short Cycling

Lean (storage)	60 s	0 ppm	10%	0%	5%	5%	Bal
Rich (regeneration)	5 s	0 ppm	0%	3.4%	5%	5%	Bal

TPR DeSulfation

Rich T: 325-700C	n.a.	0 ppm	0%	0.1%	5%	5%	Bal
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4 Consecutive S/De-S Cycles Performed

#	[SO ₂] (ppm)	Exposure Time (h)	Total S Loading (g/L)	Note
1	20	2+4	5.1	H ₂ /CO for short NSR; long NSR performed
2	40	1+1	3.4	
3	40	1+1	3.4	OSC performed; longer soak time for deS
4	20	2+2	3.4	OSc performed

Results similar for all studies

Here we focus on the 4th S/De-S cycle

Sulfation Results to be Discussed

- **Nitrogen Selectivity**

- **NO_x Storage and Reduction, NSR**
 - Spatiotemporal NO_x profiles
 - Spatiotemporal H₂ profiles

- **Oxygen Storage Capacity, OSC**
 - Spatiotemporal H₂-consumption\OSC profiles

- **Reaction exotherms**
 - Spatiotemporal Temperature profiles

- **TPR DeSulfation**

- **Conceptual model of NSR Sulfur Effects**

Outline of Sulfation Results

- **Nitrogen Selectivity**

- **NO_x Storage and Reduction, NSR**
 - Spatiotemporal NO_x profiles
 - Spatiotemporal H₂ profiles

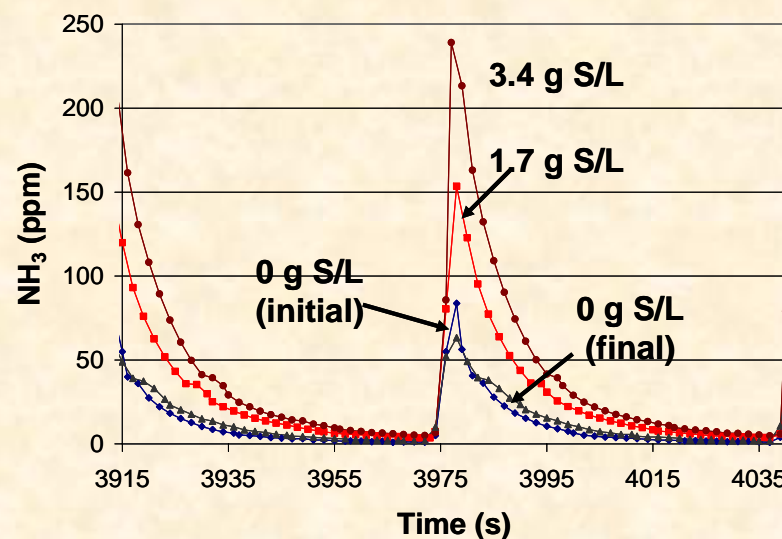
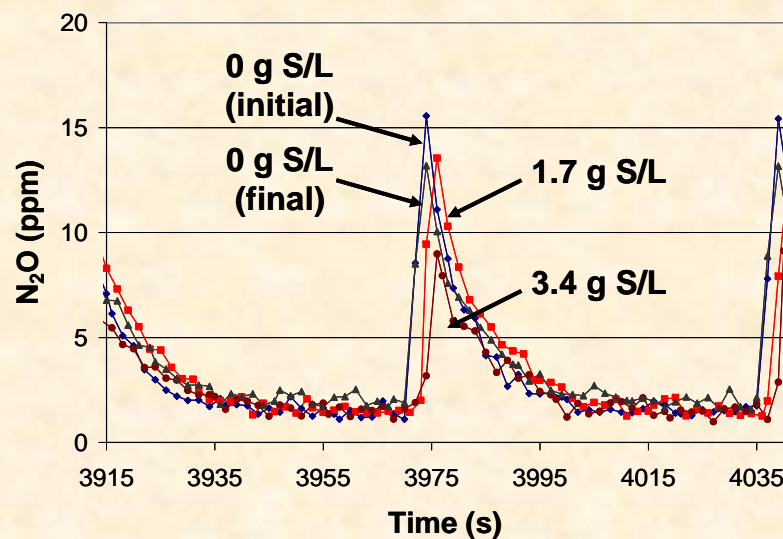
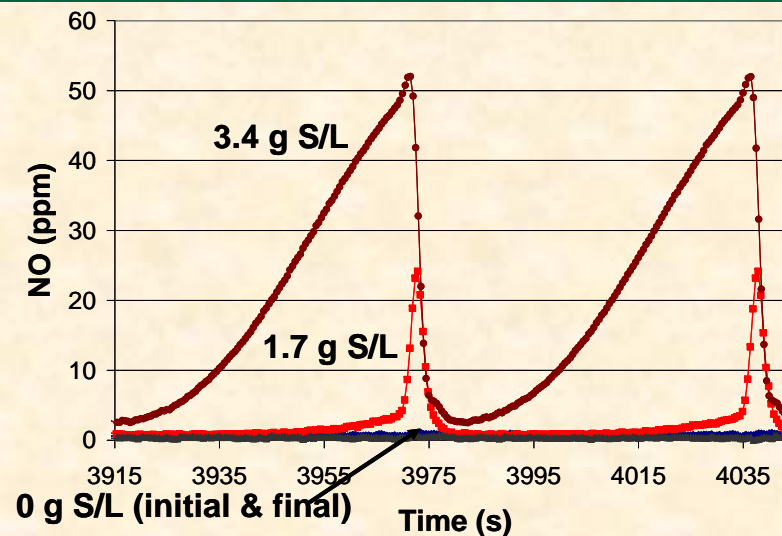
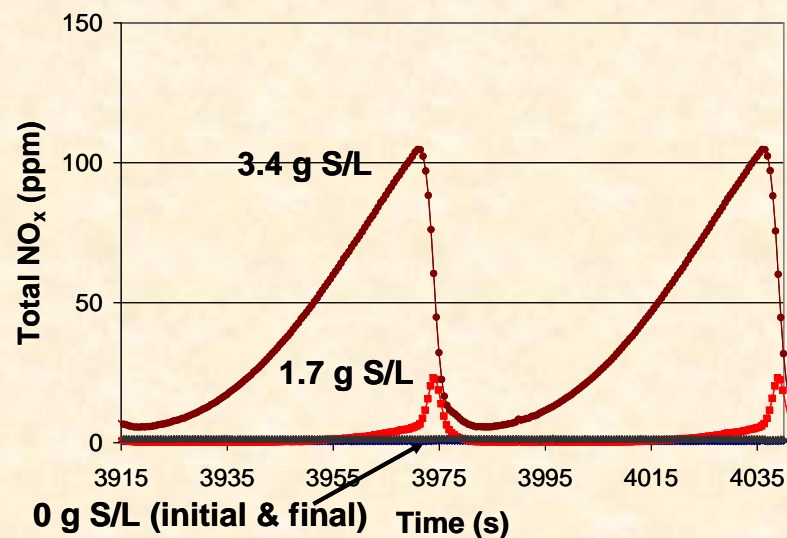
- **Oxygen Storage Capacity, OSC**
 - Spatiotemporal H₂-consumption\OSC profiles

- **Reaction exotherms**
 - Spatiotemporal Temperature profiles

- **TPR DeSulfation**

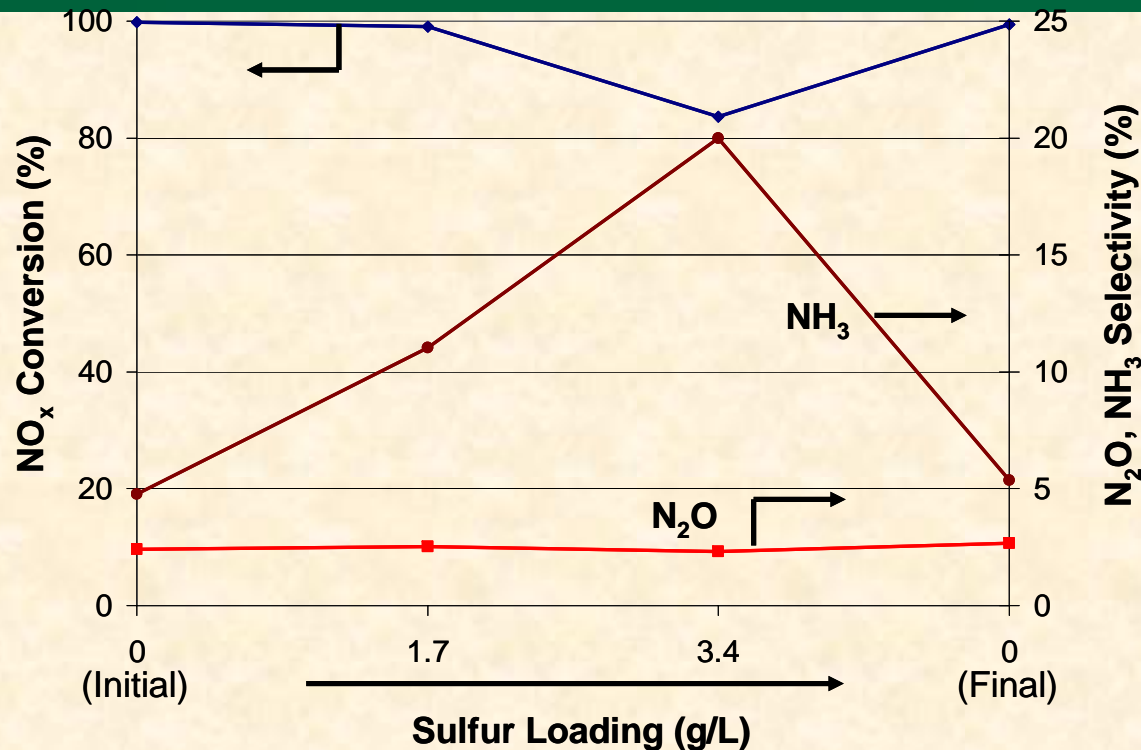
- **Conceptual model of NSR Sulfur Effects**

Sulfation clearly impacts integral LNT performance



Sulfation strongly influenced NH_3 formation and NO_x slip

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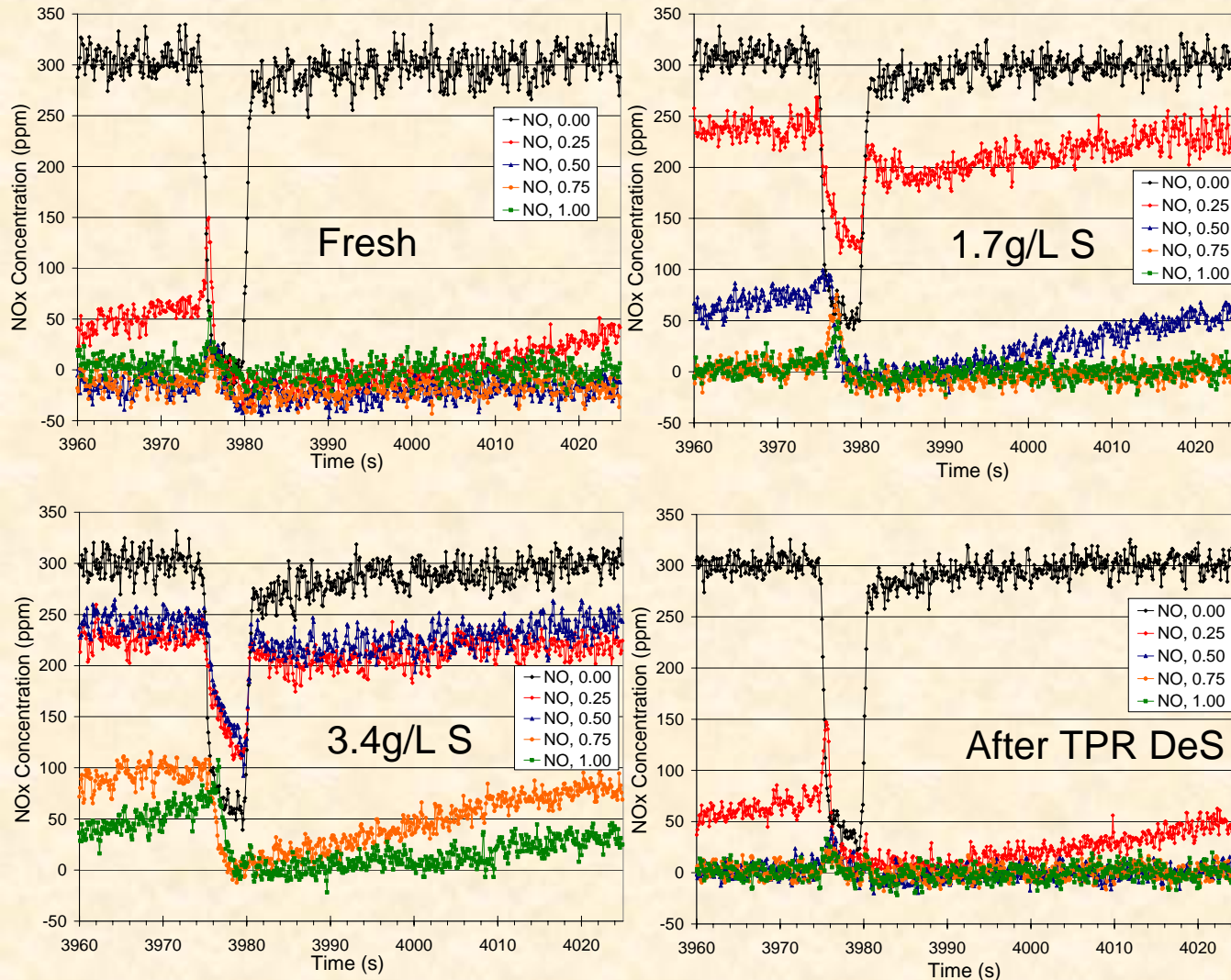


- NO_x conversion decreased with increasing sulfur loading
- NH_3 increased with each S dosing (largest effect)
- N_2O unaffected by sulfation
- Sulfur effects were stable w/o desulfation
- TPR DeSulfation recovered original performance

Outline of Sulfation Results

- Nitrogen Selectivity
- **NO_x Storage and Reduction, NSR**
 - Spatiotemporal NO_x profiles
 - Spatiotemporal H₂ profiles
- Oxygen Storage Capacity, OSC
 - Spatiotemporal H₂-consumption\OSC profiles
- Reaction exotherms
 - Spatiotemporal Temperature profiles
- TPR DeSulfation
- Conceptual model of NSR Sulfur Effects

Sulfation produced distinct stepped effect on NO_x-loading profiles



Fresh-state:

- NSR localized

1st Sulfation:

- Poisoned 1st quarter
- NSR localized in 2nd 1/4

2nd Sulfation:

- Poisoned 2nd quarter

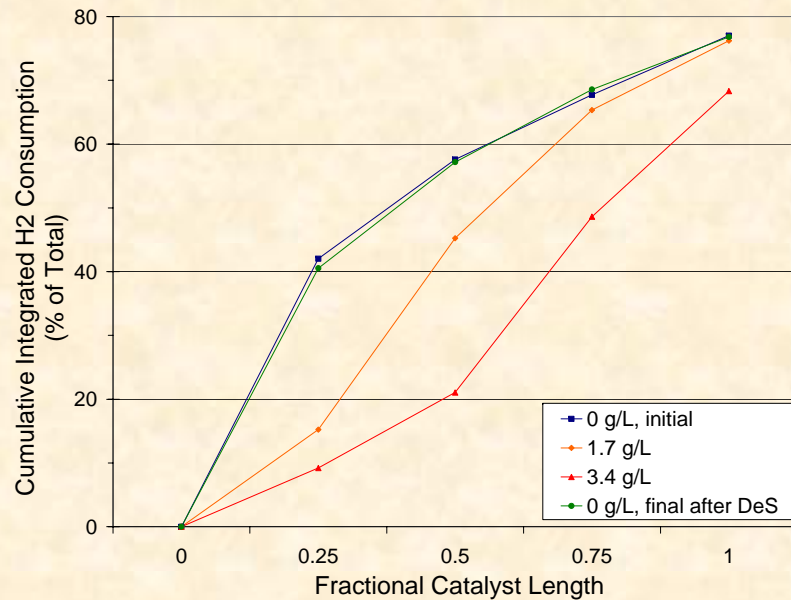
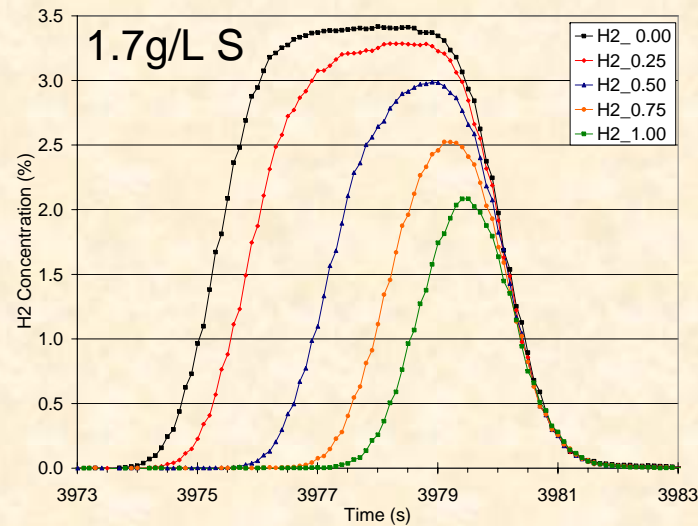
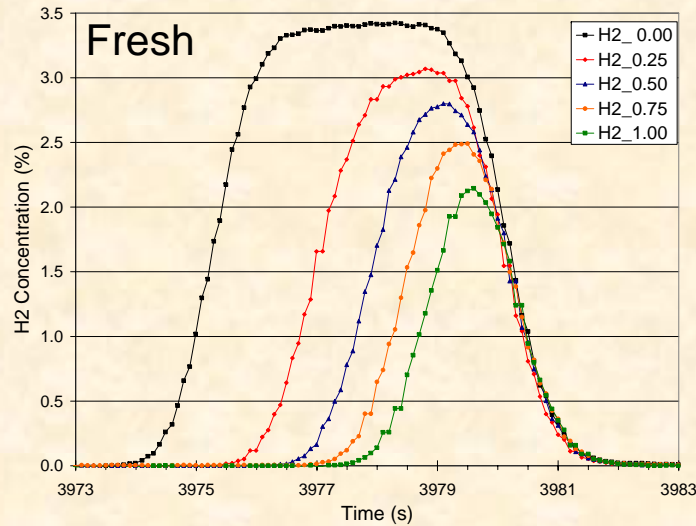
TPR DeS:

Recovered fresh performance

Non-inlet NO_x slip @ saturation:

- Differing NO/NO₂ response at 30amu
- ~240ppm NO₂ + 60ppm NO
- ~ equilibrium

NSR H₂ utilization showed similar distinct stepped response to sulfation¹²



Fresh-state:

- NO_x-attributable H₂-Util. localized in front
- H₂-Util. in back ¾ due to OSC

Sulfated-state:

- Localized NO_x-attributable H₂-Util. moved downstream
- OCS remained active in sulfated region
- Post-NSR OSC region shortened by sulfation

TPR DeSulfation:

- Fresh performance recovered

NSR sulfur response was plug-like and progressive¹³ with increasing sulfur dosing

- **NSR was localized in fresh and sulfated states**
- **NSR completely poisoned in sulfated zone**
- **Localized NSR zone moved progressively down the catalyst axis with increasing sulfur dosing**
- **Post-NSR OSC-only zone shortened by sulfation**
- **NO oxidation could equilibrate in S-poisoned zone and affect downstream NSR**

Outline of Sulfation Results

- Nitrogen Selectivity

- NO_x Storage and Reduction, NSR
 - Spatiotemporal NO_x profiles
 - Spatiotemporal H_2 profiles

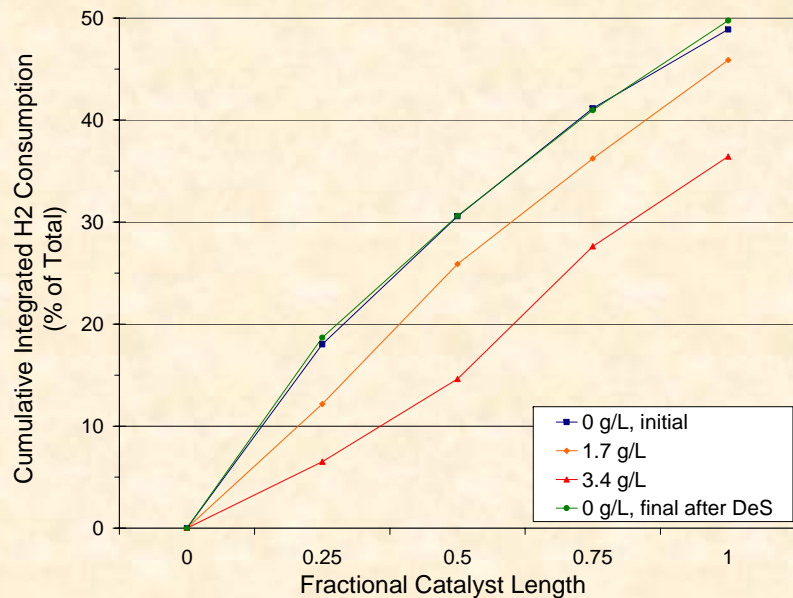
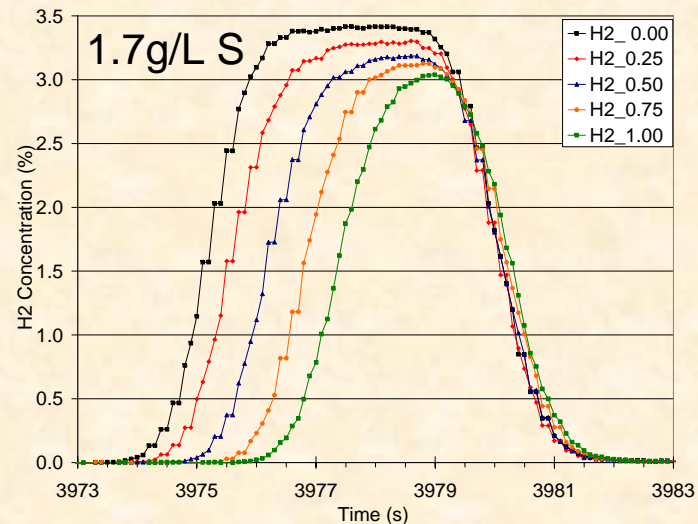
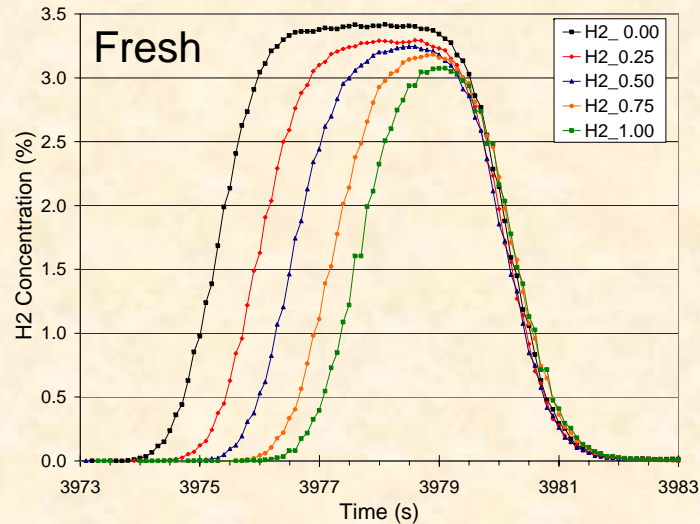
- **Oxygen Storage Capacity, OSC**
 - Spatiotemporal H_2 -consumption\OSC profiles

- Reaction exotherms
 - Spatiotemporal Temperature profiles

- TPR DeSulfation

- Conceptual model of NSR Sulfur Effects

OSC sulfur response was plug-like and progressive¹⁵ with increasing sulfur dosing



OSC is uniformly distributed in the fresh-state and unsulfated zones

1st sulfur dosing:

- Impacts 1st 1/4 only
- Did not completely poison 1st 1/4 OSC

2nd sulfur dosing:

- Continued 1st 1/4 OSC degradation
- Proceeded to 2nd 1/4 OSC

Downstream OSC ~ constant

TPR DeS recovered Fresh performance

Outline of Sulfation Results

- Nitrogen Selectivity

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 - Spatiotemporal NO_x profiles
 - Spatiotemporal H_2 profiles

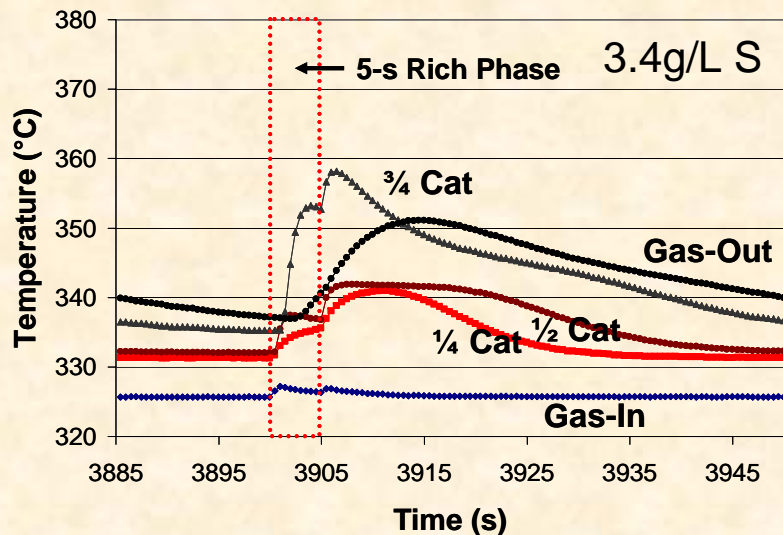
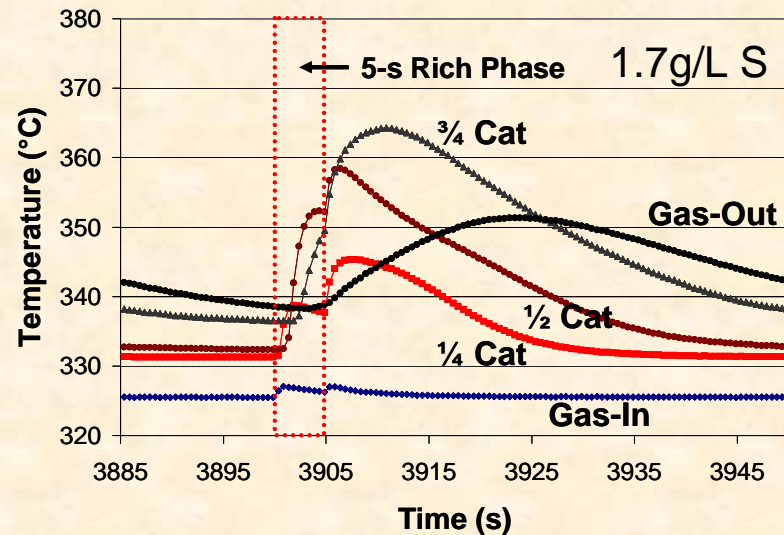
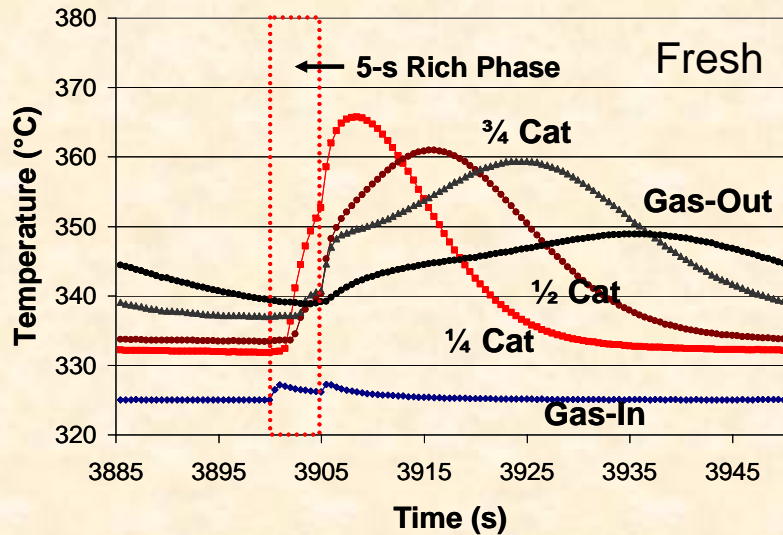
- Oxygen Storage Capacity, OSC
 - Spatiotemporal H_2 -consumption\OSC profiles

- **Reaction exotherms**
 - **Spatiotemporal Temperature profiles**

- TPR DeSulfation

- Conceptual model of NSR Sulfur Effects

Reaction exotherms confirmed localized nature of NSR, and plug-like NSR response to sulfation ¹⁷



NSR-attributable exotherm was localized & S-effect was localized and progressive

- 1st 1/4 : Fresh state
- 2nd 1/4 : 1.7g/L Sulfur state
- 3rd 1/4 : 3.4g/L Sulfur state

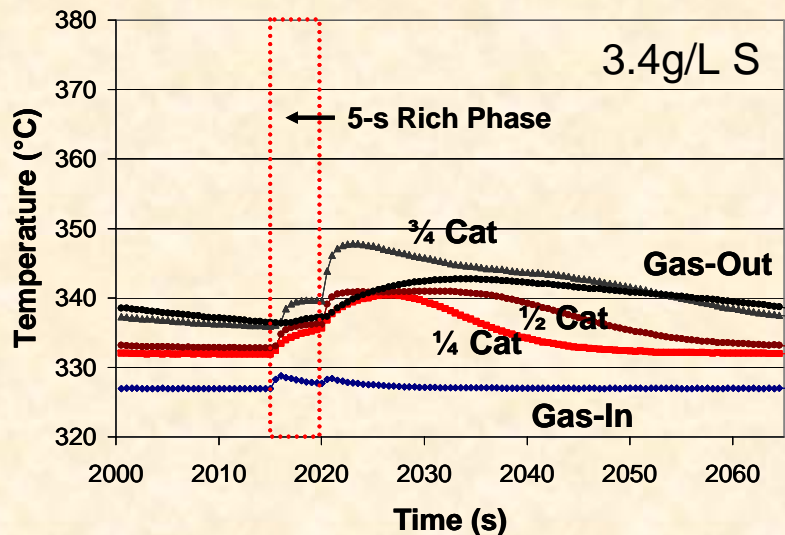
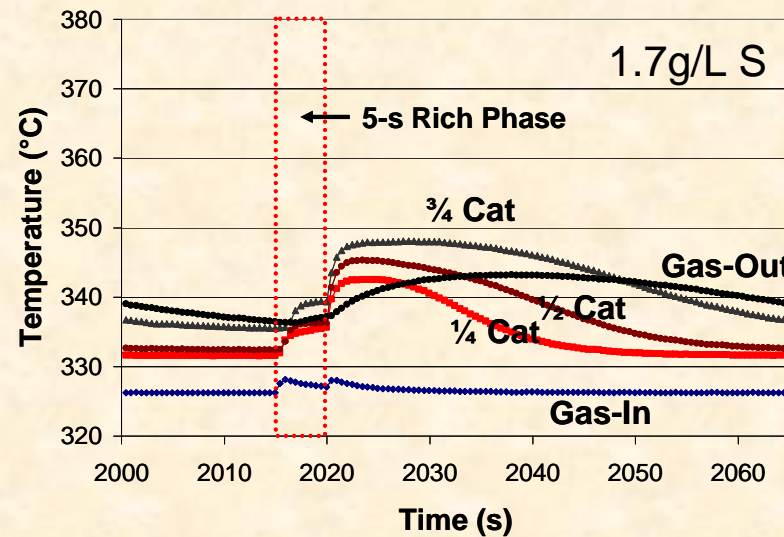
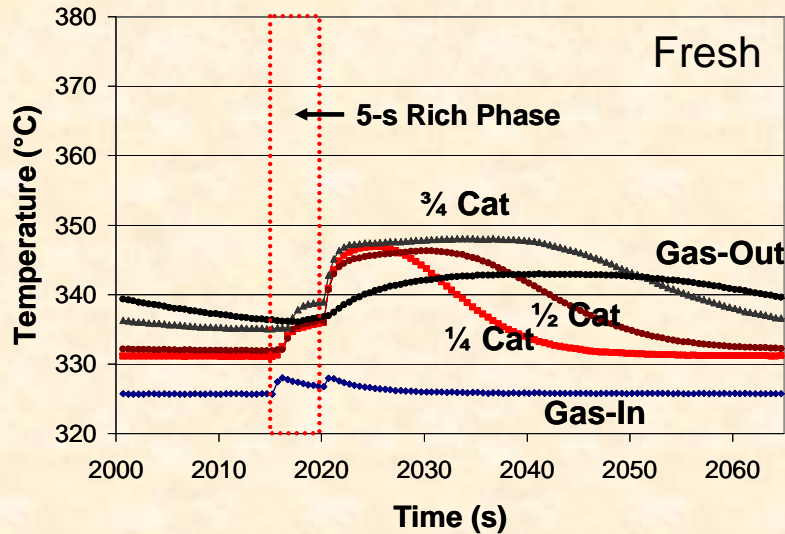
Total NSR exotherm moved but was ca. constant as sulfation progressed

TPR DeS recovered Fresh-state performance

Confirmed SpaciMS-based observations

NSR and OSC S-zone exotherms similar

Reaction exotherms confirmed distributed nature of OSC, and OSC response to sulfation



- Rich-phase exotherm : $H_2 + \text{stored } O_2$
- Rich-Lean transition exotherm : reoxidation of Ce_2O_3 to $2CeO_2$
- OSC-attributable exotherms :
 - evenly distributed in fresh state
 - progressively poisoned by sulfur exposure
- Sulfated-zone NSR exotherm due to OSC
 - Note similarity in sulfated zones

Outline of Sulfation Results

- Nitrogen Selectivity

- NO_x Storage and Reduction, NSR
 - Spatiotemporal NO_x profiles
 - Spatiotemporal H₂ profiles

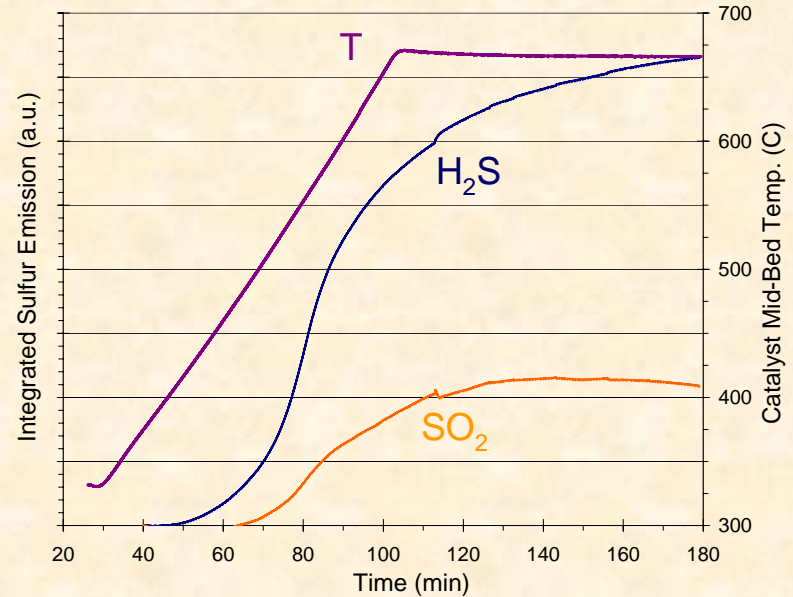
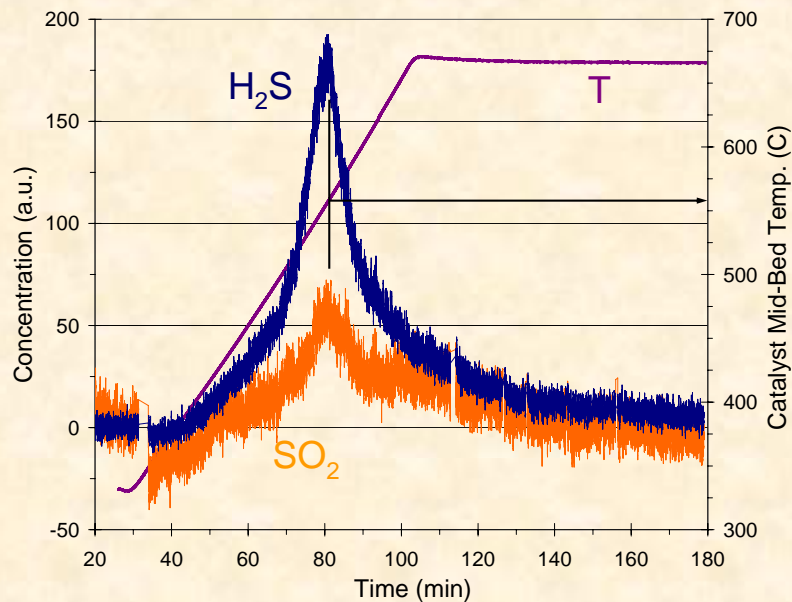
- Oxygen Storage Capacity, OSC
 - Spatiotemporal H₂-consumption\OSC profiles

- Reaction exotherms
 - Spatiotemporal Temperature profiles

- **TPR DeSulfation**

- Conceptual model of NSR Sulfur Effects

H₂S was major product of TPR DeSulfation



H₂S was major product of TPR DeSulfation

- SO₂ more significant after 1st DeS
- COS negligible

H₂S & SO₂ emission ~symmetric around peak at 560C

Major sulfur was released between 500 and 600C

H₂S & SO₂ emission slow after peak

- ~80min high-Temp soak required to reach baseline

Outline of Sulfation Results

- Nitrogen Selectivity

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 - Spatiotemporal H₂ profiles

- Oxygen Storage Capacity, OSC
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- TPR DeSulfation

- **Conceptual model of NSR Sulfur Effects**

Observation Summary

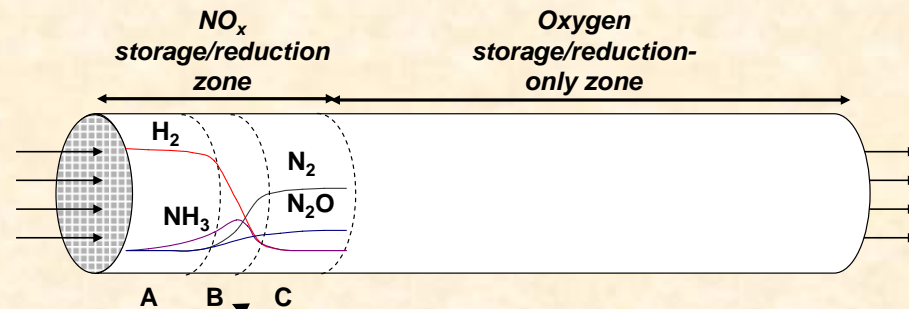
NSR was localized and OSC was distributed

Sulfation:

- Impacted NSR and OSC in a progressive plug-like manner
- Poisoned NSR but only degraded OSC
- Results in varying and different NSR and OSC zones
 - Displaced NSR zone downstream
 - Created upstream OSC-only zone
 - Shortened downstream OSC-only zone
- Could change NO_x partitioning at NSR zone
- Could change peak lean-phase T
- Affected NO_x conversion
- Affected NH₃ selectivity (major LNT-out effect)

TPR DeSulfation recovered fresh-state performance

Conceptual LNT model: fresh state



Fresh State:

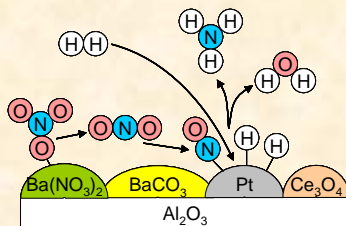
- NSR localized at catalyst front
- OSC evenly distributed throughout
- Residual H_2 and NH_3 from NSR zone oxidized in OSR-only zone

Three NSR Sub-Zone Regions:

(ref: Pihl, SAE 2006-01-3441)

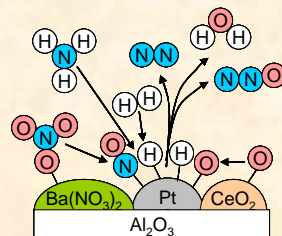
1. Mostly Regenerated
 - $H_2 + \text{Slow } NO_x \text{ sites} \rightarrow NH_3$
2. Regeneration Front
 - NO_x and O_2 reduction via H_2 and NH_3 (upstream product)
3. Un-Regenerated
 - Minor desorption

Sub-zone A:
mostly
regenerated



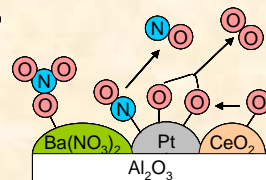
*NH₃ formation at slow
NO_x release sites*

Sub-zone B:
regeneration
front



*NO_x & oxygen reduction
with H₂ & NH₃*

Sub-zone C:
un-regenerated



*Minor desorption of
adspecies due to oxygen
depletion*

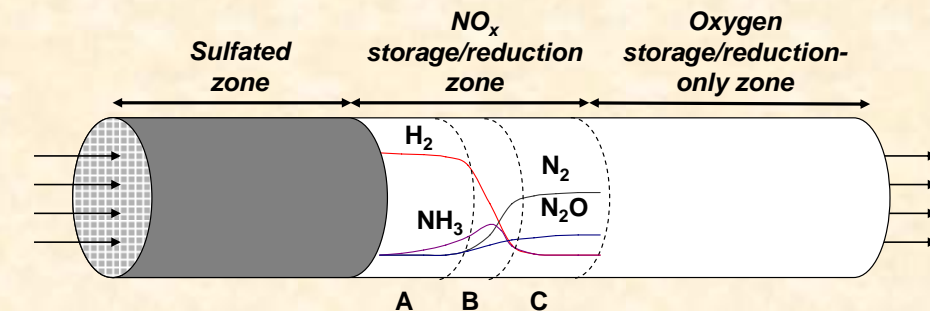
Conceptual LNT model: sulfated state

Sulfation:

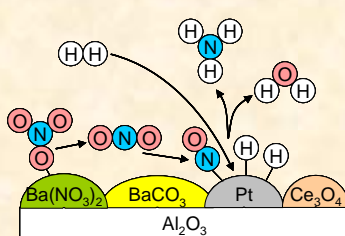
- Creates a localized sulfation zone at the catalyst front
- Sulfation front is distinct and progresses like a plug
- In the sulfated zone:
 - NSR is inactive
 - OSC is active but degraded
- OSC in sulfated zone:
 - Equilibrates NO/NO₂
 - reduces reductant to NSR zone

As Sulfation progresses:

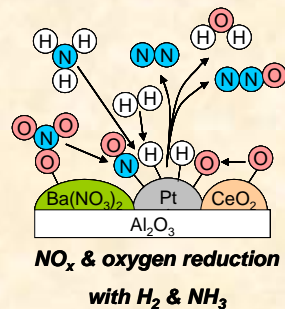
- NSR zone moves downstream
- OSC-only zone is reduced by advancing NSR zone
- NH₃ slip increases due to reduced OSC-only zone



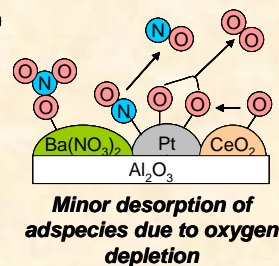
Sub-zone A:
mostly
regenerated



Sub-zone B:
regeneration
front



Sub-zone C:
un-regenerated



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