Accelerated Aging and Characterization of Emissions Control Devices



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NTRC

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History of rapid aging protocol project

- DOE funding from Fuels Technology APBF
- Started with survey of industry in 2003
 - Catalyst, engine, auto, lube industries, and universities
- Goals focused by survey and further discussions
 - Phosphorous poisoning of DOC
 - Thermal aging of LNT
 - Ash loading of DPF
 - Thermal aging of zeolite SCR resulting from DPF regeneration
- Overall Vision:
 - As fuels and catalyst formulations change, can a simple easy to implement protocol be developed that adequately replicates typical on-road conditions without running end-of-life tests
- Secondary vision:
 - Can the results be used for general emissions control models
 - Both materials changes models and kinetic models



Approach

- Small engine and bench reactor tests
- Commercial catalysts, real suppliers
- Extensive materials characterization
- Develop rapid protocols for aging and poisoning
- Determine what materials characterization most useful, how materials changes relate to performance changes
- Compare rapid protocols to field aged parts
- Model aging results to provide more insight and usefulness





Exhaust





LNT

Bench Core-Reactor Based Thermal Aging



Thermal-Aging with Exotherm in a Furnace





	Lean (130s)	Rich (50s)
NO _x	300 ppm	300 ppm
CO ₂	5%	5.00%
CO	0	5.10%
H ₂	0	3.25%
0 ₂	11%	4.00%
H ₂ 0	4.2%	4.20%
N ₂	balance	balance

- Low Temperature Ba-only LNT (fully-formulated)
- The center of the catalyst reaches a nominal aging temperature of ~900°C
- The front section of the catalyst experiences higher aging temperature
- Similar to profiles seen in our single cylinder engine



Aging at 700 and 800C has small impact



- Performance measured periodically using CLEERS fast cycling protocol
- Minimal effects observed beyond two standard deviations of fresh measurements
 - Aging at 800C denotes a mild deactivation at 200 & 400C



Both Step Changes and Gradual Deactivation Observed when aging at 900 and 1000C



- Evaluations at 400C show immediate step change in activity after only 5 cycles
- Evaluations at 200 and 300C demonstrate a more gradual deactivation



Specific Surface Area Correlates well to NOx Performance Losses at 400C



- Surface area decreases as aging temperature and aging time increase
 - Consistent with earlier study on model LNT and high temperature LNT
- Remaining materials characterization is incomplete, so other correlations incomplete
 - Probably very intertwined and modeling necessary to accurately capture materials/performance effects



XRD Can Measure Average PGM Size of Highly Sintered Particles



for the Department of Energy

- Pt aged at 700 and 800C too small to measure w/ XRD
- Maximum particle size of ~24 nm occurs after aging at 1070°C for 350 cycles
- Thermocouple location and sectioning of LNT allows precise measurements with known T history
- Some BaAl₂O₄ formation also apparent



Specialized TEM measurements detect size distribution...and eventually PGM size after 700 & 800C



for the Department of Energy

EPMA Ba Elemental Maps of LNTs Aged at 1000°C for 350 Cycles



Comparisons to Field Aged Catalysts...

- Have had difficulty obtaining field aged LNTs, therefore difficult to compare
 - General trends seem to fall in line with reported effects
- Have geared NOx performance measurements to allow modeling of results
 - Implemented CLEERS protocol portions when possible
 - Have focused on extensive materials characterization to allow performance correlations and modeling
 - Hopefully...allows modeling of both material effects and performance while accounting for material effects



DPF

Engine-based Rapid Ash Loading of series of diesel particulate filters



DPF Durability Issues

- Durability of DPF systems is affected by accumulation of residual ash
- Excellent correlation has been demonstrated between lube-oil consumption and ash formation
- As the ash forms on the channel walls, some breaks loose and deposits in the rear of the DPF





X-Ray analysis of DPF after 90,000 km of operation[#]

* - Bardasz, E., Cowling, S., Panesar, A., Durham, J., & Tadrous, T., SAE Tech. 2005-01-2168.

- Blanchard, G., Seguelong, T., Michelin, J., Schuerholz, S., & Terres, F., SAE Tech. 2003-01-0378.



Test Protocol

- 5% Lube oil mixed with ULSD fuel
- Backpressure threshold for each cycle
 - Loading Cycle
 - Baseline backpressure recorded after stabilization
 - Active regeneration when 0.2 psi above the baseline



- Target soot loading of 2 grams (~3 grams per liter)
- Regeneration
 - Initiated with fuel injection into engine exhaust
 - Initial set point to keep DPF front below 600°C
 - to avoid uncontrolled regeneration
 - Adjust set point such that DPF exit gas temperature to ~725°C
 - All regeneration cycles last 30 minutes



Experimental Setup

- 517cc Hatz Engine is operated at 1500 RPM continuously
- DPF easily removed for periodic weighing
- 700cc DPFs evaluated
 - Cordierite, SiC, mullite
 - High PGM, Low PGM, and uncoated filters

Mullite



Typical Temperature, Pressure and Mass Profiles During Loading and Regeneration

1412

Π

500

1000

1500

Time (minutes)

2000

- Results for Cordierite (high PGM)
- Steady increase in backpressure
 - High backpressures resulted in higher soot accumulation rate
- Rate of soot accumulation is consistent and repeatable
 - Loading cycle consistently achieves soot loading of 2 grams
 - Regeneration strategy effectively removes soot from the DPF





Linear (Loaded)

3000

2500

Backpressure Increases with Ash but Varies with DPF Material

Cordierite and SiC are ~parallel

Rate is 0.036-0.040 psi/g_{ash}

Mullite increase is 6x slower
 – Rate is 0.006 psi/g_{ash}



Lightoff Temperature Increases w/ash on C-DPFs

- Lightoff temperature recorded when backpressure begins to decrease
- Highly-catalyzed substrates initially demonstrate lower lightoff temperature
 - After ~5 grams of ash accumulation, comparable to other substrates
- Ash has minimal effect on lightly-catalyzed and non-catalyzed substrates



Managed by UT-Battelle for the Department of Energy

SEM and Elemental maps of Calcium demonstrate ash gradient along DPF channels





50-h Ash Deposits Similar to 600h Engine-aged DPF



 Compares well with ash deposits in extended engine dynomometer tests Nemoto et al. ash layers in DPFs after 600 hrs*



Middle



Outlet



Test No.1 DH-1 dass

* - Nemoto, S., Togawa, S., Kishi, Y., Ishikawa, T., Matsuura, K., Hashimoto, T., Miura, M., & Yamazaki, T., SAE Technical Paper Series 2004-01-1887.



SCR

Engine-based Thermal-aging of DOC-SCR-DPF Emissions Control "System"



SCR: Thermal Aging using System Approach

- Focus on zeolite-based SCR
 - Fe-based zeolites from Catalytic Solutions
- Thermal Aging Approach
 - Engine-based: active DPF regeneration
 - Engine aging \rightarrow engine and bench evaluation



- 670°C with 28ppm SO₂ for 64 hrs
- Field-aged catalysts already received (Euro bus)



Typical Temperatures Profile in the engine bench at 850°C

- Temperature profile is repeatable
- Axial temperature variation along SCR catalyst





Deactivation Occurs more Significantly in Front

- Front section of aged catalysts deactivated severely
- Maximum NO_x conversion of catalyst aged at 850°C is 45%
- Maximum NO_x conversion of catalyst aged at 750°C is 76%
- Engine Aging at 650°C coming soon



Catalytic Solutions Fe-SCR-2; NO_x conversion as a function of temperature; evaluated with 14% O_2 , 5% H_2O , 5% CO_2 , 350ppm NO, 350ppm NH_3 and N_2 balance at 30,000 h⁻¹; Catalyst aged on the engine bench



Composition of simulated exhaust gases

• Evaluations and hydrothermal aging performed at 30,000 h⁻¹

NO_x conversion evaluation

Gas Species	Concentration
H₂O (g)	5%
O ₂	14%
CO2	5%
NO	350ppm
NH₃	350ppm
N₂	balance

Hydrothermal Aging

Gas Species	Concentration
H₂O	5%
O 2	14%
CO ₂	5%
SO ₂	28ppm
N ₂	balance



NO_x conversion of Furnace-aged Fe-Zeolite SCR catalyst at 670°C

•Aging has not significantly affected NO_x conversion

- •Aging performed with 28ppm SO₂
- Middle of catalyst maintained at 670°C during aging



Catalytic Solutions Fe-SCR-2; NO_x conversion as a function of temperature: evaluated with 14% O_2 , 5% H_2O , 5% CO_2 , 350ppm NO, 350ppm NH₃ and N_2 balance at 30,000 h⁻¹; Catalyst aged on BFR

XRD patterns of fresh and aged Fe-Zeolite (Fe-SCR-2) SCR CATALYSTS

•Significantly less zeolite detected in aged samples

•Fe₂O₃ and Al₂O₃ peaks are discernable in aged samples

•Zeolite lattice begins to breakdown around 800°C for many high-silica zeolites

-Zhdanov

 Surface area impact also being measured



Field-aged Fe-SCR shows similar Deactivation Characteristics of Engine Aging at 750C

100%

- Front section of field-aged catalyst severely degraded
- Field-aged catalyst performance slightly worse
 - Suggests actual field conditions exceeded 750C
 - But less than 850C
- Overall vehicle performance was maintained in field



Field Aged-Front

NOx Conversion characteristics of field-aged and fresh SCR catalysts; evaluated at varying temperatures, 5% H2O, 5% CO2, 14% O2, 350ppm NO, 350ppm NH3, N2 balance, GHSV=30,000



Summary and Future Directions

- Project has laid groundwork for diesel emissions control accelerated aging/poisoning protocol development
 - Phosphorous poisoning
 - Bench and engine-based thermal aging
 - Ash loading
- Results have been incorporated into models
 - PSAT aging impact on LNTs
 - MIT starting modeling effort on DOC
 - More incorporation desired and needed

Future plans

- Biodiesel constituent effects (Na/K) on emissions control devices...performance and materials characterization study
- Collaboration with EPA for possible certification of protocols to allow industry implementation
 - Similar to existing gasoline-based accelerated aging protocols



Supplemental Slides



Development of Rapid Protocols





Experimental equipment

- Small diesel engine (ORNL)
- Bench reactor (UTK)
- Catalytic micro-reactor (ORNL)
- Materials characterization (UTK and ORNL)
- Catalysts
 - BASF, Delphi, Catalytic Solutions, Cummins Filtration
 - Some model catalysts, most fully formulated
- Substrates
 - NGK, Dow
- Thanks again to the catalyst and substrate companies who helped us in this research



Engine Bench Setup





Bench-Flow Reactor





Catalytic Micro-reactor Apparatus





200°C Evaluation of Fresh LNT



- Steady state NO_x Conversion is not achieved after ~2 hours of evaluation
- NO_x excursion continues to increase for the duration of the experiment

Managed by UT-Backle storage capacity also continues to fall throughout the experiment for the Department of Energy



200°C Evaluation of LNT Aged at 1000°C for 5 cycles



- Steady state NO_x Conversion is not achieved after ~2 hours of evaluation
- NO_x excursion begins to fall after peaking at ~325 ppm

• NO_x storage capacity continues to decrease throughout the experiment for the Department of Energy



Thermocouple Locations



 By measuring the catalyst's temperature at multiple axial locations we are able to obtain the axial temperature profile during aging and evaluation and to determine the effect of aging temperature on the performance of the catalyst



Performance of LNTs Aged at 700, 800, 900, and 1000°C Using 200°C Evaluation Cycle



• Conversion is lowest after aging at 1000C

Managed by UT-Battelle • Results are not consistent for the Department of Energy



Performance of LNTs Aged at 700, 800, 900, and 1000°C Using 300°C Evaluation Cycle



• Conversion falls with aging temperature as expected



Performance of LNTs Aged at 700, 800, 900, and 1000°C Using 400°C Evaluation Cycle



• Conversions after aging at 900°C and 1000°C are very similar



DPF Durability Issues

 Accelerated lube-oil consumption methods have been shown to produce ash layers that compare favorably with layers formed solely with natural lube-oil consumption



a. 1118 Hr Experiment

b. 165 Hr Experiment

Ash Laver 1.5 inches 4.5 inches 6.5 inches High Ash Oil High Ash Oil Low Ash Oil (with Doping) (no Doping) (with Doping) SEM images of ash layers for

EPMA images of ash layer formed from natural oil consumption (left) and oil misting (right)^[11]

three oil formulations and consumption mechanisms^[15]

[11] Sutton, M., Britton, N., Otterholm, B., Tengström, P., Frennfelt, C., & Murray, I. (2004). Investigations into lubricant blocking of diesel particulate filters. (SAE Technical Paper Series 2004-01-3013).
 [15] Bardasz, E., Mackney, D., Britton, N., Kleinschek, G., Olofsson, K., Murray, I., & Walker, A. (2003). Investigation of the interactions between lubricant-derived species and aftertreatment systems on a state-of-the-art heavy duty diesel engine. (SAE Technical Paper Series 2003-01-1963).



Characterization of ash with ICP

- Identify and quantify (?) ash constituents in oil and fuel
- Quantify (?) recovery of ash constituents in filter

- Results indicate a varying degree of recovery rate
 - up to ~90% for both Calcium and Zinc, but varies from sample to sample
 - May be difficult to completely rely on this technique for quantification
- Distribution within filter corresponds to ash layer gradient along axis
 - Layer thickness is minimal near the inlet
 - Thickness increases along the length of the filter



Mullite Materialistic Details

Mullite (High PGM)

- Initial backpressure lowest of all tests
 - Most porous material

Fewest regenerations required

- Extremely effective in continuously regenerating soot
- Still gain ~2 grams of soot between regenerations
- Backpressure slope is roughly horizontal for the first 2/3 of the experiment, then begins to increase



 Textured surface increases filtration surface area and appears to maximize soot-catalyst interaction when catalyst is present





- Ash recovery is calculated by quantifying the presence of calcium, zinc, phosphorus and sulfur in the fuel, lube-oil and ash-loaded substrate
- Recovery efficiencies as high as 91% were calculated for calcium

- Lengthwise distribution can be analyzed by cutting the sample into sections for individual analysis
- Additional analysis revealed the presence of iron in the ash, accounting for up to 5% of the DPF mass increase

		Test 3 - Cordierite (Low PGM)			Test 4 - SiC (No PGM)			Test 5 - Mullite (High PGM)					
		Zn	S	Ca	Р	Zn	S	Ca	Р	Zn	S	Ca	Р
Average Concentration	wt %	0.548	0.506	1.424	0.000	0.279	0.183	0.601	0.115	0.449	0.270	0.712	0.125
Mass in DPF	g	2.290	2.114	5.948	0.000	1.349	0.888	2.908	0.558	1.894	1.139	3.002	0.527
Concentration in Oil	wt%	0.129	0.749	0.351	0.119	0.124	0.380	0.231	0.110	0.124	0.380	0.231	0.110
Mass in Oil	g	3.424	19.898	9.335	3.174	3.296	10.103	6.143	2.932	3.296	10.103	6.143	2.932
Recovery	%	66.88%	10.62%	63.72%	0.00%	40.94%	8.79%	47.34%	19.04%	57.46%	11.27%	48.87%	17.97%
Concentration in Fuel + Oil	wt%	0.005	0.029	0.013	0.004	0.006	0.032	0.011	0.006	0.006	0.032	0.011	0.006
Mass in fuel	g	2.504	14.924	6.721	2.249	3.066	16.457	5.571	2.964	3.066	16.457	5.571	2.964
Recovery	%	91.42%	14.17%	88.50%	0.00%	44.01%	5.39%	52.20%	18.83%	61.76%	6.92%	53.89%	17.77%



- Test 3: Cordierite (low PGM)
 - Mass of ash increases along the length of the DPF
 - Greatest concentrations observed 4" from inlet
 - Concentrations in rear decrease due to ash plugging and the presence of cordierite channel plugs
 - Presence of phosphorus is negligible
 - Highest ash recovery (~90%)





- Test 4: SiC (no PGM)
 - Calcium concentrations increase along the length of the channel
 - Phosphorus presence is no longer negligible
 - Zinc, calcium and phosphorus appear fairly evenly distributed
 - Ash recovery based on these constituents is fairly low (52%)





- Test 5: Mullite (high PGM)
 - Concentration of all four constituents increase along the length of the channels
 - Ash recovery based on these constituents slightly higher than that calculated for SiC (~61%)





- Ca, Zn, S and P regarded as primary ash constituents
- Compounds identified:
 - CaSO₄
 - Most prevalent in literature (~75%)
 - $Ca_3(PO_4)_2$
 - Zn(PO₃)₂
 - Second-most prevalent in literature (~15%)
 - ZnO and CaO
 - Present in some form in addition to other compounds not easily identifiable
 - CaSO₃
 - Difficult to discern presence as some peaks are masked by other compounds



- XRD (X-Ray Diffraction) used to identify compounds in ash
 - Ca, Zn, S and P regarded as primary ash constituents
 - Compounds identified:
 - CaSO₄
 - Most prevalent in literature (~75%)
 - Ca₃(PO₄)₂
 - Zn(PO₃)₂
 - Second-most prevalent in literature (~15%)
 - ZnO and CaO
 - Present in some form in addition to other compounds not easily identifiable
 - CaSO₃
 - Difficult to discern presence as some peaks are masked by other compounds



- Test 4: SiC (no PGM)
 - SiC and CaSO₄
 responsible for the largest peaks
 - Remaining elements are likely present, but difficult to identify due to peak overlap





- Test 5: Mullite (high PGM)
 - Mullite and CaSO₄ are again readily identified
 - Ca₃(PO₄)₂ is also identified as a strong candidate
 - It is again difficult to confirm the presence of the remaining elements





- Test 7: Cordierite (high PGM)
 - Cordierite and CaSO₄ are the only compounds identified with any certainty
 - There are fewer pronounced peaks
 - Analysis of the numerous smaller peaks further complicates identification





Summary of Accelerated DPF Ash Loading

- Ash loading protocol developed to deposit ash gradient along channel
 - Similar to the gradient observe in unaccelerated tests
- Backpressure increases with all substrates
 - SiC and Cordierite show same rate of increase
 - Mullite shows smaller rate of increase
- Active regeneration light-off temperature increases with ash loading
 - No trend apparent based on substrate material
 - PGM does not appear to assist active soot regeneration when significant ash is present



Application of what we have learned to evaluation of useful life

- Start from field samples and application data
- Evaluate performance in test cell
- Cut sections from catalysts
 - Cores for bench reactor evaluation
 - Materials characterization
 - Micro reactor evaluation
- Correlate materials changes to application data
- Project performance to full useful life
- Model rates of change to provide input for vehicle or system models and to help understand and define phenomena

